

Development of a Novel Process for the Production of Calcium Sulfoaluminate

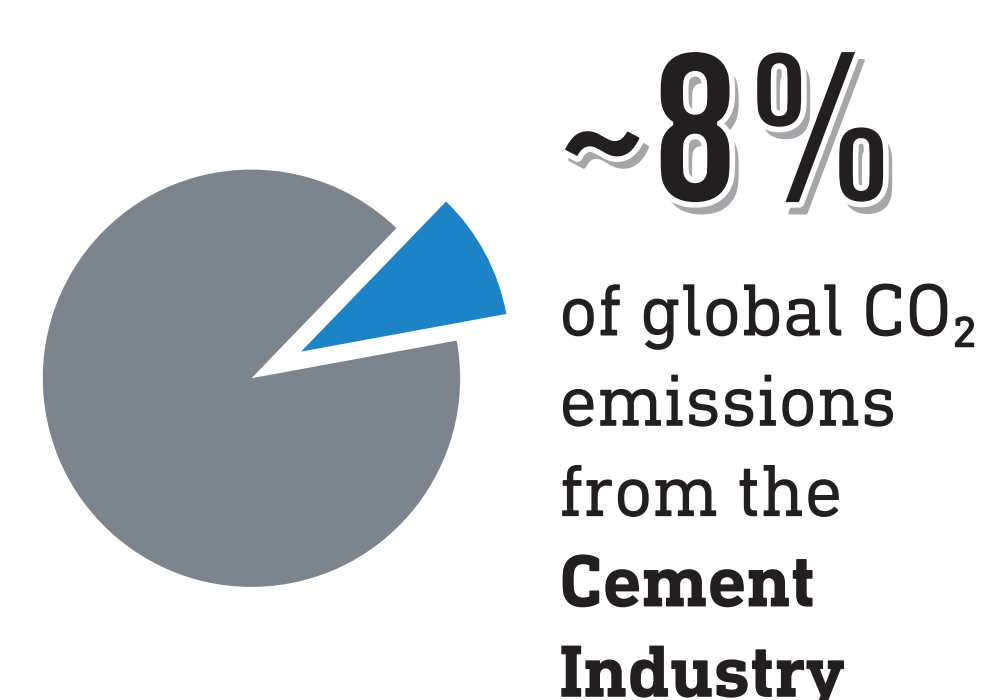
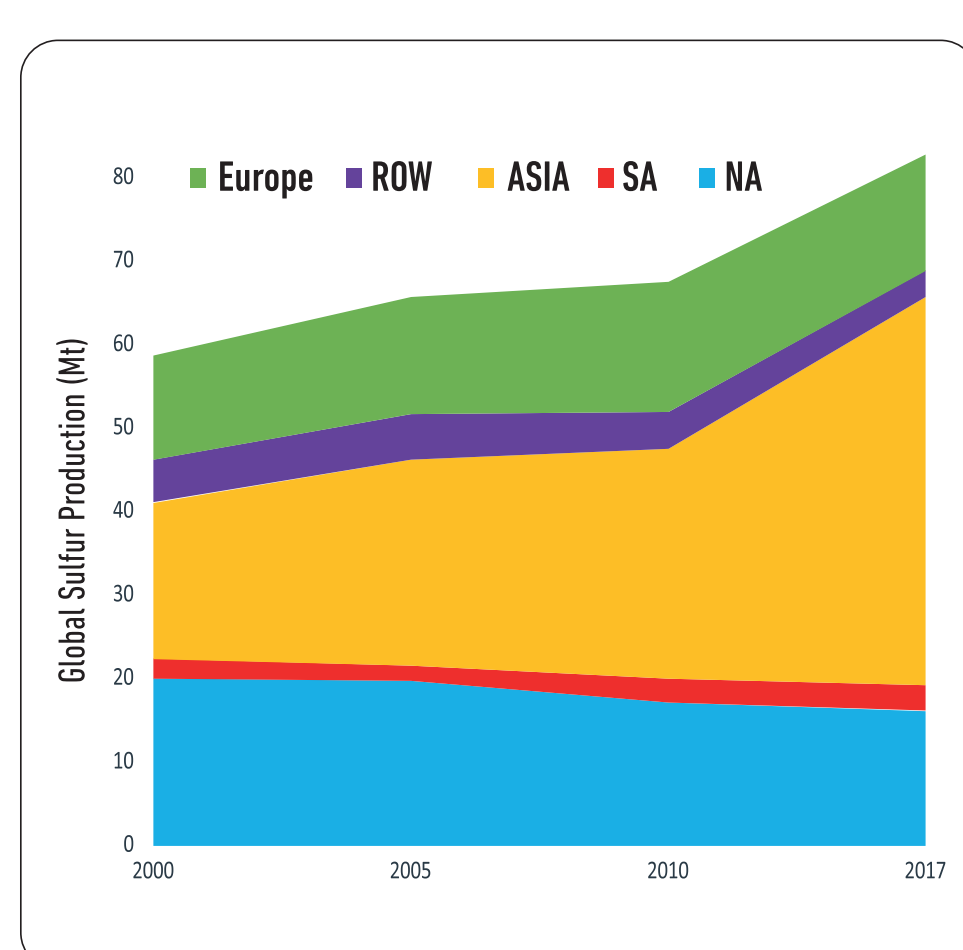
Ammar Elhoweris^{a,b}, Yousef Alhorr^a, José Luis Gálvez Martos^{b,c}, Marcus Campbell Bannerman^b

^a Gulf Organisation for Research and Development, QSTP, Tech 1, Level 2, Suite 203, P.O. Box: 210162, Doha, Qatar

^b Department of Engineering, Fraser Noble Building, University of Aberdeen, AB24 3UE, Aberdeen, UK

^c Systems Analysis Unit, IMDEA Energy, 28935, Móstoles, Spain

Motivations



The surplus of elemental sulfur from the **oil and gas industry** has become a major global challenge.

To deal with the scale of this overproduction, new uses must be found in products which are produced on an equally massive scale.

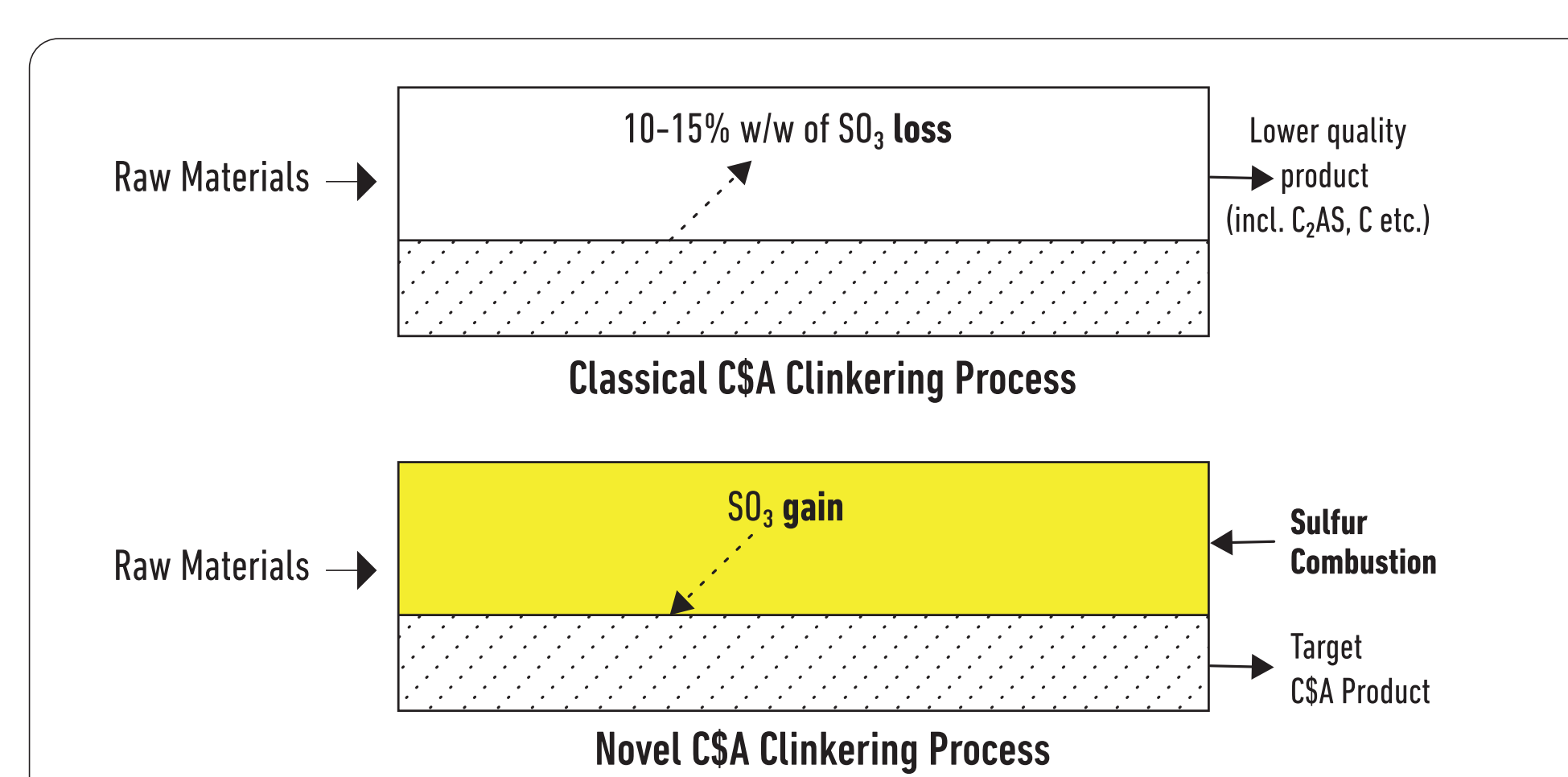
The cement sector is required to make drastic changes to mitigate increasing demands.

Calcium Sulfoaluminate (C\$A) cements are attractive from an environmental perspective as they have considerably lower raw-material and fuel-derived CO₂.

Proposed Solution

Herein, a novel process for C\$A production is proposed whereby SO₃ is completely/partly supplied through the **combustion of sulfur**. The objective of introducing elemental sulfur is twofold:

- as a fuel, reducing the need for hydrocarbon fuels and thereby lowering carbon dioxide emissions
- to supply the SO_x necessary to synthesise the desired clinker mineralogy while avoiding sulfur volatilisation.



Process Economics

C\$A cements have a lower carbon footprint relative to OPC (Figure 3). However, they require an extra source of aluminium oxide which has an important effect on process economics (Figure 4). Using a life cycle assessment, our first estimation of CO₂ avoidance costs is below 30 USD per tonne of CO₂ eq.

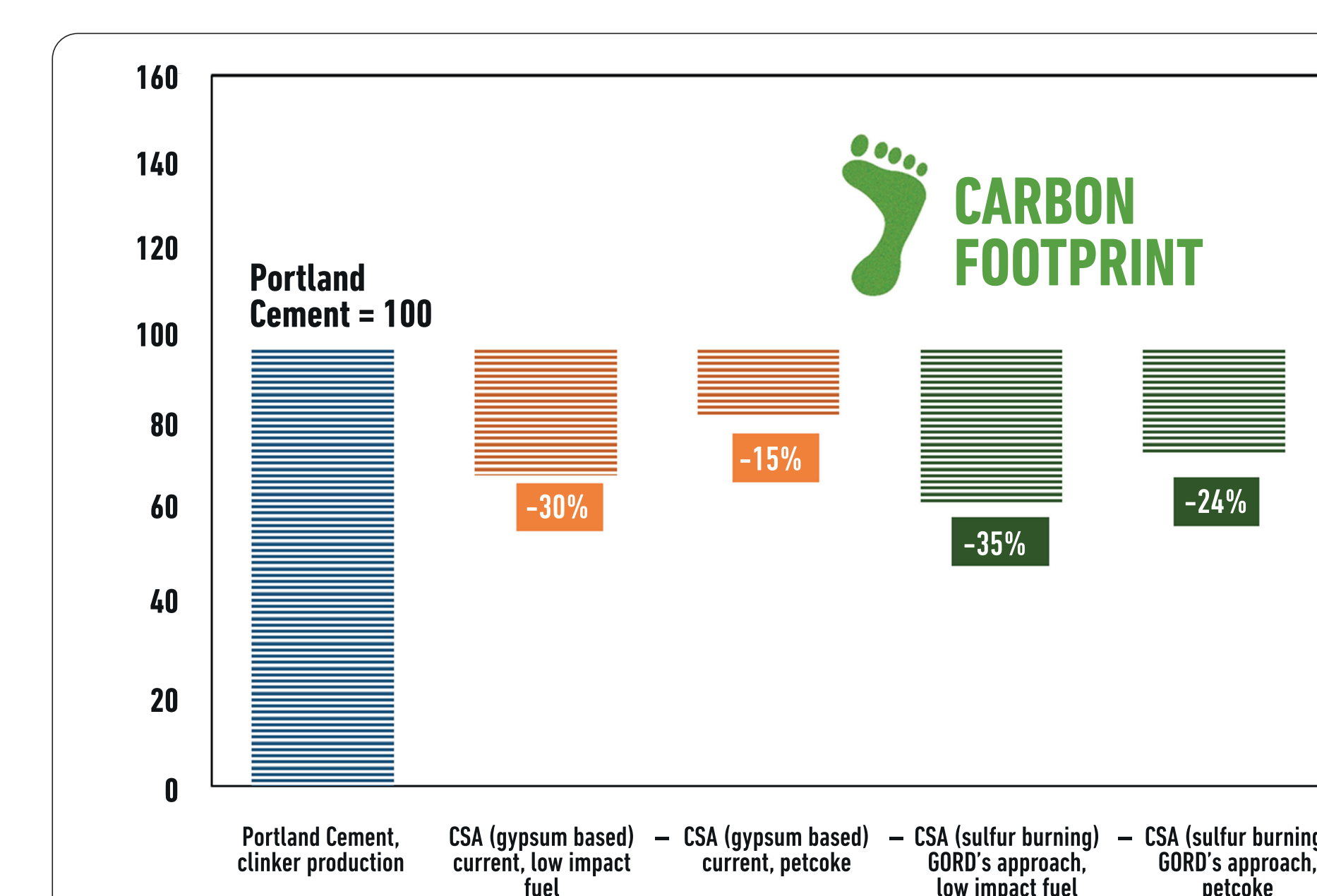


Figure 3: Relative carbon footprint of C\$A cements

Pilot Scale Trials

Following investigations conducted at lab scale for the synthesis C\$A clinker in SO_x containing atmospheres, a series of pilot trials were conducted using a direct fired rotary kiln at IBU-Tec AG (Figure 1). Results from the latest pilot trials in August 2018 (Table 1) showed successful control of kiln conditions which achieved satisfactory SO₃ sequestration into the clinker solids with minimal formation of residual anhydrite. The compressive strengths of the produced mortar are also shown in Figure 2.

	C ₄ A ₃ S	C ₂ S	C ₂ AS	CS	Minor Phases
Avg	42.53 (40.00)	41.53 (48.88)	7.66 (0)	3.15 (0)	5.13 (11.13)

Table 1: Average clinker composition for samples that were "in spec" with the target values predicted by thermodynamics (in parenthesis) for the final pilot trials.



Figure 1: The KDO kiln at IBU-Tec AG was used for the pilot trial series.

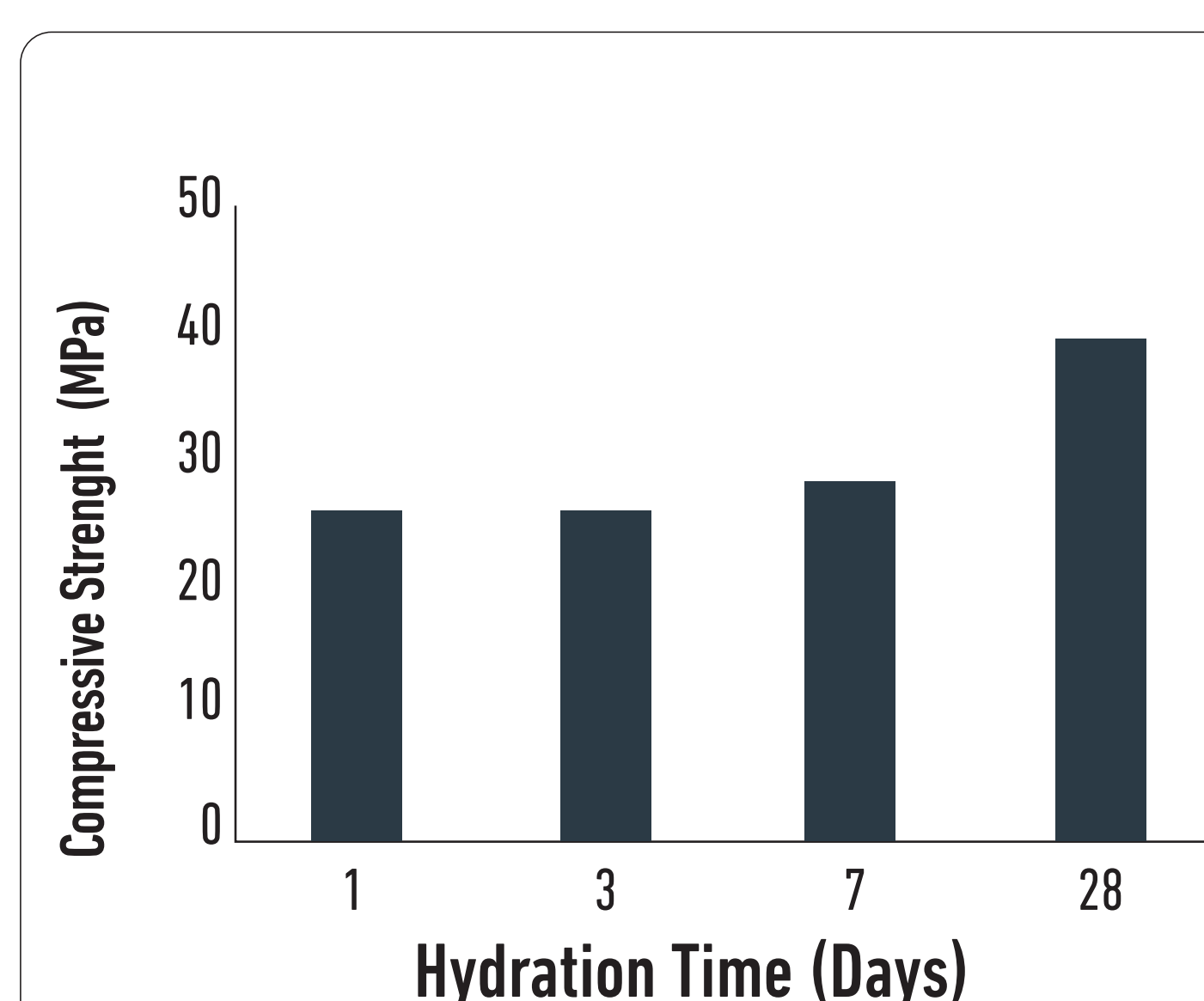


Figure 2: Mortars were prepared using 26% added gypsum and 0.5 w/c.

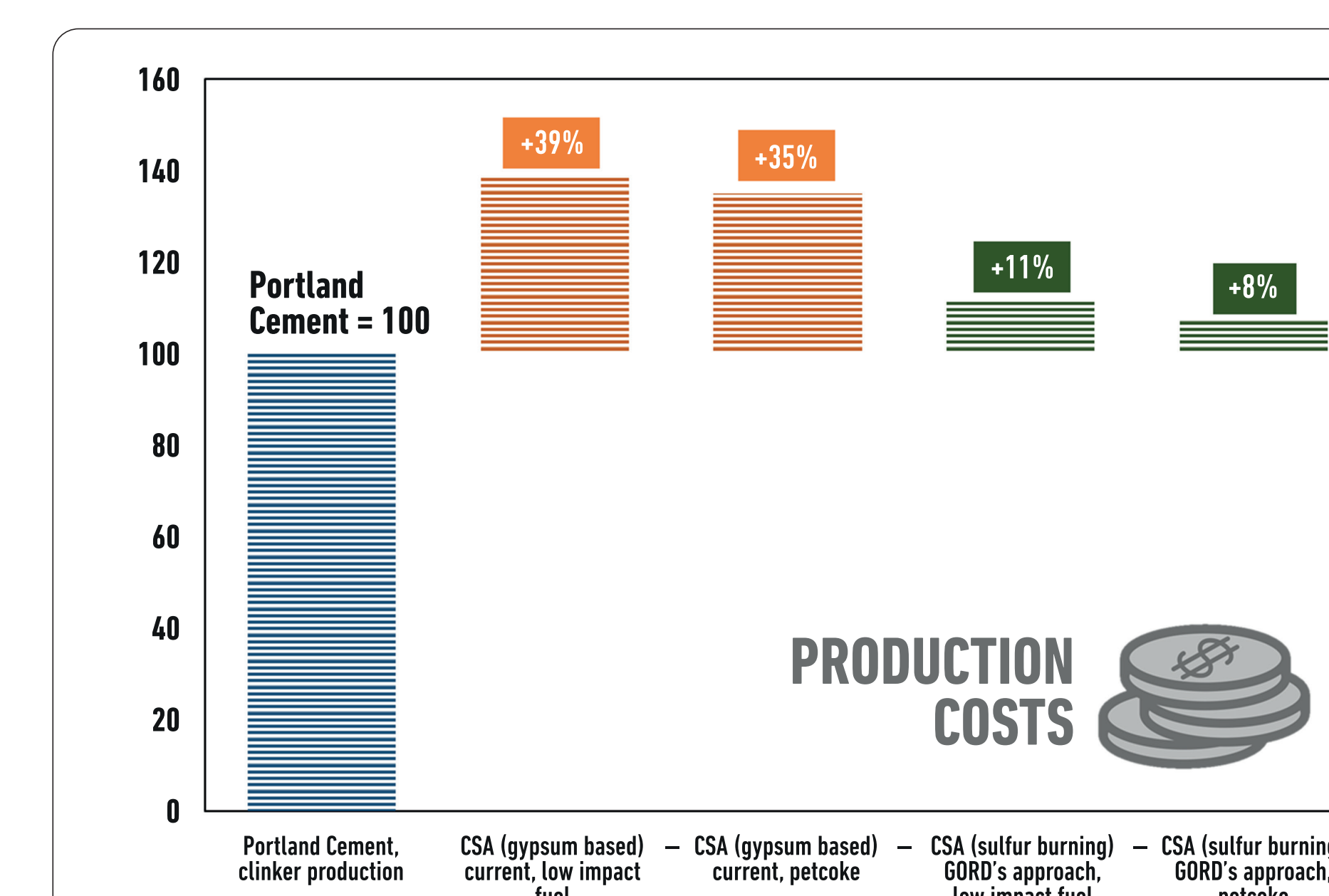
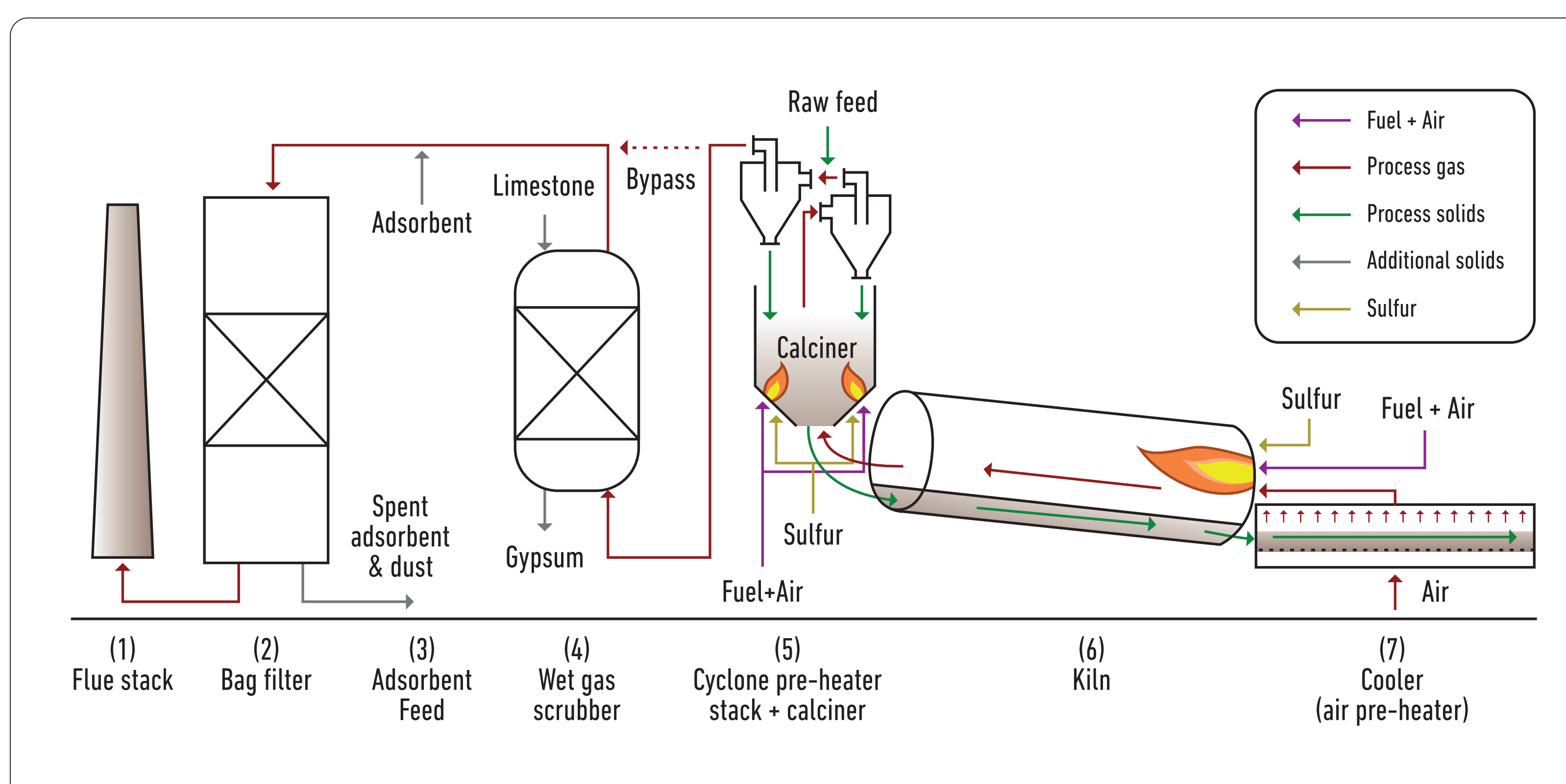


Figure 4: Relative production costs of C\$A cements

Key Process Technology



- The technology used here utilises sulfur combustion in the kiln (6) to produce a **protective atmosphere preventing volatilisation of SO₃** from the clinker phase, ensuring ye'elinite does not decompose.
- Additional sulfur combustion in the calciner (5) can significantly **offset fuel demand**, even producing excess SO_x which is sequestered via a scrubber (4) in gypsum that is required to control setting time.
- The technology is **compatible with cheap sulfurous fuels** such as coal or sour crude oil as well as waste H₂S from the Oil & Gas refineries saving the expense/energy of operating a Claus process to convert the H₂S to solid sulfur for transport/storage.

Acknowledgements

GORD and its collaborators would like to acknowledge the Green Concrete Project team for their efforts towards the development of this work. The authors would also like to acknowledge the Qatar Science and Technology Park, Kuwait National Petroleum Company and Holcim AG for their support during the pilot scale trials.

References

United States Geological Survey, Mineral Commodity Summaries (2001–2018)

Olivier, Jos G.J., K. M. Schure, and J. A. H. W. Peters. "Trends in global CO₂ and total greenhouse gas emissions." PBL Netherlands Environmental Assessment Agency (2017)