

The effect of CO₂ phase on drainage process by analysis of transient differential pressure

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Abstract

CO₂ sequestered in deep geological formations can be in gaseous, supercritical or liquid state, depending on subsurface pressure and temperature. In this work, CO₂ core flooding experiments are carried to investigate the effect of CO₂ phase on drainage process, especially parameters, such as capillary displacement pressure, relative permeability and displacement efficiency. The results indicate that CO₂ phase significantly affect their breakthrough. The breakthrough took place with the least volume of injected liquid CO₂ (LCO₂), and with the largest volume of gas CO₂ (gCO₂). The capillary displacement pressure can be measured based on the jumps in the pressure profiles, and it shows the largest jump in gCO₂ drainage. The relative permeability is the largest in LCO₂-water displacement and the smallest in gCO₂-water displacement. The displacement efficiency can be improved by increasing capillary number when it is smaller than a critical value around 2×10^{-8} .

Keywords: CO₂ phase; CO₂ breakthrough time; capillary displacement pressure; relative permeability; water recovery

1. Introduction

CO₂ storage in reliable reservoirs, aquifers and unminable coal seams has been seen as a promising technology to reduce CO₂ emissions into atmosphere (Safi et al., 2016; Shi and Durucan, 2005). As the pressure and temperature in these formations vary with the depth of the reservoirs, the stored CO₂ can be in different phases (Saraji et al., 2014). In formations with a temperature lower than 30.98 °C, CO₂ can be in gaseous state (gCO₂) e.g., Alabama Black Warrior Basin ~ 70 bar, 23 °C, or in liquid state (LCO₂), for instance West Sak reservoir ~ 110-125 bar, 24 °C. While in a formation with a temperature higher than 30.98 °C, CO₂ can be in supercritical state (sCO₂), for instance, Weyburn oil field ~ 140 bar, 59 °C (Espinoza and Santamarina, 2010; Li et al., 2005; Saraji et al., 2014).

The process of injection, migration and storage of CO₂ in reservoirs, aquifers or unminable coal seams are significantly affected by the interfacial tension (IFT) and contact angle (CA) in the fluid-CO₂-mineral systems, fluid property, physical and chemical properties of pore surfaces, and pore structure, etc. (Sarmadivaleh et al., 2015). Previous research has demonstrated that the interfacial tension and contact angle of CO₂ on minerals in subsurface vary significantly with CO₂ phase, temperature and pressure (Bikkina, 2011; Espinoza and Santamarina, 2010). Chi et al. found that water-CO₂ contact angle on coal surface changed from 84 °C (hydrophilic) to 145 °C (hydrophobic) when CO₂ varied from gas phase to liquid phase (Chi et al., 1988). Jung et al. (Jung and Wan, 2012) observed a sudden increase in water-CO₂ contact angle on silica surfaces when CO₂ phase changed from gas to supercritical state. Similar phenomenon was observed from the measurement of CO₂-water contact angle in single capillaries. Li et al. (Li and Fan, 2015) experimentally demonstrated that the CO₂-fluid contact angles vary considerably with CO₂ phase in oil-wet pores, but not in water-wet pores. Besides contact angles, intensive investigations focus on the effect of CO₂ phase on interfacial tension between CO₂ and water/Brine under various temperature and pressure. Espinoza et al. and Bachu et al. have demonstrated that interfacial tension between CO₂ and water decreases greatly with increase of CO₂ pressure at room temperature (298 K), and remains almost constant at the CO₂ vapour-liquid boundary (Bachu and Bennion, 2008; Espinoza and Santamarina, 2010). This trend has

been further confirmed by molecular dynamics computations (Iglauer et al., 2012). Therefore, CO₂ phase could have a notable influence on capillary entry pressure, displacement pressure, differential pressure, displacement efficiency, relative permeability and storage capacity, etc. (Krevor et al., 2012; Riazi et al., 2011).

Numerous research work has been designed to investigate CO₂ breakthrough (BT) time, capillary pressure, relative permeability and sub-core scale heterogeneity, etc. (Hildenbrand et al., 2002; Perrin and Benson, 2010; Wang and Alvarado, 2016). In CO₂ dynamic displacement, CO₂ breakthrough time, usually represented by the amount of CO₂ injected, is an important parameter and has significant effect on displacement efficiency, the calibration of real time production profile and the determination of relative permeability (Cao et al., 2014). For example, an early gas breakthrough at CO₂ enhanced oil recovery (EOR) could lead to poor sweep efficiency and the reduction of oil production (Gozalpour et al., 2005). In traditional measurements, breakthrough time is usually determined by the turning point on the differential pressure profiles as the production of the first-oil or the first-water in displacement experiment is invisible (Liang et al., 2016). However, it is sometimes hard to find the turning point under a low injection rate or in a high permeability core with little differential pressure change (Liang et al., 2016), and it is hard to determine the breakthrough time in a heterogeneous core with a complex differential pressure (Perrin and Benson, 2010). The method frequently used to determine the breakthrough time is CT scan through visualising changes in situ saturation distribution during core flooding experiments (Perrin et al., 2009). Shi et al. observed a CO₂ breakthrough by CT images after 0.42 pore volume (PV) of supercritical CO₂ was injected into a heterogeneous water-saturated Tako sandstone (Shi et al., 2011). Liang et al. designed a two phase detector probe connected to outflow endcap of core holder to identify the CO₂ breakthrough time for oil-water unsteady-state displacement experiments (Liang et al., 2016). Once the CO₂ breakthrough time is determined, the relative permeability of CO₂ and water/oil in the unsteady state displacement can be estimated through JBN method based on Buckley-Leverett theory (Buckley and Leverett, 1942; Johnson et al., 1959). JBN method is more efficient than steady method to get relative permeability curve with

continuous saturation distribution. Steady method can get only one point each time (Liu et al., 2010). Xu et al. used the JBN method to predict the relative permeability of $s\text{CO}_2$ and water in the process of injecting $s\text{CO}_2$ into porous media containing water. The predicted results showed a good agreement with the experimental data (Xu et al., 2011). Liu et al. modified JBN method by considering capillary pressure gradient to calculate the relative permeability of oil and water for particular low permeability reservoirs. Their results showed that the modified JBN method can predict the relative permeability more accurately than empirical formula for extra low permeability Reservoir (Liu et al., 2010). Chen et al. extended the JBN method to obtain relative permeabilities of CO_2 and brine at each pressure tap of the core (not just at the core outlet) by measuring the section-wise pressure drop, local saturation and local phase fluxes (Chen et al., 2016).

However, most research work has focused on single phase of CO_2 , especially supercritical CO_2 in core flooding experiments. For example, Tokunaga et al. measured the drainage and imbibition capillary pressure for supercritical CO_2 and brine in quartz sand and found that wettability of silica surfaces was influenced by the exposure time to $s\text{CO}_2$ (Tokunaga et al., 2013). Parvazdavani et al. showed that the relative permeability of $g\text{CO}_2$ and oil calculated by JBN method was similar to the results predicted by sophisticated inverse modelling method when the sandstone was highly fractured (Parvazdavani et al., 2017). Though the effect of CO_2 phase on the differential pressure and water production in a core flooding experiment was studied by Ebraheam et al. (Al-Zaidi et al., 2018), the understanding of its impact on the drainage process, especially essential parameters, such as CO_2 breakthrough time, capillary displacement pressure, relative permeability and displacement efficiency, is still limited.

In this study, unsteady-state drainage experiments in a sandstone core sample was designed to investigate the effect of CO_2 phase on the drainage process, especially flow parameters, such as CO_2 breakthrough time, capillary displacement pressure and relative permeability. A simple and effective method was proposed to determine the CO_2 breakthrough through placing a membrane at the outlet of the core and the breakthrough point was found as the turning point

on differential pressure curve. The impact of CO₂ phase on the capillary displacement pressure was studied by comparing the pressure jumps on the pressure profiles. The effect of CO₂ phase on the relative permeability of water displacement were measured and compared with JBN method. The uncertainties of relative permeabilities were also assessed. The relevance between water recovery and capillary number was finally examined to investigate the effects of viscous and capillary forces on the displacement efficiency.

2. Materials and experimental setup

Sandstone core samples were provided by Anasuria Operating Company, UK. Double deionized water was used with an electrical resistivity greater than 18.18 MΩ·cm. Dry CO₂ was supplied by BOC with a guaranteed minimum purity level 99.995% in a cylinder fitted with a dip tube to permit withdrawal of liquid phase of CO₂.

2.1. Core sample

The core plug sample used for CO₂ flooding experiment has a diameter of 2.3 cm and a length of 3.7 cm and an average porosity of 0.23. The absolute permeability of the core sample was measured by DI water under different confining pressure and at a flowrate range from 0.05 to 0.4 mL/min, as shown in Fig. 1. The results are fitted linearly and the permeability can be derived from the slope by Darcy's law, which are 152, 138 and 129 mD (1 mD = 10⁻¹⁵ m²) under confining pressures of 1000, 1500 and 2000 PSI respectively. The reduction of permeability under high confining pressure could be caused by the closure of the elliptic pores rather than the presence of fractures (Shi et al., 2011). All other experiments were conducted at the same confining pressure (2000 PSI) to keep a consistent permeability of the core (129 mD). The wettability of the core sample was measured with a Drop Shape analyser (DSA 100). 10 μL of ultrapure water with 18.2 MΩ·cm was placed on the surface of the air-saturated (dry) core at ~ 20 °C by a pipette and then a photographic digital image of the droplet on the surface was taken by DSA 100. After setting the baseline, the air-liquid-rock contact angle calculated was around 77 ° by the software, which showed that the core sample was hydrophilic. In Li's work,

the CO₂-fluid contact angles were independent of CO₂ phase in a water-wet pore (Li and Fan, 2015).

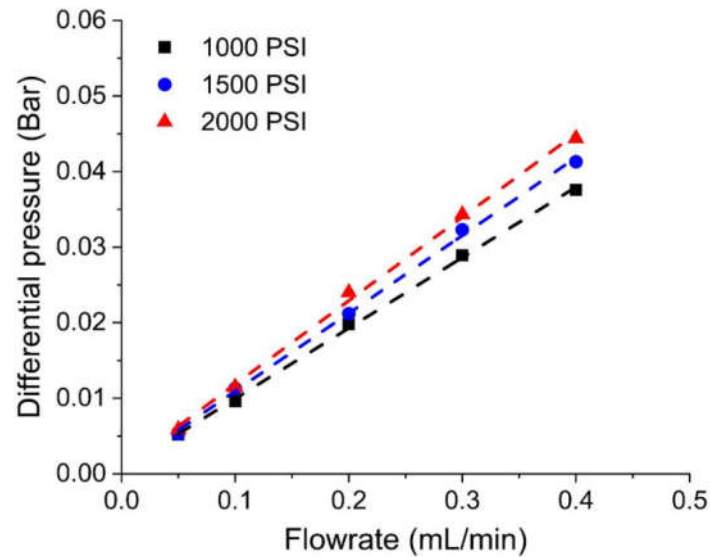


Fig. 1. The differential pressure versus flowrate under different confining pressures.

2.2. Experimental setup

An unsteady-state core flooding system was designed for CO₂-water dynamic displacement experiments, which is shown in Fig. 2. The system is consisted of two high-pressure (up to 695 bar) syringe pumps (ISCO syringe pump, 100DM) with a flowrate ranging from 0.00001 to 30 ml/min for CO₂ injection and water collection, an overburden-pressure water pump (Milton Roy, CM4000) to create the confining pressure, a vacuum pump to remove trapped gas, a core holder with a shrinkable Teflon sleeve to place the core plug and a hot water bath (Grant Instruments, GD100) to control the temperature with a precision of ± 0.02 °C. A LabVIEW software was built to automatically record the data of inlet and outlet pressure which was measured by two pressure transducers (UNIK, 100 bar gauge with a precision of ± 0.001 bar).

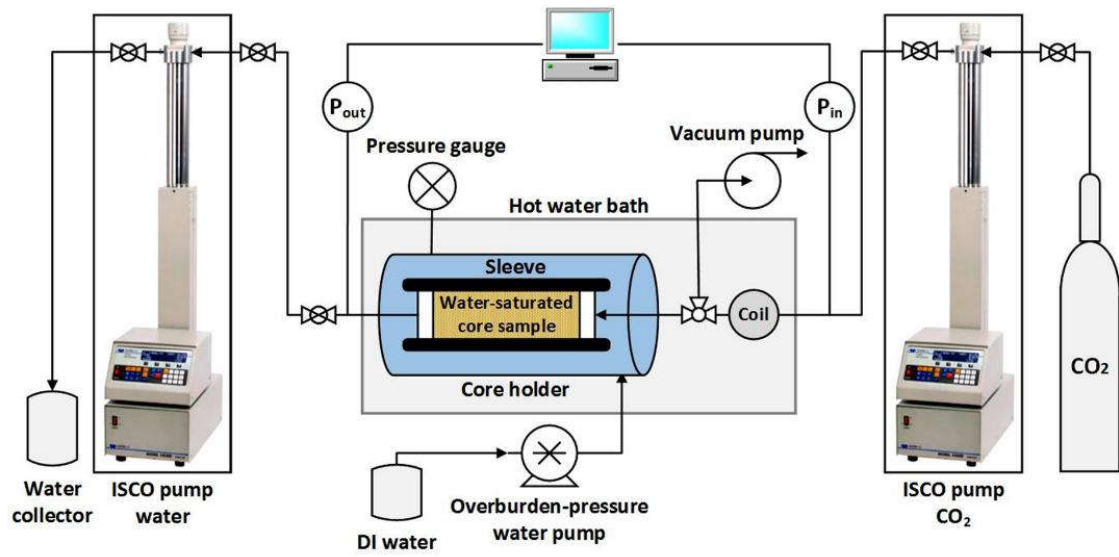


Fig. 2. Schematic diagram of CO₂ core flooding system.

Table 1. Operating conditions of the experiments.

Pressure (Bar)	Temperature (°C)	Phase state	Flowrate (mL/min)	IFT of CO ₂ and water (mN/m)	Viscosity ratio (CO ₂ /water)	Capillary number $N_{ca}(\times 10^{-8})$
45	17	Gas	0.1-0.3	40.4	0.015	0.7-2.1
80	40	Supercritical	0.1-0.3	31.3	0.035	1.4-4.3
80	22	Liquid	0.1-0.3	29.8	0.075	4.3-12.8

2.3. CO₂-water displacement procedure

Experiments were conducted under various reservoir conditions where CO₂ can be in different phases. CO₂ injection flowrate was in a range from 0.1 to 0.3 mL/min, which correspond to a superficial flow velocity of 4 to 12 μm/s. For each experiment, the injection flowrate was constant. The experimental conditions are presented in table 1. The interfacial tension between water and CO₂ is obtained from literature (Bachu and Bennion, 2008) and the viscosities of water and CO₂ are available from NIST CHEMISTRY Webbook website (Chase Jr and Tables, 1998).

The core plug was pre-saturated with pure water and then wrapped in a shrinkable rubber sleeve and placed in a horizontal core holder. The core holder was immersed in the hot water bath and kept at a targeted temperature for several hours to achieve the thermal equilibrium. The confining pressure, 13.79 MPa (2000 psi), which was always larger than pore pressure, was firstly built by the overburden-pressure water pump to ensure the tight fitting between the rubber sleeve and the core sample, and thus to prevent fluid bypassing through the space between the rubber sleeve and the outside of the core (Shiraki and Dunn, 2000). Prior to the core flooding experiment, at least 30 pore volume of pure water was injected by syringe pump into the core at a high flowrate of 20 mL/min to remove the air in the system and to dissolve any CO₂ present in the core, and thus to ensure a complete saturation of the core with water (Kyte et al., 1956; Narayanan et al., 2018). Some more water was injected into the core to build the back pressure. After removing the trapped air by the vacuum pump, CO₂ was injected at the targeted pressure by the syringe pump (the CO₂ injecting pump in Fig. 2) into the inlet pipe. A coil placed at the inlet was to ensure that CO₂ was achieved thermal balance before it entered into the core sample. Keeping the system balanced at the targeted pressure for an hour, the core flooding experiment was then started by switching the CO₂ injection syringe pump from constant pressure to constant flowrate. During the experiment, the transient data of inlet pressure, outlet pressure and production of water and CO₂ was carefully monitored. The inlet and outlet pressures were measured by two pressure transducers and automatically recorded every 6 seconds, by the LabVIEW software, which were later used to calculate the differential pressure across the core sample. Another water syringe pump was used to collect the production of water and CO₂. The cumulative production volume of water and CO₂ was recorded every minute. Each experiment was terminated after injecting about 2 pore volume of CO₂. After each experiment, the weight of the core was measured using the Sartorius weighing scale with a resolution of 0.0001 g to calculate the residual water saturation.

2.4. Baseline of the differential pressure

In order to have a clear observation of the evolution of the differential pressure when CO₂ started to flow within core sample, around 0.7 mL water was kept in the inlet prior to CO₂

injection into the core. Before CO₂ entered the core, the differential pressure was dominated by single phase flow of water in the core and its value corresponded to viscous pressure loss of water flow. Once CO₂ arrived at the inlet of the core, the pressure in injected CO₂ built up to overcome the capillary pressure caused by the interfacial tension between water and CO₂. Pressure jumps were observed in all the CO₂-water displacement experiments regardless of what phase CO₂ was, which indicates the value of capillary displacement pressure as discussed in section 4.2.2.

3. Results

3.1 Differential pressure

During the CO₂-water displacement experiments, the inlet and outlet pressure of the core sample were recorded every 6 seconds. The temporal resolution has been proved to be sufficient to capture differences in fluid front configurations in dynamic flow experiments, such as Haines jumps (Berg et al., 2013). The differential pressures of drainage processes by different CO₂ phases are shown in Fig. 3. At injection flowrate ranging from 0.1 to 0.3 mL/min, the drainage process is a typical of slow capillary-dominated drainage with intermittent fluctuations (Primkulov et al., 2019). At the first stage of the experiments, the differential pressure is determined by water flow in the core sample as water prior to the inlet was pushed into the core sample by CO₂. This can be verified by the small pressure variation at the beginning of the profiles in Fig. 3(a)-(c). When CO₂ arrives at the inlet of the core sample, the pressures suddenly built up to overcome the capillary force caused by the CO₂-water interfacial tension. After CO₂ enters into the core sample, the pressure is dominated by evolution of CO₂-water interface, that is, cooperative pore filling events (like burst, touch or overlap) defined by Cieplak & Robbins (Cieplak and Robbins, 1988), resulting in the fluctuations on the pressure profiles.

It is worth to note that injection flowrate of CO₂ set by the injection pump is sometimes different from the CO₂ flowrate in the core sample as the temperature in the pump differs from that in the water bath. This leads to a variation in CO₂ density and CO₂ phase. For example, to obtain a CO₂ flowrate of 0.1 mL/min in the core at 40 °C and 80 Bar, the CO₂ flowrate in the

pump at room temperature and the same pressure needs to be set as $Q_p = \frac{\rho_{CO_2}^{40^\circ C \ 80 Bar}}{\rho_{CO_2}^{RT \ 80 Bar}} \times 0.1$ mL/min.

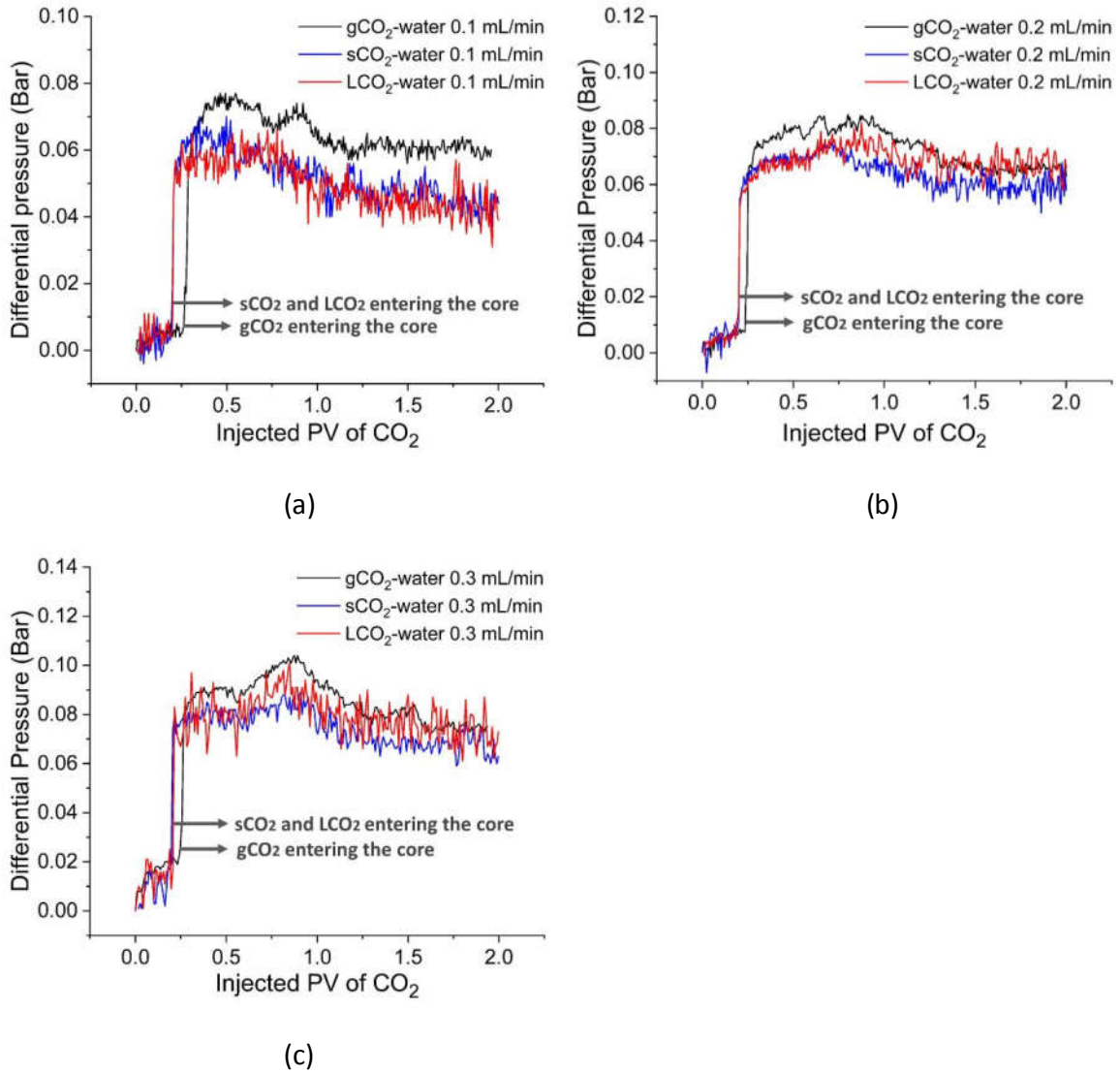
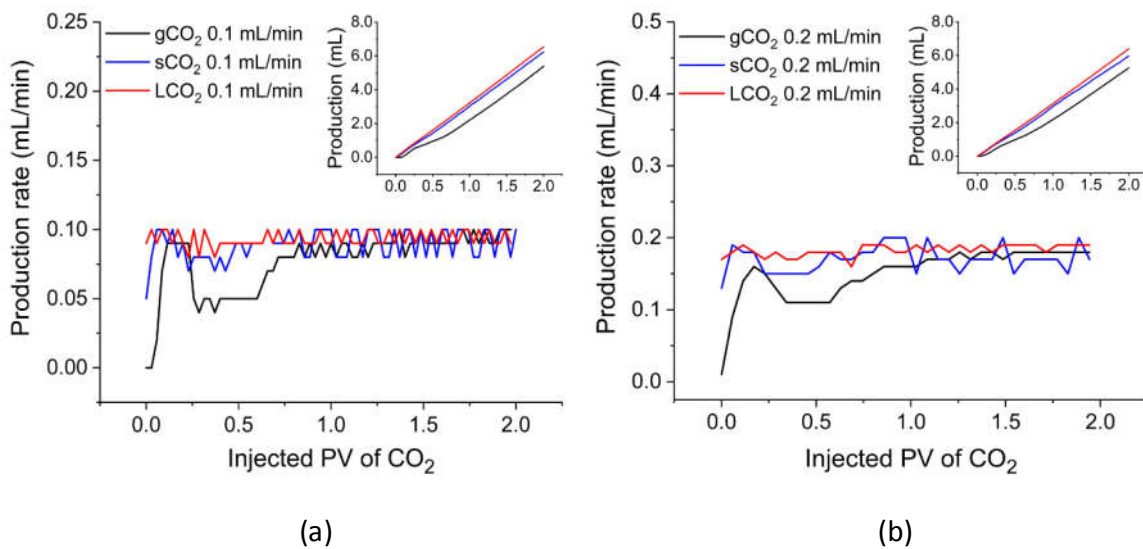
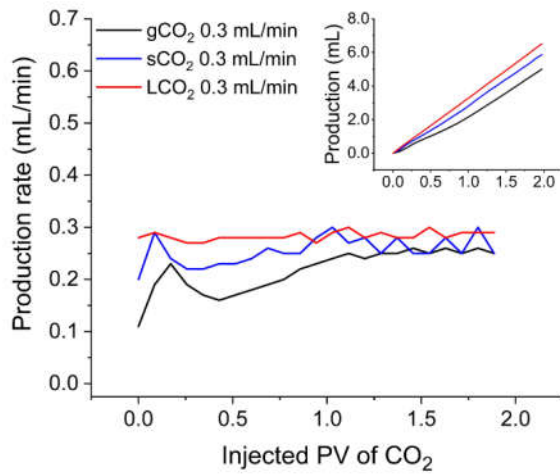


Fig. 3. Differential pressure across the core sample for the displacement of water by gCO₂, sCO₂ and LCO₂ at an injection flowrate of (a) 0.1 mL/min, (b) 0.2 mL/min and (3) 0.3 mL/min.

3.2 CO₂ and water production

CO₂ and water production were collected by the pump at the outlet of the core sample every minute. The effect of CO₂ phases on their production rate and cumulative production is presented in Fig. 4(a)-(c). The results indicate that the production rate and cumulative production vary significantly with CO₂ phases. For LCO₂-water displacement, the collective production rate is very close to the CO₂ injection flowrates and the production is always higher than that in gCO₂, sCO₂ injection processes. For gCO₂-water displacement, the production rate varies greatly during the injection process. Before CO₂ entering the core sample, the production rate increases quickly to the given injection flowrate. When CO₂ starts to enter the core, the production rate decreases rapidly as part of injected gas CO₂ has to be compressed to build up the pressure. After the injection of around 1 PV CO₂, the production starts to be stabilized and close to the given injection flowrate. The cumulative production during gCO₂ injection is the smallest. The production rate and cumulative production of sCO₂ displacement varies between LCO₂ and gCO₂ displacement as its density, viscosity and compressibility, are between LCO₂ and gCO₂.





(c)

Fig. 4. Effect of CO₂ phase on water production rate and corresponding cumulative production (inset) during drainage process at an injection flowrate of (a) 0.1 mL/min, (b) 0.2 mL/min and (c) 0.3 mL/min.

4. Discussions

4.1. Determination of breakthrough time

An accurate identification of CO₂ breakthrough time is a first-line step for further determining parameters such as water/CO₂ fractional flow, relative permeability, and displacement efficiency. Though there is a straight and visualized way to observe CO₂ breakthrough by CT images, the core flooding rig equipped with CT is more complicated and expensive. An economic and convenient method present in this paper to determine the CO₂ breakthrough time is through placing a membrane (Nuclepore Track-Etch Membrane-Whatman) with a low permeability of ~ 0.01 mD at the outlet of the core sample. The breakthrough time can be easily seen from the differential profile of the core sample with a membrane.

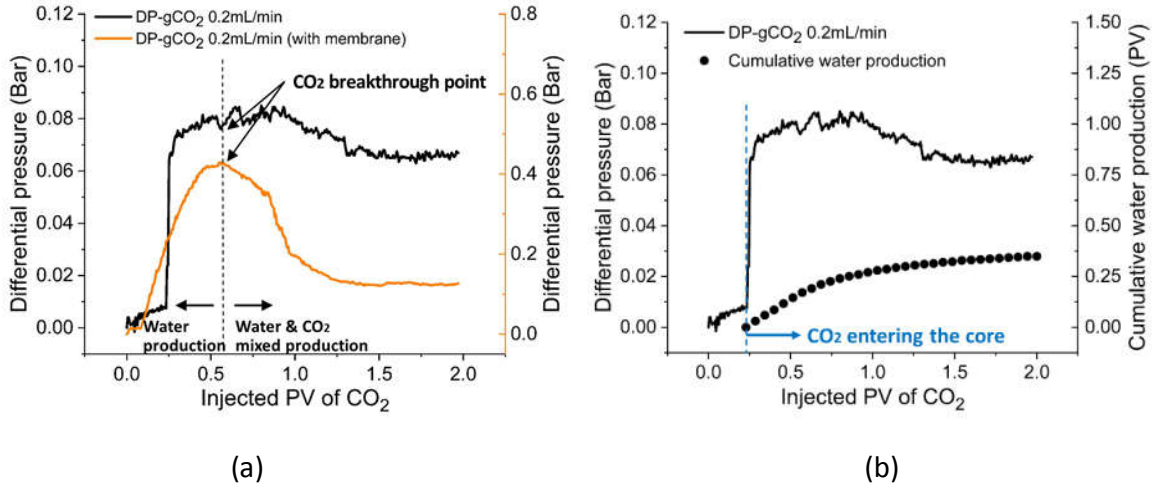


Fig. 5. Take gas CO₂-water displacement at injection flowrate of 0.2 mL/min as an example to determine (a) CO₂ breakthrough time and (b) cumulative water production.

Taking gas CO₂-water displacement with 0.2 mL/min injection flowrate as an example (Fig. 5(a)), the black line is the original differential pressure across the core sample without a membrane and the yellow line is the differential pressure with a membrane. The application of the membrane gives very different differential pressure profiles for the flooding process. After the application of a membrane, the magnitude of the differential pressure is much larger than that in the original core sample, and there is a maximum value in the pressure profile. The time corresponding to the maximum value is the breakthrough time as shown in Fig. 5(a). This is due to the difference in the viscosity of water and CO₂.

The total differential pressure (ΔP_T) along the core sample and membrane is the sum of capillary pressure (P_c) and viscous pressure drop (ΔP_v), i.e.,

$$\Delta P_T = P_c + \Delta P_v \quad (1)$$

As the change of the capillary pressure is not obvious during the displacement in the core sample, the significant change in the differential profile for the core sample with a membrane (the yellow line in Fig. 5(a)) is contributed by the viscosity difference between water and CO₂. As the water viscosity is much larger than CO₂ viscosity, the viscous pressure drop of water through the membrane is much larger than CO₂. Once CO₂ reaches a breakthrough, CO₂

immediately enters into the membrane, and the pressure drop caused by viscosity drops significantly.

Applying Leverett extension of Darcy's law to multiphase flow in porous media, the pressure for each phase ($i=n, w$) is (Ramakrishnan and Chugunov, 2013),

$$\frac{\partial P_i}{\partial x} = - \frac{\mu_i Q_i}{K K_{ri} A} \quad (2)$$

where P , μ , Q and Kr are the pressure, viscosity, volume flow rate and relative permeability for phase i . K , A and x are the intrinsic permeability, cross-sectional area and length of the porous media. The viscous pressure loss of each phase flowing across the membrane is approximately 50 times larger than that in core sample based on Eq. (2). The variation of the differential pressure is therefore largely dominated by the water and CO₂ flow through the membrane. The viscosity ratio between water and CO₂ varies significantly with CO₂ phase. The viscosity of water is 60, 30, 15 times larger than that of gCO₂, sCO₂ and LCO₂ respectively. The viscous pressure drop caused by water is much larger than CO₂. Once CO₂ enters into the membrane, the differential pressure across the membrane decreases as shown in Fig. 5(a). CO₂ breakthrough time thus can be clearly determined by the turning point on the differential pressure profile with a membrane. Among many factors, CO₂ breakthrough time depends critically on the relative permeability of each phase which relates to the distribution of permeability within a reservoir, injection flowrate and viscosity ratio of invading phase to displaced phase, etc. (Jessen et al., 2005). The variation of injection flowrate in this experiment hardly affected CO₂ breakthrough time and CO₂ breakthrough happened after 0.3, 0.23 and 0.19 pore volume of CO₂ was injected for gCO₂, sCO₂ and LCO₂-water displacement experiment, respectively.

Before CO₂ breakthrough, only water production was collected from the outlet, and after that, both CO₂ and water were collected until the equilibrium (steady-state) period when only CO₂ was collected as the water saturation in the core sample become irreducible. The water production before CO₂ breakthrough can be calculated from the increase volume of the receiving pump. After CO₂ breakthrough, the water production rate $v(t)$ can be correlated by the power-law equation (Toth et al., 2001; Toth et al., 2002),

$$v(t) = at^b \quad t \geq t_{BT} \quad (3)$$

where t is dimensionless time, a and b are constant parameters that can be determined by calculating the water production rate at the CO₂ breakthrough point and the integral of water production rate from CO₂ breakthrough to the end. The cumulative water production then can be calculated as shown in Fig. 5(b), which indicated most water production happened at the initial stage of CO₂ injection and it almost stabilized after injecting 1.5 PV of CO₂.

4.2. Effect of CO₂ phase on differential pressure

4.2.1. Pressure fluctuation

The differential pressure profiles are split into the general trend (red line) and fluctuation components (blue line) by wavelet decomposition, as shown in Fig.6 (a). The fluctuation has been explained by Bauyrzhan et al. (Primkulov et al., 2019). It is due to the reconfigurations of the CO₂-water interface when invading phase (CO₂) enters into the pores. However, their explanations are based on the numerical simulation, and it might be different from experiment. For example, before CO₂ enters into the core sample, some pressure fluctuation was also observed from the profiles. This may be caused by the running of the pump and intrinsic error of pressure transducer. The standard deviation of pressure fluctuation of CO₂-water flow has been presented in Fig. 6(b) and it shows that fluctuation is almost independent of injection flowrate but related to the CO₂ phase. LCO₂-water displacement has the largest pressure fluctuation while gCO₂-water displacement gives the smallest one. To invade the pores and push out of the defending fluid, the pressure needs to overcome the capillary pressure at the pore throat, which is controlled by Young-Laplace equation. The general form of the equation is,

$$P_c = \sigma_{CO_2,w} \times \left(\frac{1}{a} + \frac{1}{h} \right) \quad (4)$$

where $\sigma_{CO_2,w}$ is the IFT between CO₂ and water, a is the width of the pore throat and h is the height of the pore. During the displacement, the variation of P_c was dominated by the invaded pore size. Take the variation of P_c with a and h ,

$$\delta P_c = \frac{\sigma_{CO_2,w}}{a^2} \delta a + \frac{\sigma_{CO_2,w}}{h^2} \delta h \quad (5)$$

Eq. (5) indicates that the pressure fluctuation is related to the interfacial tension between two fluids and the variation of a and h . A higher mobility ratio (μ_{CO_2}/μ_{water}) usually favours

displacement, which means CO₂ can invade in more pores with smaller size (Lenormand et al., 1988). The mobility ratio of LCO₂-water displacement is much larger than gCO₂, sCO₂-water displacement, which therefore shows a larger fluctuation in differential pressure profiles.

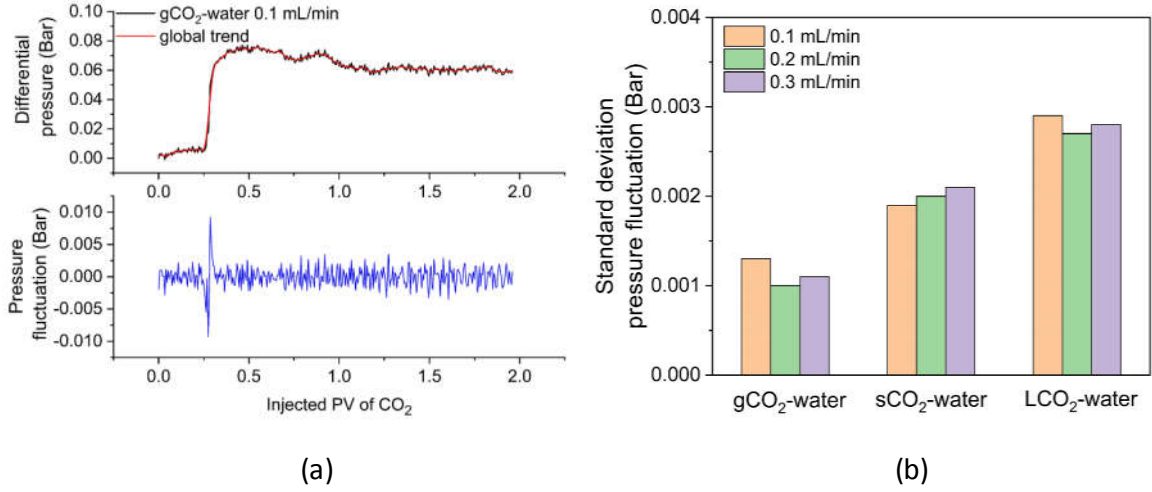


Fig. 6. Wavelet decomposition analysis of the differential pressure: (a) splitting the pressure profile into global trend (red line) and fluctuation (blue line) and (b) standard deviation of pressure fluctuation for gCO₂, sCO₂ and LCO₂-water displacement at different injection flowrate.

4.2.2. Capillary displacement pressure

As mentioned in section 2.4, pressure jumps were observed in all CO₂-water displacement experiments. The values of pressure jumps are shown in Fig. 7, which are clearly independent on injection flowrate, but dependent on the CO₂ phase. In Egermann's experiment (Egermann et al., 2006), the value of the pressure jump observed at the moment when CO₂ initially entered into the core were close to displacement pressure (P_{cd}) derived from the mercury-injection curve. The capillary displacement pressure in a tube is caused by the curved interface between the non-wetting phase (CO₂) and wetting phase (water). It can be described by the modified Young-Laplace equation (Chalbaud et al., 2007),

$$P_{cd} = P_{CO_2} - P_{water} = \frac{2\sigma_{CO_2,w}\cos\theta}{R_{throat}^{max}} \quad (6)$$

where P_{cd} is the capillary displacement pressure value, $\sigma_{CO_2,w}$ is the IFT between CO₂ and water, θ is the contact angle and R_{throat}^{max} corresponds to the largest connected pore throats.

The IFT between gCO₂ and water is usually larger than that between sCO₂ or LCO₂ and water as shown in Fig. 8, which leads to higher capillary displacement pressure for gCO₂ displacing water according to Eq. (6) and thus requires larger external pressure difference to overcome it. Assuming the core sample as a bundle of capillary channels with the same radius, the relationship between permeability and channel diameter was $K = \phi d^2 / 32\tau$ (Kozeny, 1927), where ϕ is the porosity of the porous media, d is the channel diameter and τ is tortuosity defined as the ratio of actual length of flow path in the porous medium to the length of flow path in the absence of porous medium. Assuming τ equals to unit, the estimated displacement pressure can be expressed as,

$$P_{cd}^{est} = \frac{2\sigma_{CO_2,w}\cos\theta}{\sqrt{2K/\phi}} \quad (7)$$

The estimated values of capillary displacement pressure from Eq. (7) is presented in Fig. 7. They are 28%, 25% and 27% larger than the experiment results in gCO₂-water, sCO₂-water and LCO₂-water displacement, respectively. The overestimation of the displacement pressure is because the pore size in Kozeny model is assumed as uniform, which is smaller than the size of largest pore throats of the core sample.

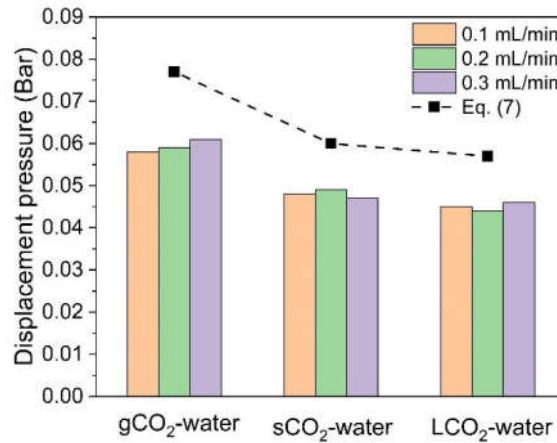


Fig. 7. Values of pressure jumps at the beginning of drainage and estimated capillary displacement pressure by Kozeny model (Kozeny, 1927)

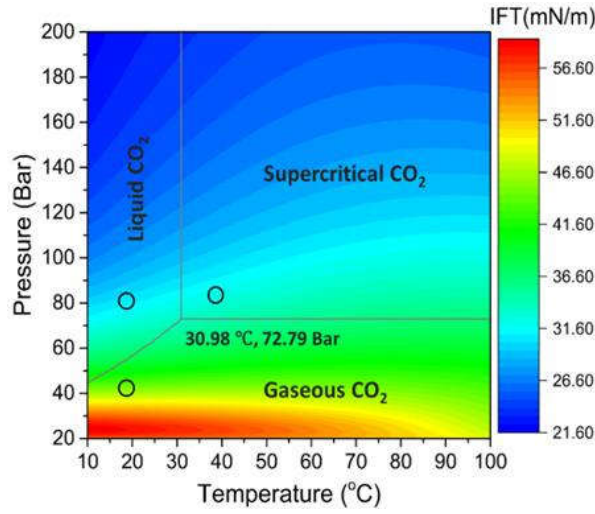


Fig.8. A contour map of CO₂-water IFT as a function of pressure and temperature calculated from equations in literature (Li et al., 2012). Our experimental conditions are shown by the circles.

4.3. Effect of CO₂ phase on the relative permeability

Relative permeability is a vital parameter in subsurface hydrology. It reflects the capability of a specific formation to produce a combination of oil, water or gas more accurately than the absolute permeability under multiphase flow conditions (Chen et al., 2016). The JBN method is a popular method and can be used to quickly determine the relative permeability of the invading phase (CO₂) and the defending phase (water) at the outlet of the core sample across a range of saturations. To apply this method to the unsteady-state drainage experiments, we assume (Johnson et al., 1959): (1) stabilized displacement can be satisfied with high enough flow velocity and (2) incompressible fluids can be obtained by maintaining gas phase at a high enough pressure level (Johnson et al., 1959). During displacement, the differential pressure across the core sample and the phase fractional flow need to be measured. By applying generalized Darcy's law and fractional flow equation to Buckley-Leverett theory (continuity equation) with those assumptions, relative permeability of CO₂ and water and CO₂ saturation at the outlet of the core can be calculated as Eqs. (8)-(11) (Buckley and Leverett, 1942; Johnson et al., 1959),

$$f_{w,2} = \frac{dS_{CO_2,av}}{dt_d} \quad (8)$$

$$K_{rw,2} = \frac{f_{w,2} L Q_i \mu_w}{KA} \frac{d(1/t_d)}{d(\Delta P/t_d)} \quad (9)$$

$$K_{rCO_2,2} = k_{rw,2} \frac{\mu_{CO_2} (1-f_{w,2})}{\mu_w f_{w,2}} \quad (10)$$

$$S_{CO_2,2} = S_{CO_2,avg} - f_{w,2} t_d \quad (11)$$

where subscript 2 denotes the outlet of the core; $f_{w,2}$ is fractional flow of water at the core outlet; $S_{CO_2,av}$ is the average CO₂ saturation of the core; t_d is dimensionless time indicated by injected pore volume; $K_{rw,2}$ and $K_{rCO_2,2}$ represent relative permeability of water and CO₂ at the core outlet respectively; Q_i is volumetric injection flowrate; μ_w is viscosity of water; ΔP is the measured pressure difference across the core; K , A and L are absolute permeability, cross section area and length of the core, respectively; $S_{CO_2,2}$ and $S_{CO_2,av}$ are core outlet and core-average CO₂ saturation, respectively.

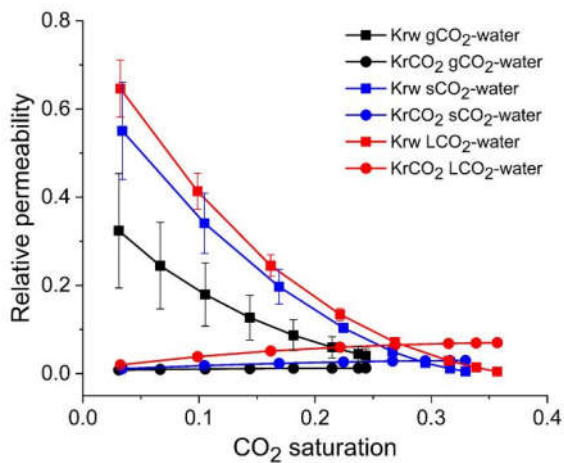
Rapoport (Rapoport and Leas, 1953) suggested a scaling coefficient of $\mu_w L Q_i / A$ ($\frac{cm^2}{min} \cdot cP$) to evaluate the capillary end effect. If the coefficient is larger than 0.1, it is applicable to assume the flow is stabilized. In this experiment, the coefficient ranges from 0.1 to 0.3, stabilised flow conditions can be achieved during the drainage. The incompressible condition assumes the inlet injection flowrate equals to the outlet production flowrate ($Q_i = Q_o$). However, the measured flowrate from outlet is smaller than the injection flowrate, especially in gCO₂-water displacement. Therefore, the uncertainty of relative permeability determined by this method can be assessed as,

$$\delta K_{rw,2} = \frac{f_{w,2} L \mu_w}{KA} \frac{d(1/t_d)}{d(\Delta P/t_d)} \frac{(Q_i - Q_o)}{Q_i} \quad (12)$$

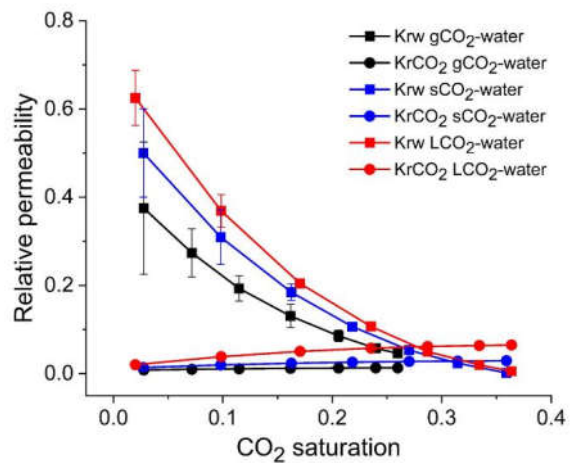
$$\delta K_{rCO_2,2} = \frac{\mu_{CO_2} (1-f_{w,2})}{\mu_w f_{w,2}} \delta K_{rw,2} \quad (13)$$

Relative permeability curves of CO₂ and water are shown in Fig. 9(a)-(c) at an injection flowrate from 0.1 mL/min to 0.3 mL/min. The error bars indicate the uncertainty of the relative permeability caused by the assumption of incompressible fluids. Larger uncertainty appears at smaller CO₂ saturation. gCO₂-water displacement always gives the largest uncertainty because gaseous CO₂ was compressed to build up the capillary pressure at the beginning of displacements. During the drainage process, LCO₂-water displacement always gives the largest

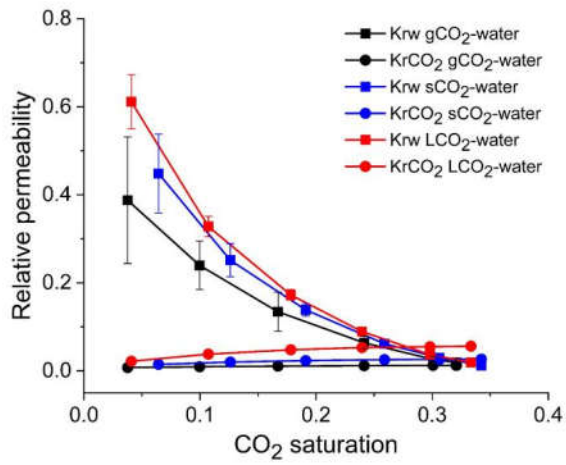
relative permeability for both water and CO₂ while the relative permeability of gCO₂-water displacement is smallest under the same injection flowrate condition. Eq. (9) indicates that the water relative permeability is mainly related to water fractional flow, water viscosity and differential pressure. As pressure difference across a high permeability core in the drainage process changes slightly, its influence on the relative permeability is insignificant (Liang et al., 2016). The differences between the relative permeability of water in different CO₂ phase core flooding is mainly caused by the water fraction flow. Fig. 10(a)-(c) present the fractional flow of water versus CO₂ saturation at various injection flowrate in different CO₂ phase displacements. It can be noted that the value of water fractional flow is greatest in LCO₂-water displacement and lowest in gCO₂-water displacement at the same CO₂ saturation under the same injection flowrate condition. According to Eq. (10), the low CO₂ relative permeability is mainly caused by the large difference between the viscosity of CO₂ and water. Besides, it is found that the injection flowrate hardly affects the relative permeability for both water and CO₂ in different CO₂ phase drainage processes, which is also proved in previous study (Chen and Wood, 2001). Therefore, the relative permeability obtained from the laboratory experiments can be also applied to the reservoir condition where the injection flowrate is smaller.



(a)

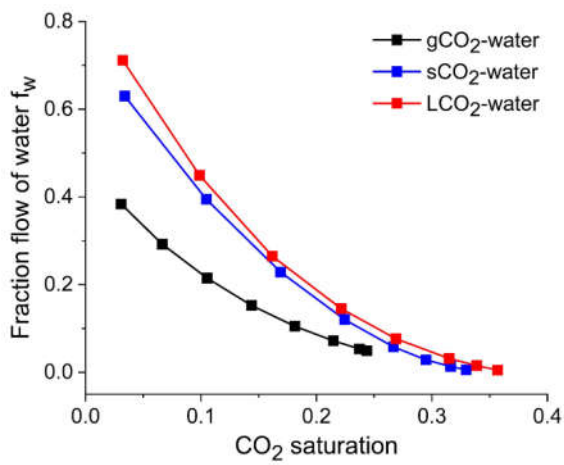


(b)

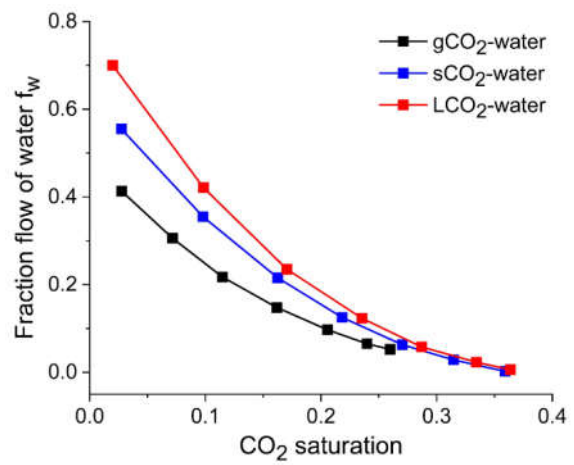


(c)

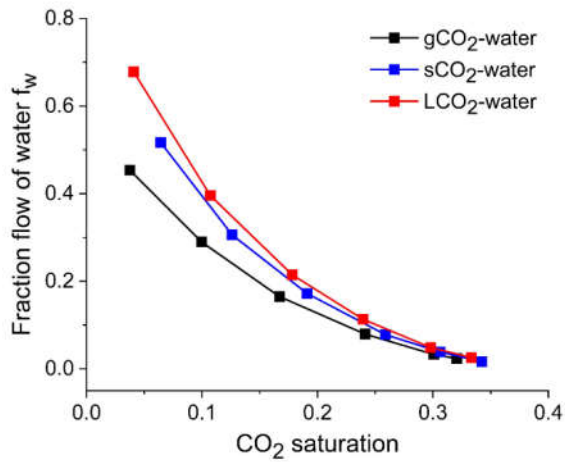
Fig. 9. Effect of CO₂ phase on the relative permeability in displacement of water at injection flowrate of (a) 0.1 mL/min, (b) 0.2 mL/min and (c) 0.3 mL/min.



(a)



(b)



(c)

Fig. 10. Effect of CO₂ phase on water fractional flow in displacement of water at injection flowrate of (a) 0.1 mL/min, (b) 0.2 mL/min and (c) 0.3 mL/min.

4.4. Relationship between water recovery and capillary number

The method to calculate the water production was discussed in section 3.1. Fig. 11(a)-(c) show the cumulative water production denoted by the pore volume as a function of CO₂ injected pore volume at an injection flowrate varying from 0.1 mL/min to 0.3 mL/min. The results indicate that the cumulative water production reaches its maximum after injecting ~ 2 pore volumes of CO₂ and the displacement of water by LCO₂ gives the highest water production, while gCO₂-water displacement gives the lowest the water production CO₂. Commonly, the cumulative water production increases fast at the beginning (around CO₂ breakthrough point), and the growth rate slows down with the rise of injected CO₂ volume as water in larger pores is easily driven out first (Duan et al., 2016). Fig. 11(d) presents the terminal water production represented in PV as a function of injection flowrate. It is worthwhile to note that the ultimate water production increases with the increase of CO₂ injection flowrate, while the increasing trend tends to slow down. This is mainly because initial rise of injection flowrate could increase viscous force to push out the water which is occupying in large pores dominated by small capillary force. Further increasing flowrate would be more difficult to drive out the water remaining in small pores due to the large capillary force (Nobakht et al., 2007).

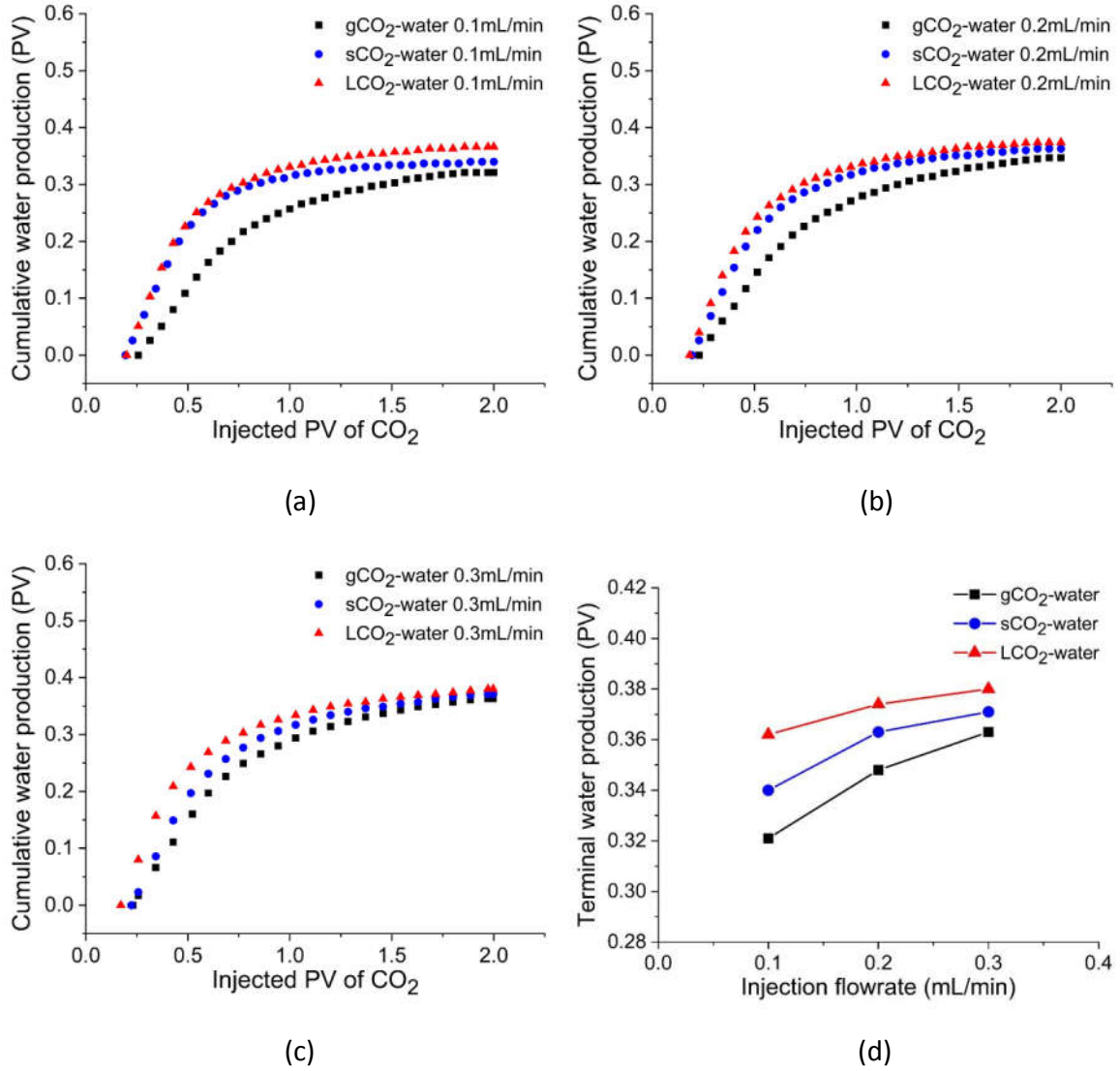


Fig. 11. Impact of CO₂ phase on water production as a function of injected pore volume of CO₂ at injection flowrate of (a) 0.1 mL/min, (b) 0.2 mL/min, (c) 0.3mL/min and (d) terminal water recovery as a function of injection flowrate.

In general, a two-phase flow in horizontal core sample is controlled by the capillary force and viscosity force. The quantitative relative effect between them can be evaluated as,

$$\frac{P_{vis}}{P_{cap}} \sim \frac{\mu v r / K}{\sigma / r} = \frac{r^2}{K} N_{ca} \quad (11)$$

where r is the grain size and K is the permeability of the core sample. The typical grain size is in order of 100 μm and the permeability in this experiment is about 100 mD. Thus, the value of

r^2/K is around 10^5 and for capillary number N_{ca} around 10^{-5} , viscous and capillary force are about the same. In this study, the capillary number $N_{ca} = \frac{v_{CO_2}\mu_{CO_2}}{\sigma_{CO_2,w}}$, given in terms of invading phase (CO_2), is ranging roughly between 10^{-9} between 10^{-7} , which suggests that capillary forces are more important. Fig. 12 shows the effect of capillary number on water recovery. The water recovery is defined as the ratio of cumulative water production volume to pore volume of the core. The results indicate that water recovery increases generally with the capillary number though the growth rate seems different. When N_{ca} is smaller than $\sim 2 \times 10^{-8}$, water recovery increases notably with the increase of capillary number because some water still remains in large pores after CO_2 flooding with small viscosity number. The increased viscous force pushes out more water. When N_{ca} is larger than $\sim 2 \times 10^{-8}$, the increase rate in water recovery slows down, as most unrecovered water remains in smaller pores after flooding with larger viscous force. Further increasing viscous force is hardly to overcome the strong capillary force to drive the remaining water out (Nobakht et al., 2007). Therefore, increasing capillary number is a more effective way to increase the displacement efficiency when the capillary number is smaller than $\sim 2 \times 10^{-8}$.

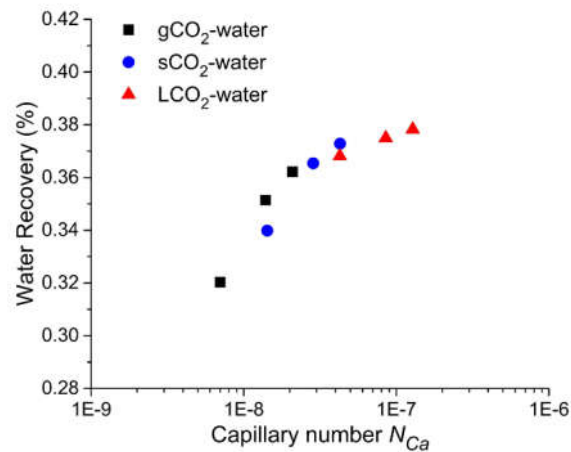


Fig. 12. Water recovery as a function of capillary number

5. Conclusions

The impact of CO₂ phase on the drainage process, especially the key parameters, such as CO₂ breakthrough time, capillary displacement pressure, relative permeability and displacement efficiency, has been thoroughly investigated in this study. The results indicate that:

- 1) CO₂ breakthrough point can be obtained from the turning point on the differential pressure curve by placing a low permeability membrane at the outlet of the core sample. The breakthrough is affected notably by CO₂ phase, but is independent of injection flowrate. The breakthrough happened earliest in LCO₂-water displacement, while it happened latest in gCO₂-water displacement.
- 2) The differential pressure fluctuates remarkably in LCO₂ drainage, but it is much smoother in gCO₂ drainage. The capillary displacement pressures for gCO₂, sCO₂ and LCO₂-water displacement are 0.059, 0.048 and 0.046 Bar, which are indicated by the pressure jumps and their values close to the results estimated by Kozeny model.
- 3) The relative permeability varies greatly with CO₂ phase, but is affected hardly by CO₂ injection flowrate. At the same injection flowrate, LCO₂-water displacement gives the largest relative permeability while gCO₂-water displacement has the smallest relative permeability. The uncertainty of relative permeability caused by incompressible assumption is largest for gCO₂-water displacement, especially at the beginning of the drainage.
- 4) CO₂ phase significantly affects the cumulative water production. LCO₂-water displacement gives the highest cumulative water production, while gCO₂-water displacement gives the lowest one at the same injection flowrate. Water recovery generally increases with the increase of N_{ca} . It increases remarkably when N_{ca} is smaller than $\sim 2 \times 10^{-8}$ and after which the growth trend slows down.

Acknowledgements

Authors gratefully acknowledge the financial supports from Chinese Scholarship Council (CSC) and UK India Education & Research Initiative (UKIERI).

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***Declaration of Interest Statement**

Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests: