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The HABIT (HabitAbility: Brine Irradiation and Temperature) environmental instrument for the ExoMars 2022 Surface Platform

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2

3 **The HABIT (HabitAbility: Brine Irradiation and Temperature) environmental instru-**
4 **ment for the ExoMars 2022 Surface Platform**

5

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14

15 *: JMT and MPZ led this work

16 **Abstract**

17

18 The HABIT (HabitAbility: Brine Irradiation and Temperature) instrument is a European pay-
19 load of the ExoMars 2022 Surface Platform Kazachok that will characterise the present-day
20 habitability at its landing place in Oxia Planum, Mars. HABIT consists of two modules: (i)
21 EnvPack (Environmental Package) that monitors the thermal environment (air and ground),
22 the incident ultraviolet radiation, the near surface winds and the atmospheric dust cycle; and
23 (ii) BOTTLE (Brine Observation Transition To Liquid Experiment), an In-situ Resource Uti-
24 lization instrument to produce liquid water for future Mars exploration. BOTTLE will be used
25 also to investigate the electrical conductivity properties of the martian atmosphere, the pre-
26 sent-day atmospheric-ground water cycle and to evaluate if liquid water can exist on Mars in

27 the form of brines, and for how long. These variables measured by HABIT are critical to de-
28 termine the present and future habitability of the martian surface. In this paper, we describe in
29 detail the HABIT instrument and sensors, together with the calibration of its Flight Model
30 (FM) and the Engineering Qualification Model (EQM) versions. The EnvPack module has
31 heritage from previous missions operating on the surface of Mars, and the environmental ob-
32 servations of its sensors will be directly comparable to those delivered by those missions.
33 HABIT can provide information of the local temperature with $\pm 0.2^\circ\text{C}$ accuracy, local winds
34 with ± 0.3 m/s, surface brightness temperature with $\pm 0.8^\circ\text{C}$, incident UV irradiance with 10%
35 error of its absolute value in the UV-A, UV-B, UV-C ranges, as well as in the total UV-ABC
36 range, and two additional wavebands, dedicated to ozone absorption. The UV observations
37 can be used to derive the total opacity column and thus monitor the dust and ozone cycles.
38 BOTTLE can demonstrate the hydration state of a set of four deliquescent salts, which have
39 been found on Mars (calcium chloride, ferric sulphate, magnesium perchlorate and sodium
40 perchlorate) by monitoring their electric conductivity (EC). The EC of the air and the dry salts
41 under Earth ambient, clean room conditions is of the order of $0.1 \mu\text{Scm}^{-1}$. We have simulated
42 HABIT operations, within an environmental chamber, under martian conditions similar to
43 those expected at Oxia Planum. For dry, CO_2 atmospheric conditions at martian pressures, the
44 air EC can be as low as $10^{-8} \mu\text{Scm}^{-1}$, however it increases with the relative humidity (RH)
45 percentage. The laboratory experiments show that after an increase from 0 to 60% RH within
46 a few hours, the EC of the air increased up to $10^{-1} \mu\text{Scm}^{-1}$, magnesium perchlorate hydrated
47 and reached values of $10 \mu\text{Scm}^{-1}$, whereas calcium chloride deliquesced forming a liquid state
48 with EC of $10^2 \mu\text{Scm}^{-1}$. HABIT will operate with a regular cadence, through day and night.
49 The Electronic Unit (EU) is protected with a heater that is activated when its temperature is
50 below -33°C and disabled if the temperature of the surface platform rises above -30°C . Addi-
51 tionally, the heaters of the BOTTLE unit can be activated to dehydrate the salts and reset the

52 experiment. HABIT weighs only 918 g. Its power consumption depends on the operation
53 mode and internal temperature, and it varies between 0.7 W, for nominal operation, and 13.1
54 W (when heaters are turned on at full intensity). HABIT has a baseline data rate of 1.5
55 MB/sol. In addition to providing critical environmental observations, this light and robust in-
56 strument, will be the first demonstrator of a water capturing system on the surface of Mars,
57 and the first European In-Situ Resource Utilization in the surface of another planet.

58

59

60 **Keywords:** Mars, ExoMars, Surface Platform, instrumentation, habitability, water, ISRU,
61 Atmosphere, regolith, brines, astrobiology

62

63 **1 Introduction**

64

65 The search for present life and habitability on Mars is conditioned by the availability of liquid
66 water. Also, the future potential exploration of Mars by humans is conditioned by the availa-
67 bility to produce in-situ critical products such as water, oxygen or propellant. Recently it has
68 been shown that liquid water can be stable on Mars in the form of brines (Martín-Torres et al.,
69 2015). Two other environmental conditions constrain the habitability of the near surface of
70 Mars: the thermal range and the ultraviolet radiation dose (Rummel et al., 2014). The quest
71 about the habitability of present-day Mars is still open and a crucial aspect for the future hu-
72 man space exploration. The HabitAbility: Brines, Irradiance and Temperature (HABIT) in-
73 strument, will focus on this subject. HABIT is one of the two European payloads of the Sur-
74 face Platform *Kazachok* that is part of the Exomars 2022 mission to Mars. ExoMars is a joint
75 program by the European Space Agency (ESA) and the Russian Space Agency (Roscomos)
76 which comprises two missions, in 2016 and 2022. The 2016 mission is called Trace Gas Or-

77 biter (TGO), which is now in operation on Mars monitoring traces gases, such as methane and
78 water in the atmosphere (Vandaele et al., 2019; Korablev et al. 2019). In addition to its own
79 scientific goals, TGO will serve as relay for the ExoMars 2022 surface mission. This mission,
80 to be launched in 2022 (final windows date to be decided), shall land at Oxia Planum
81 (18.20°N, 335.45°E) on Mars, a plain with an elevation more than 3,000 metres below the
82 martian mean. It will search for signs of present or past life on Mars with its rover and surface
83 platform and will bring the first European/Russian rover and Surface Platform (SP) to the sur-
84 face of Mars.

85

86 HABIT will characterize the habitability at the landing site of the mission, Oxia Planum
87 (Bhardwaj et al., 2019; Fonseca et al., 2019), in terms of Ultra-Violet (UV) radiation, air and
88 ground temperature (T), as well as liquid water availability and in-situ atmospheric water ex-
89 traction usage. It will also monitor the near surface winds, and the atmospheric dust cycle.
90 Additionally, it will provide measurements of the air electric conductivity (which changes
91 with the relative humidity of the air) and over time, when dust is accumulated in one of the
92 open cells of HABIT, it will provide an estimate of the electric conductivity of the dust. Hav-
93 ing in-situ information about these processes is also key for the design of future missions to
94 Mars.

95

96 The ExoMars mission will pursue one of the outstanding questions of our time by attempt-
97 ing to establish whether life ever existed or is still active on Mars today (Vago et al., 2017).
98 This is specifically the goal of the payloads of the rover
99 (<http://exploration.esa.int/mars/45103-rover-instruments/>) and HABIT, mounted on the Sur-
100 face Platform, will contribute to this later goal, namely it will assess the present-day habitabil-
101 ity of Oxia Planum, in terms of key environmental variables. The thermal ranges, water avail-

102 ability and UV radiation doses are critical environmental parameters that condition the habit-
103 ability of Mars. The correct characterization of the wind and heat fluxes on Mars are also im-
104 portant for the operational performance of spacecraft platforms and for our understanding of
105 the boundary layer dynamics. The dust cycle and the electric conductivity may also be poten-
106 tial hazards to the future exploration of Mars.

107

108 Finding unequivocal proofs of liquid water on present day Mars is a prominent domain of
109 Mars research with implications for the conditions promoting habitability and the future of
110 Mars exploration as well as for the definition of planetary protection protocols (Bhardwaj et
111 al., 2019; Rummel et al., 2014). HABIT includes the BOTTLE (Brine Observation Transition
112 To Liquid Experiment) module to capture at night-time atmospheric water by hydration and
113 deliquescence of salts, and three environmental sensors – Ground Temperature Sensor (GTS),
114 Air Temperature Sensor (ATS), and UltraViolet Sensor(UVS) – devoted to monitoring the
115 full diurnal and seasonal variations of the ground and air temperature, the near surface winds
116 and the UV irradiance. BOTTLE also includes two open containers to study the electrical
117 conductivity properties of the martian atmosphere. The information provided by the three en-
118 vironmental sensors of HABIT will allow to constrain the habitability at the landing site in
119 terms of metabolic and reproduction temperature, calculating the heat-flux and the UV bio-
120 logical dose, study the atmosphere/surface water interchange, providing information about
121 winds (which shall be useful also for the rover drilling and sampling operations and for vali-
122 dation of global and meso-scalar circulation models), about thermal inertia and subsurface
123 thermal profile and hydration level complementing the studies of the other platform and rover
124 instruments. The analysis of the measured UV irradiance may also be used to provide the
125 concentration of the atmospheric trace gas ozone (complementing orbiter observations from
126 NOMAD-TGO) and the total column of dust or opacity (providing continuous monitoring of

127 the dust cycle). Additionally, the environmental observations of HABIT will be compared
128 with those of REMS on board the Curiosity rover of the Mars Science Laboratory mission
129 (Gómez-Elvira et al., 2012; Gómez-Elvira et al., 2014) and of other environmental sensors
130 which are designed with heritage of REMS, such as Twins on InSight and MEDA on the
131 Mars 2020 NASA Perseverance rover. The BOTTLE module is furthermore designed as an
132 ISRU (In-situ Resource Utilization) demonstrator that shall quantify the amount of water (and
133 derived products such as H₂ and O₂) available for future landed missions on Mars. Thus,
134 HABIT will allow for extra long-term climate and atmospheric monitoring and will provide
135 an unequivocal proof, for the first-time, of the existence of liquid water on Mars, while
136 demonstrating the ISRU capabilities of salts.

137

138 The scientific objectives of the ExoMars program are defined as follows (Vago et al., 2017):
139 1) To search for signs of past and present life on Mars; 2) To investigate the wa-
140 ter/geochemical environment as a function of depth in the shallow subsurface; 3) To study
141 martian atmospheric trace gases and their sources; and 4) To characterize the surface envi-
142 ronment.

143

144 The scientific objectives of HABIT are:

145

- 146 1. Investigate the habitability of the landing site in terms of the three most critical envi-
147 ronmental parameters for life as we know it: availability of liquid water, UV biologi-
148 cal dose and thermal ranges (on Earth, microbial metabolism has only been found
149 above 240 K and reproduction above 255 K);

150

- 151 2. Provide environmental information (air and ground temperature, ground relative hu-
152 midity (RH) and UV irradiance), to investigate the atmosphere/regolith water inter-
153 change, the subsurface hydration, as well as the ozone, water and dust atmospheric cy-
154 cle and the convective activity of the boundary layer;
- 155
- 156 3. Demonstrate an In-Situ Resource Utilization technology for future Mars exploration.
- 157
- 158 4. The clear detection of a high conductivity signal within BOTTLE would reveal that
159 the deliquescence mechanism is in effect on Mars. This would have major implica-
160 tions. First, it would confirm the possibility of a present-day atmosphere-regolith in-
161 teraction that may lead to the formation of briny mixtures which, in the case of deep
162 slopes may be observed from orbit through large-scale features like Recurrent Slope
163 Lineae. This is still a controversial topic, i.e., see Bhardwaj et al. (2019) and Dundas
164 et al. (2020). The identification of the transient liquid water and its stability time
165 scales would open the discussion about the possibility of life in present day Mars and
166 various other key points linked with the future human exploration. It is also expected
167 that the demonstration of the formation of liquid brine would have implications on the
168 definition of special regions defined for planetary protection purposes.

169

170 HABIT is designed to be a robust, small and light instrument that complies with the ExoMars
171 science objectives and priorities:

172

- 173 1. Search for signs of past and present life on Mars: The clear detection of transient liq-
174 uid conditions is key to assessing the possibility of present life on Mars, since water is
175 a requirement for life as we know it. The determination of the surface and subsurface

176 thermal range will allow evaluating if the landing site complies with the known ther-
177 mal limits of terrestrial microbial reproduction and metabolism;

178

179 2. Investigate the water/geochemical environment as a function of depth in the shallow
180 subsurface: The investigation of the diurnal cycle of water, the formation of frost and
181 the hydration of salts at the surface and the subsurface will allow to investigate the
182 water environment as a function of depth in the upper meters of the regolith where the
183 diurnal and annual thermal wave act;

184

185 3. Study martian atmospheric trace gases and their sources: Water is a trace gas on Mars
186 and HABIT will investigate the water cycle, its sources (when released from the soil
187 dehydration) and sinks (absorbed by the soil). O_3 is a trace gas produced by atmos-
188 pheric photolysis with UV, the concentration of O_3 can be monitored with the UV
189 sensor, its cycle anti-correlates with atmospheric water. The UV irradiance also drives
190 the concentration of other minor gases that are produced from CO_2 : CO, O_2 ;

191

192 4. Characterize the surface environment: HABIT will evaluate the habitability of the
193 near surface environment, in terms of thermal range, water availability and UV dose.
194 It will also provide critical environmental information regarding the dust cycle, and
195 wind activity, as well as the thermal tide response, heat-fluxes and planetary circula-
196 tion at the boundary layer.

197

198 This paper is aimed at describing the HABIT instrument, its operation and sensing elements
199 and the calibration and validation of the HABIT Flight Model (FM) and HABIT Engineering
200 Qualification Model (EQM) which have been developed by Omnisys, Sweden, and the Group

201 of Atmospheric Science at Luleå University of Technology (LTU), with support of the Swe-
202 dish National Space Agency (SNSA). Following ESA policy, the HABIT data will be re-
203 leased for public use and archived at Roscosmos/IKI and mirrored at the Planetary Science
204 Archive (PSA), and thus this document shall serve as reference for the future scientific analy-
205 sis of the HABIT data. It is also the intention of this article to describe the potential and limi-
206 tations of the instrument, so that the data can be interpreted correctly.

207

208

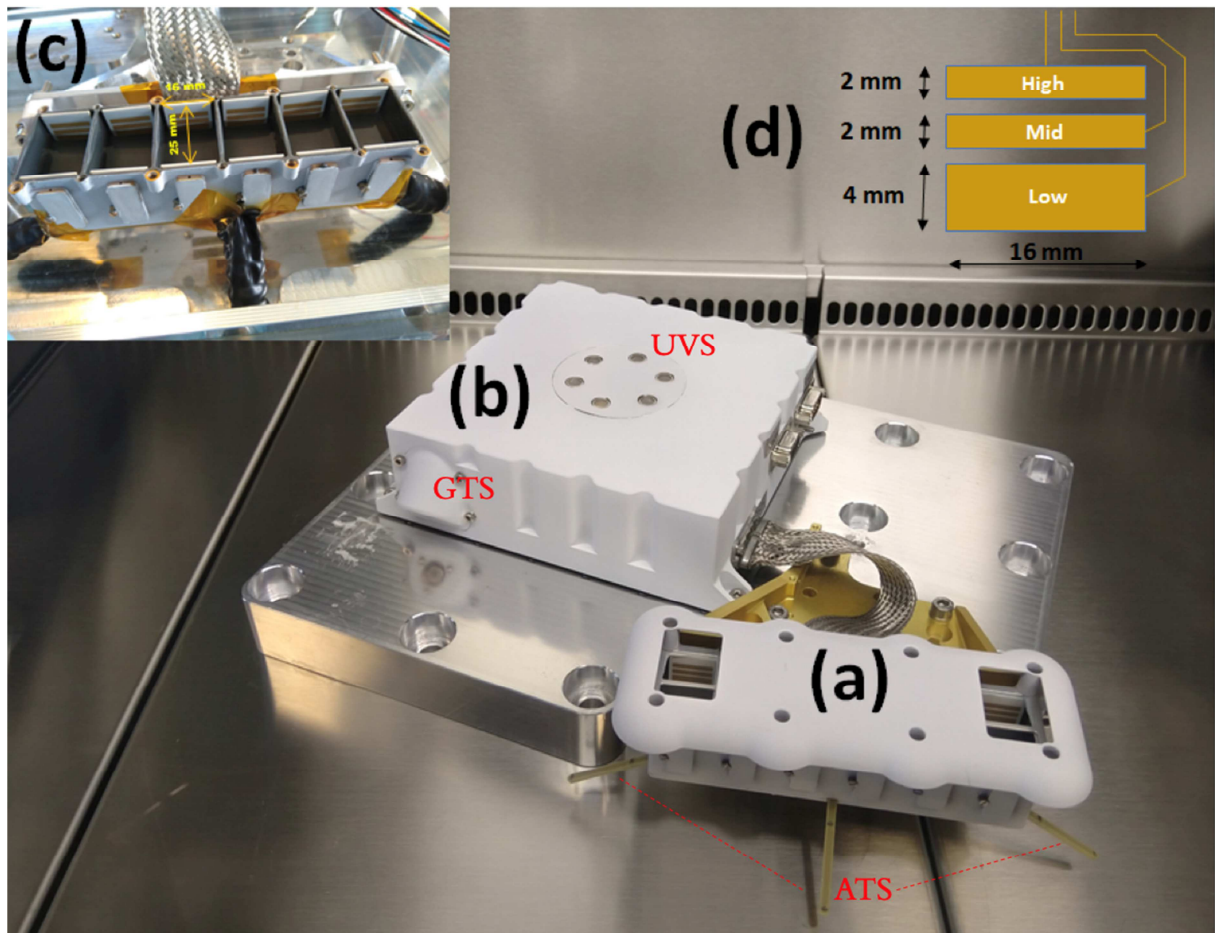
209 **2. Instrument design**

210

211 HABIT and its modules are shown in Figure 1. The dimensions of the Container Unit
212 (CU) the Electronic Unit (EU) are 150 mm x 63 mm x 43 mm and 172 mm x 172 mm x 55
213 mm (L x B x H) respectively. The Electronic Unit (EU) is protected with a heater that is acti-
214 vated when its temperature is below -33°C and disabled if the temperature of the Surface Plat-
215 form rises above -30°C . This implies that the CU, at nighttime when the RH% peaks, will be
216 warmer than the martian ground. The instrument weights 918 g; requires a power consump-
217 tion that ranges between 0.7 and 13.0 W (when the heaters are activated) and a baseline data
218 rate of 1.5 MB/sol. The CU, the lid and the filter holder are manufactured in aluminium, the
219 electrodes are made of gold, to prevent corrosion, and the magnetic rings around the UV pho-
220 todiodes are made of samarium cobalt. The instrument measures at 1 Hz, on its nominal op-
221 eration mode, during five minutes at the beginning of each hour. It can also operate in extend-
222 ed mode, within programmable continuous periods of time.

223

224



225

226

227 **Figure 1:** HABIT Flight Model (FM): (a) Container Unit (CU) with BOTTLE and the
 228 three ATS, (b) Electronic Unit (EU) with the UVS and GTS, (c) Open view of the empty con-
 229 tainers of BOTTLE with the golden electrodes, and (d) Electrode dimensions and separations.

230

231 The container Unit (CU) which includes the six containers of BOTTLE and the three Air
 232 Temperature Sensors (ATS) will be exposed to the free flow of air, mounted on the edge of
 233 the ExoMars SP. The Electronic Unit (EU) is mounted on top of the SP deck, with the ultra-
 234 violet sensors (UVS) pointing to the sky, and the ground temperature sensor (GTS) pointing
 235 to the side, with a clean field-of view, to monitor the temperature of the surface of the ground.
 236 The UVS is located in the EU lid centre and it includes six different photodiodes embedded
 237 within magnetic rings to deflect the trajectory of in-falling airborne dust and extend the life-

238 time of the sensors by keeping their field of view clean. The UVS has a free view of sky of
239 60° solid angle which make HABIT exposed to both direct sunlight and diffuse irradiance
240 without any shielding. The GTS is located on the short edge of the instrument. This sensor is
241 pointing toward the ground in an angle of 45°. The CU contains the six electric conductivity
242 cells of BOTTLE and three ATS. Two cells are open and four are sealed with a High Effi-
243 ciency Particle Air (HEPA) filter, with pore size 0.3 μm , and covered by a roof that allows the
244 flow of air while preventing direct obstruction by dust. The filter has two purposes: it prevents
245 inbound and outbound bioburden (organics or microbes) contamination and protects the salts
246 from dust while allowing air to flow in. The two open cells are aimed at investigating the
247 electrical conductivity of the martian air, as it changes throughout days and seasons depend-
248 ing on the amount of atmospheric water. Also, when airborne dust starts to accumulate at the
249 bottom of these containers, the electrodes will provide a measurement of the conductivity of
250 dust. The four central cells of BOTTLE are filled with four deliquescent salts that have been
251 reported on Mars, and a super absorbent polymer that will form a hydrogel when the deli-
252 quescent salts absorb water. This hydrogel will keep the brine in a semi-solid state, avoiding
253 capillarity of the pure liquid phase while supporting the capture of water and thus allowing to
254 measure electrical conductivity. The two open air cells have two measuring electrodes,
255 whereas the cells with the encapsulated products have three measuring electrodes. The quan-
256 tity of products is chosen such that the dry product covers fully the lower electrode, allowing
257 for expansions and contractions as the products absorb water, freeze or melt. When the salt
258 turns into liquid brine the volume increases, and the mid electrode may also be in contact with
259 the salt. The upper electrode is for control in case of inclination of the platform.

260

261 The data acquired by the sensors will be complemented with the timestamp of the acquisi-
262 tion moment and the platform inclination will be taken into account as ancillary data. This

263 ancillary information is needed to interpret the solar UV measurements, because the response
264 of the sensors depends on the angle of incidence of the direct irradiance from the Sun, also the
265 estimate of the total column of dust depends on the traversed slant path from the Sun to the
266 platform. A set of engineering housekeeping temperature sensors monitors the temperature of
267 the UVS, GTS and the EU and CU. The UVS and GTS temperature sensors are needed to ap-
268 ply the adequate calibration functions and retrieve the observable variables. Whereas the EU
269 and CU temperature measurements are needed to activate the heaters.

270

271 The instrument shall operate autonomously at day- and night-time and communicate with
272 the central computer once per sol. The regular, nominal, cadence of observation will be five
273 minutes at the beginning of each hour, acquiring data at 1 Hz. The EU includes different elec-
274 tronic functions such as start-up heaters, power supply, data handling and sensitive analogue
275 sensor circuits.

276

277 **3. Air temperature and wind retrieval from the Air Tempera-** 278 **ture Sensors (ATS)**

279

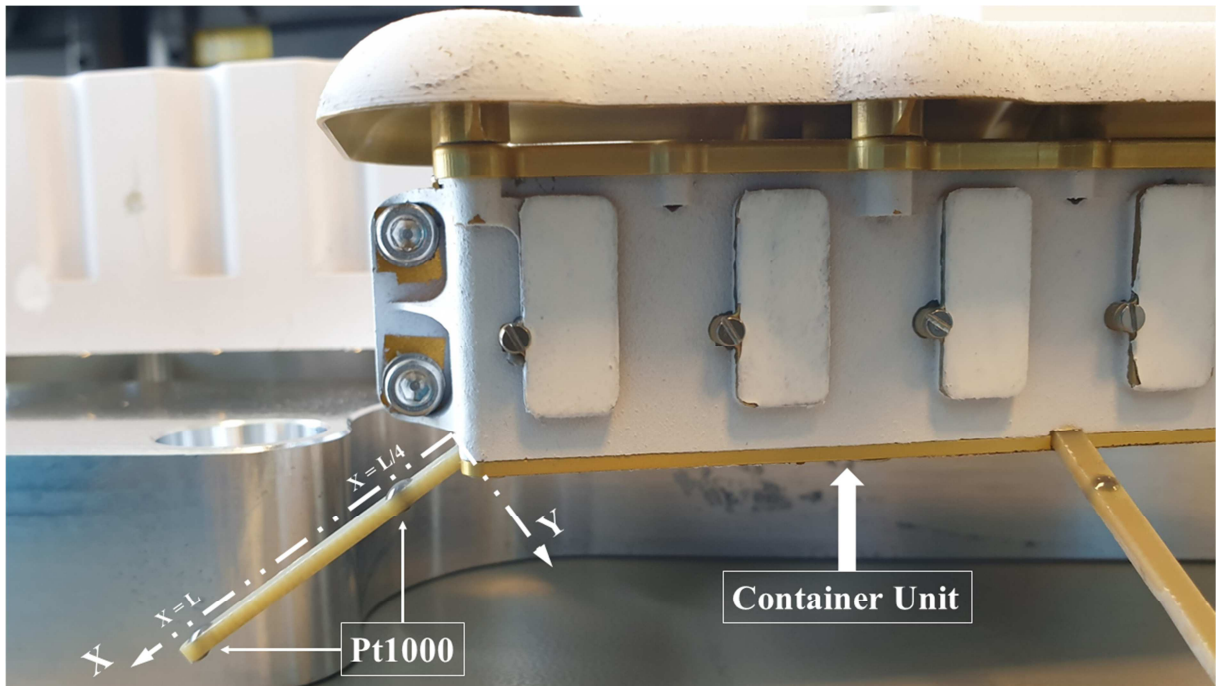
280 The HABIT Air Temperature Sensors (ATS) design, main working principles, and operation
281 modes are illustrated in section 4.1. Also, the calibration procedure of the HABIT FM ATS is
282 included in that section, together with results of the FM calibration campaign at Omnisys In-
283 struments AB facilities. The ATS in HABIT is an evolved version of the REMS ATS current-
284 ly operating on the surface of Mars on board the NASA Mars Science Laboratory/ Curiosity
285 rover. In this section we summarize a novel wind retrieval method that is based on the analy-

286 sis of the heat-flux of the ATS. In section 4.2, we will present an example of the calibration
287 of the ATS for wind retrieval under martian conditions, in the wind tunnel facility of Aarhus.

288 a. Design and operation of the ATS.

289 The near-surface air temperature changes mostly with the solar irradiance, varying about 90 K
290 between the day and night following the diurnal thermal tide; and following a seasonal behav-
291 iour along the year, depending on the solar longitude. The temperature also affects the martian
292 air density and it is important to describe the near surface heat-transfer processes. Further-
293 more, the temperature may show slow fluctuations in response to natural convection that takes
294 place at the planetary boundary layer during the day; it may also show the temperature of air
295 masses brought by wind from other locations, such as in the case of katabatic winds; and it
296 may also vary rapidly in response to very quick phenomena, such as dust-devils. To monitor
297 the air temperature in the vicinity of the platform, HABIT incorporates three ATS in three dif-
298 ferent orientations.

299 Figure 2 shows a detail view of ATS rod 1 mounted on the HABIT FM CU structure. Each
300 ATS consist on three rods of length $L = 3.6$ cm made of FR4 material with three Pt1000
301 thermistors attached at three different positions, to measure the temperature profile along the
302 rods. The temperatures recorded by each Pt1000 are expected to be thermally influenced by
303 the conductive heat exchange between the rods and the HABIT structure, which is in turn
304 thermally coupled to the SP. By monitoring these temperature profiles, the ATS can obtain an
305 ambient temperature corresponding to the film temperature of the SP. The material of the rods
306 has a low thermal conductivity, of $k = 0.7$ W/mK.



307

308 **Figure 2:** Detailed view of one of the ATS 1 attached to the side of the HABIT FM CU, and
 309 partial view of the frontal ATS. Three Pt1000 thermistors are located at $x = 0$, $x = L/4$ and $x =$
 310 L in order to record the temperature at the base, T_b , at the intermediate point, T_{Ln} , and at the
 311 tip of the rods, T_a , respectively. The Pt1000 at the base of the rod, $x = 0$, is not visible.

312 The air temperature acquisition performed by HABIT is similar to the REMS ATS recording
 313 (Gómez-Elvira, et al., 2012), and it is based on modelling the energy balance along a fin that
 314 is cooled by natural convection (Mueller & Abu-Mulaweh, 2006). Considering a local coordi-
 315 nate system at the ATS rods, where $x = 0$ is at the interface between the ATS rods and the CU
 316 wall (see Figure 2). Each rod has one thermistor at the tip ($x = L$) recording the temperature
 317 T_a , a second one at the base ($x = 0$) recording T_b , and a third one at an intermediate point, $x =$
 318 $L/4$, recording the temperature T_{Ln} . By assuming that the HABIT CU structure is in thermal
 319 equilibrium with the ambient, it is possible to express analytically the temperature distribution
 320 along each rod as a function of an average dimensionless variable \bar{m} , (here referred as the m-
 321 parameter), which is dependent on the thermal conductivity k , geometry, and average heat
 322 transfer coefficient \bar{h} of the rods.

$$323 \quad \theta(\chi) = \theta_b \frac{\cosh[\bar{m}(1-\chi)]}{\cosh(\bar{m})}, \quad (1)$$

$$324 \quad \bar{m} = L \sqrt{\frac{\bar{h}A_s}{kA_c}}. \quad (2)$$

325 Here, $\theta = T - T_\infty$ is the excess temperature, $\theta_b = T_b - T_\infty$ the excess temperature at the
 326 base of the rod, and T_∞ the temperature outside the thermal boundary layer of the rod. The
 327 latter was assumed to be similar to the fluid temperature $T_\infty \sim T_f$. A_s and A_c represent the
 328 lateral and cross-section areas of the rods, respectively. It should be noted that the \bar{h} coeffi-
 329 cient collects the radiative and convective terms, \bar{h}_c and \bar{h}_r , where constant emissivity, ε , and
 330 thermal conductivity have been assumed along the rods. The \bar{h} coefficient is assumed uniform
 331 and equal to:

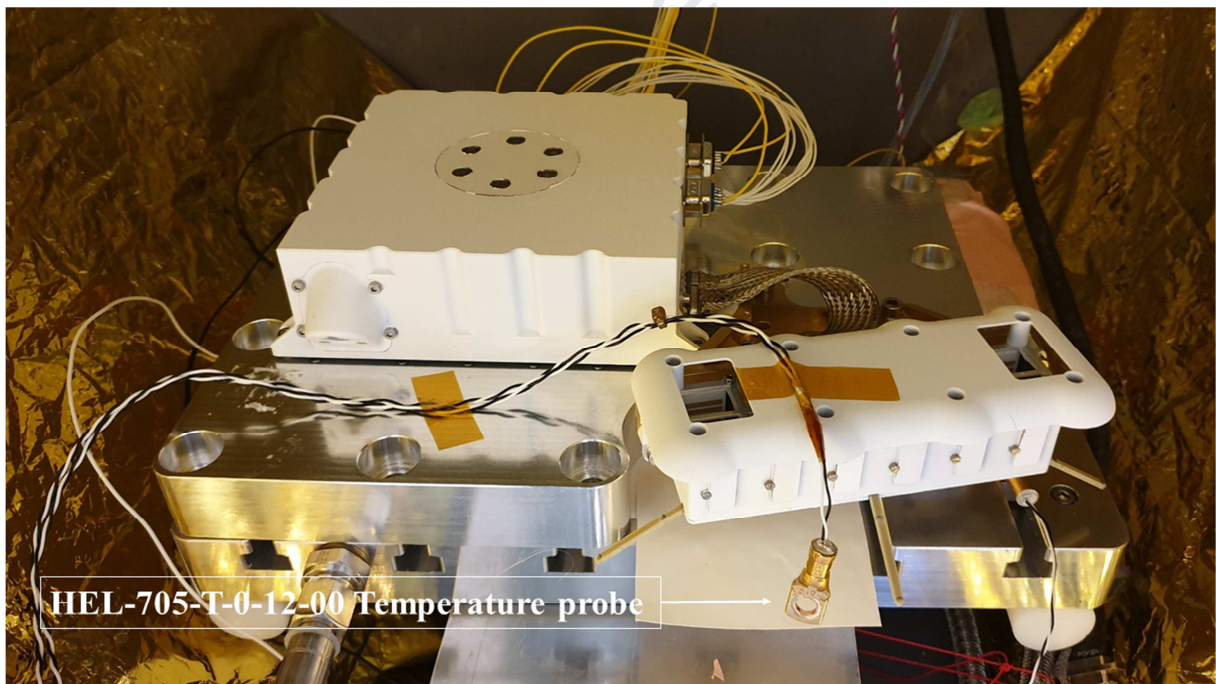
$$332 \quad \bar{h} = \bar{h}_c + \bar{h}_r = \bar{h}_c + \varepsilon\sigma(T^2 + T_\infty^2)(T + T_\infty), \quad (3)$$

333 where σ is the Stefan-Boltzmann constant. By introducing the three temperature readings at
 334 each ATS rod in Equation (1), it is possible to establish a system of two equations with the
 335 average m-parameter, \bar{m} , and the fluid temperature, T_f , as unknowns:

$$336 \quad \begin{cases} T_a - T_f = (T_b - T_f) \frac{1}{\cosh[\bar{m}]} \\ T_{Ln} - T_f = (T_b - T_f) \cdot \frac{\cosh[\bar{m} \cdot (1 - \frac{1}{n})]}{\cosh[\bar{m}]} \end{cases} \quad (4)$$

337 This approach is implemented on REMS, to obtain T_f , by solving the system (4). HABIT is
 338 designed to replicate this working principle for the ExoMars SP (Soria-Salinas, et al., 2020a).
 339 As can be inferred when solving the system of equations (4), the fluid temperature T_f depends
 340 on the absolute temperatures provided by the thermistors, but \bar{m} , as seen on Equation (1), is
 341 only dependent on the temperature differences. The temperature readings need to be calibrat-
 342 ed against a reference temperature probe to obtain the actual T_f .

343 The calibration of the thermistors was performed in a thermal vacuum chamber (TVAC) close
344 to vacuum conditions. On Mars, because of the low atmospheric density, the dominant heat-
345 exchange mechanism is radiative heat transfer, while on Earth, at ambient pressures, the high-
346 er density environment makes natural convection the dominant process. One of the calibra-
347 tions was performed under vacuum conditions in order to reduce the influence of the atmos-
348 phere and correct for the self-heating contamination produced by the initiation of the instru-
349 ment. Figure 3 shows the setup used for the HABIT flight model (FM) within the Omnisys
350 Instruments AB TVAC chamber. The test was performed at ambient temperatures for 12
351 hours using a reference temperature probe HEL-705-T-0-12-00 with an accuracy of $\pm 0.8\%$
352 (around $\pm 0.2^\circ\text{C}$ at 25°C) at a pressure of $1 \cdot 10^{-6}$ bar.

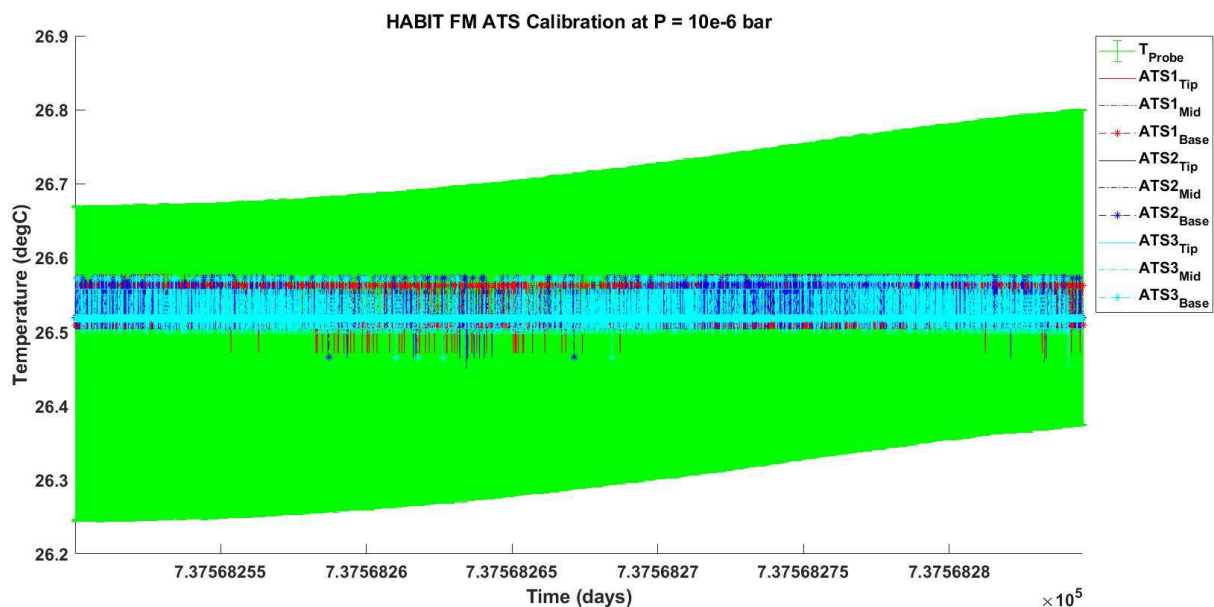


353

354 **Figure 3:** Setup for the HABIT flight model (FM) ATS calibration under low-pressure condi-
355 tions, $P = 1 \cdot 10^{-6}$ bar, and ambient temperature, $T = 25^\circ\text{C}$, in the Omnisys Instruments AB
356 TVAC. The reference temperature probe was a HEL-705-T-0-12-00 with an accuracy of \pm
357 0.8% and the pressure monitored by a ‘Leybold TR-211 Thermovac Pirani Vacuum Sensor’.

358

359 The pressure at the chamber was monitored by a ‘Leybold TR-211 Thermovac Pirani Vacuum
360 Sensor’ and maintained to $1 \cdot 10^{-6}$ bar, and the reference temperature was measured by a
361 ‘HEL-705-T-0-12-00’ probe with an accuracy of $\pm 0.8\%$. The probe was fixed close to the
362 ATS rods in order to expose the sensor to the most similar radiative environment possible. In
363 this test, all the nine temperatures provided by the Pt1000 were fitted to the reference temper-
364 ature. An average offset with respect to the reference temperature was calculated using the
365 measurements of each sensor. This allowed us to remove the initial temperature profiles orig-
366 inated from the self-heating of the instrument when switched-on. The instrument was config-
367 ured to perform an extended acquisition of 3 hours at 1 Hz for the calibration. The differences
368 were corrected with the averaged offsets that each thermistor presented during the central
369 hour of the extended acquisition period. The results of the calibration can be observed in Fig-
370 ure 4.



371

372 **Figure 4:** Results of the ATS calibration at $P = 1 \cdot 10^{-6}$ bar and $T = 25^\circ\text{C}$. conditions, in-
373 cluding the temperature probe HEL-705-T-0-12-00 absolute error of $\pm 0.8\%$, during the cen-
374 tral hour of a 3-h extended acquisition.

375 As it is shown in Figure 4, during the ATS calibration the reference probe presented an oscil-
376 lation of $\pm 0.1^\circ\text{C}$ limited by the capability of the thermal plate that was set to maintain the
377 HABIT structure temperature at 25°C . However, it must be noted that the overall accuracy of
378 the probe for the temperatures tested is around $\pm 0.2^\circ\text{C}$, as indicated in Figure 4 by the error
379 bar, which makes this variation negligible. As can be observed in the figure, the stability of
380 the Pt1000 readings in this hour of extended acquisition are always contained within the
381 boundaries of the errors of the reference temperature probe. As a result of this calibration, the
382 Pt1000 measurements at each rod are now capable of measuring temperatures at a resolution
383 of 0.05°C and with an accuracy of $\pm 0.2^\circ\text{C}$.

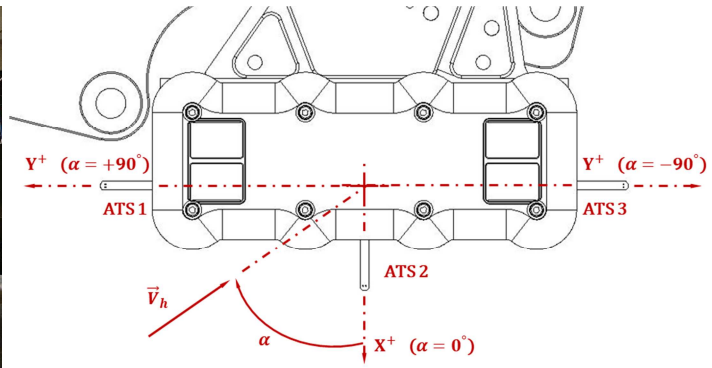
384 b. Wind retrieval from the ATS heat-flux analysis.

385 The recent results from the analysis of REMS ATS and WS dataset on Mars suggested the
386 possibility of using the REMS ATS temperature data to provide the MSL mission with wind
387 speed (Soria-Salinas, et al., 2020a). The main working principle of this wind retrieval method
388 is the use of the estimated average m-parameter at each rod, \bar{m} . This parameter was calculated
389 in the original fluid temperature retrieval performed by REMS, but it was neither stored nor
390 further processed. According to its definition, the m-parameter is correlated with the wind
391 speed through the average convective heat transfer term, \bar{h}_c , included in the average heat
392 transfer coefficient, $\bar{h} = \bar{h}_c + \bar{h}_r$ (see equation (2)). In addition, the \bar{h}_c coefficient of each rod
393 is dependent on natural and forced convection; that is, on the relative speed of the airflow that
394 surrounds the ATS rods. Thus, if the correlation between the \bar{h}_c coefficient and cooling air-
395 flow relative speed around each rod was modelled, wind speed could be retrieved, in first ap-

396 proach, from the estimated m-parameters. The proposed wind-retrieval method is thus based
397 on modelling the forced convection on the ATS rods (Soria-Salinas, et al., 2020a). This work
398 shows that this retrieval technique, when applied to REMS ATS data, matched the REMS WS
399 wind speed values (allowing for WS errors of 20%, the error of the sensor according to
400 Gomez-Elvira, J., 2013 a,b,c) up to $\sim 77\%$ of the time when the incident horizontal wind field
401 direction was $\alpha \in [12.95^\circ, 107.05^\circ]$ (α is defined as the horizontal angle clockwise from
402 REMS Boom 2 relative to Boom 1, see Soria-Salinas, et al. 2020a).

403 Because the HABIT ATS design is based on REMS ATS, the wind retrieval developed for the
404 REMS dataset is also applicable to HABIT. In this case a set of calibration experiments are
405 needed to convert the derived heat-transfer to the corresponding wind speed and to calibrate
406 the emissivity to thermal conductivity ratio. A wind tunnel campaign was performed in the
407 Aarhus Wind Tunnel Simulator (AWTS), Aarhus University, Denmark, with the main goal of
408 assessing the overall error of the wind retrieval concept when applied to the HABIT EQM
409 ATS temperature measurements. This facility has been frequently used for the testing of space
410 hardware for Mars (Holstein-Rathlou, et al. 2010). Two tests were performed so that the ATS
411 rods were exposed to a normal airflow. These tests were also used to verify the thermal con-
412 ductivity of the HABIT ATS rods after the Pt1000 thermistors and associated printed circuitry
413 (Soria-Salinas, et al. 2020b). The tunnel consists on a TVAC that possess an electric motor
414 (Danfos VLT5001) that moves an internal fan magnetically coupled. This fan creates a close
415 circuit of low-pressure air that can replicate typical low-speed winds on the surface of Mars.
416 The cylindrical tunnel is around 8 m x 2.5 m, although the effective testing area, far from wall
417 interference, is about 0.4 m in diameter (Merrison, et al., 2008). Wind speeds up to ~ 16 m/s
418 could be simulated within the tunnel, although beyond ~ 8 m/s the friction of the recirculating
419 air artificially overheats the flow (Soria-Salinas, et al. 2020b).

420 Figure 5 shows the setup for the two tests performed within the AWTS and the reference sys-
 421 tem considered. Test 1 simulated front winds ($\alpha = 0^\circ$), where the airflow is perpendicular to
 422 both ATS rod 1 and ATS rod 3. Test 2 simulated lateral winds ($\alpha = -90^\circ$); that is, airflows
 423 perpendicular to the ATS rod 2. The range of wind speeds tested in the tunnel was limited to
 424 $V \lesssim 12$ m/s because of excessive overheating of the airflow observed during the operations.



425

426 **Figure 5:** (Left) HABIT EQM setup within the AWTS. The instrument is attached to an alu-
 427 minium mast through a thick aluminium plate. The mast could rotate in the horizontal plane
 428 360° with respect to the centre of the EQM CU. The tunnel fan that suctions the air towards
 429 the EQM is visible behind the setup. (Right) HABIT CU top view including the reference sys-
 430 tem used for the tested wind orientation. Test 1 simulated front winds ($\alpha = 0^\circ$) and Test 2
 431 simulated lateral winds ($\alpha = -90^\circ$).

432 As it was mentioned earlier, the HABIT wind speed retrieval method relies on the modelling
 433 of forced convection on the ATS rods. The main purpose is then to obtain an analytical ex-
 434 pression that correlates the averaged convective heat transfer coefficient at each rod, \bar{h}_c , esti-
 435 mated from \bar{m} , with the local wind speed. This was performed through the average Nusselt
 436 number, \bar{N}_u , defined as the ratio of convective to conductive heat transfer terms within the
 437 boundary layer developed around the ATS rods, $\bar{N}_u = \bar{h}_c \cdot L_c / k$. Here, L_c represents the char-
 438 acteristic length.

439 The rectangular-based ATS rods were assumed to be cylinders and several empirical formula-
 440 tions of \bar{N}_u for normal cross-flow were assessed (Fand and Keswani, 1972; McAdams, 1954;
 441 Perkins and Leppert, 1964; Van Der Hegge Zijnen, 1956). These \bar{N}_u models were developed
 442 for similar fluid flow regimes to the expected low-density airflows on the surface of Mars and
 443 for the L_c of the heat transfer problem on the ATS rods; in this case $L_c \sim$ diameter of the cyl-
 444 inders. As a result, by comparing the \bar{h}_c coefficient estimated from \bar{m} with the \bar{h}_c resulting
 445 from the averaged Nusselt number model at each rod, it is possible to obtain the wind speed.
 446 Equation (5) summarizes the analytical expression developed for the wind speed retrieval,
 447 whose solution provides wind speed from the estimated \bar{m} values.

$$448 \quad \bar{h}_c = \frac{\bar{m}^2 \cdot K \cdot a \cdot b}{L^2 \cdot 2 \cdot (a+b)} - \sigma \varepsilon (T_s^2 + T_\infty^2) (T_s + T_\infty) = \frac{k}{L_c} \cdot \bar{N}_u. \quad (5)$$

449 Here, $a = 2 \text{ mm}$ and $b = 3 \text{ mm}$ correspond to the sides of the rectangular cross-section of
 450 each ATS rod, T_∞ is the temperature of the fluid T_f which is retrieved through equation (4)
 451 and T_s the temperature of the ATS surface which is calculated from the average of the tem-
 452 peratures provided by the three ATS thermistors. From the wind tunnel tests performed in the
 453 AWTS, and according to the conclusions from the application of the retrieval concept to the
 454 REMS temperature data when compared to the REMS WS values (Soria-Salinas, et al.,
 455 2019a), the average Nusselt number model developed by S. Whitaker (Van Der Hegge Zijnen,
 456 1956) was selected for the ATS rods (see equation (6)).

$$457 \quad \bar{N}_u = (0.4 \cdot Re^{0.5} + 0.06 \cdot Re^{2/3}) \cdot Pr^{0.4} \cdot (\mu_f / \mu_s)^{0.25}. \quad (6)$$

458 In equation (6), $Re = \rho U_\infty L_c / \mu$ is the Reynolds number, which represents the ratio between
 459 the viscous and convective terms of a fluid flow of density ρ with unperturbed wind speed
 460 U_∞ , which for our tests is V_{Tunnel} , and dynamic viscosity μ around the rods of diameter L_c .
 461 $Pr = \mu \cdot C_p / k$ is the Prandtl number, that represents the ratio between the momentum diffu-

462 sivity and thermal diffusivity within a fluid flow of specific heat at a constant pressure C_p and
463 thermal conductivity k . μ_f and μ_s correspond to the dynamic viscosity evaluated at the fluid
464 temperature, T_f , and at the averaged surface temperature of the rods, T_s , respectively.

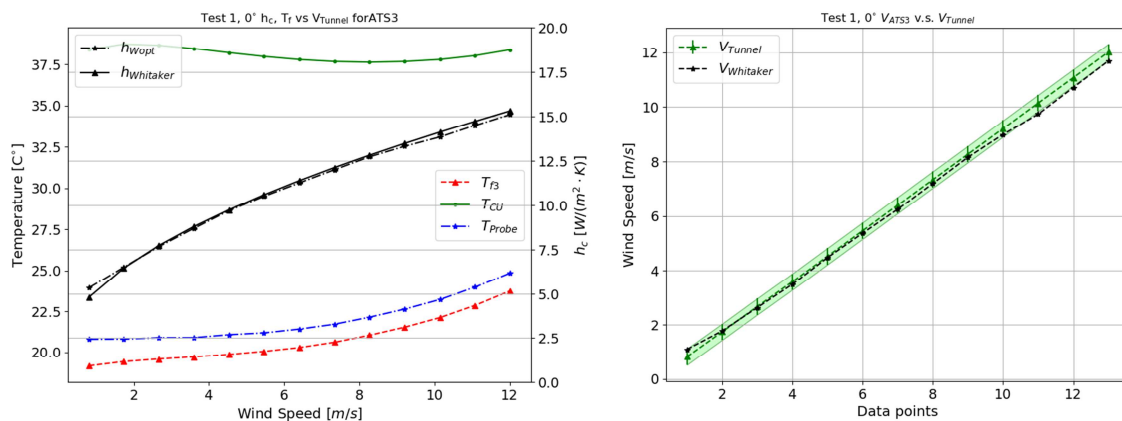
465 The final accuracy of the ATS wind speed retrieval was set based on the comparison with the
466 wind tunnel reference values. The latter were obtained after the calibration of the correlation
467 between the fan revolutions per minute, and the measurements provided by a commercial
468 Dantec laser Doppler anemometer; the nominal errors at the testing area for the selected range
469 of wind speed is ± 0.3 m/s (Merrison, et al., 2008).

470 Although both ATS 1 and ATS 3 were equally exposed to the same free stream, the experi-
471 ments suggested that the aluminium plate to which the HABIT CU was attached locally modi-
472 fied the horizontal velocity field in the vicinity of the ATS rod 1; that is, ATS rod 1 was too
473 close to the plate and the perturbation introduced in the flow by the latter modified the airflow
474 upstream. As a result, no reference value of the actual wind speed sensed by the ATS rod 1
475 was available to be compared with the retrieval prediction. When compared to the free stream
476 speed values, the maximum error observed in the ATS rod 1 retrieval, assuming the Whitaker
477 approach, was 0.81 m/s for $V_{Tunnel} \in [0,10]$ m/s. ATS 3, on the contrary, demonstrated to be
478 sufficiently separated from the plate and this hypothetical perturbation into the ATS rod 3 sur-
479 rounding airflow was not observed, providing a maximum wind speed retrieval error of 0.28
480 m/s with respect to the reference values for $V_{Tunnel} \in [0,10]$ m/s. It should be noted that both
481 the shape of the aluminium plate and CU relative orientation to the platform are not similar to
482 the final configuration on board the SP, both ATS1 and ATS3 are similarly distant from the
483 SP. However, for calibration purposes we shall use the tests of the ATS that is less perturbed
484 by its holding structure. Then on Mars, the three ATS will provide the wind measurement at
485 the place where they are installed, namely this local wind may be significantly different from

486 the free flow of air on Mars. This contamination is unavoidable for any instrumentation
 487 mounted on a landed spacecraft on Mars. As a result, from the AWTS analysis an error of
 488 0.81 m/s was assumed for both sensors to predict possible unknown interference on the re-
 489 trieval of frontal winds (Soria-Salinas, et al., 2020b).

490 Results from Test 2, where the CU was exposed to lateral winds ($\alpha = -90^\circ$), demonstrated
 491 the ATS 2 wind speed prediction capabilities, for the Whitaker approach, with an maximum
 492 error of 0.29 m/s for $V_{Tunnel} \in [0,6]$ m/s; that is, comparable to the wind tunnel wind speed
 493 reference probe error. However, ATS rod 2 results exposed a limitation in the wind speed re-
 494 trieval for $V_{Tunnel} \in [6,10]$ m/s inherent to the instrument structural design. The results sug-
 495 gested that, for horizontal lateral winds, the viscous boundary layer developed along rectangu-
 496 lar shape of the CU structure could present separation bubbles along its entire lateral surface
 497 for wind speeds > 5.5 m/s. Okajima (1990) predicted this point for a Reynolds number, based
 498 on the length of the larger side of the CU ($L_c = A$), at $Re_A \sim 200$, which agrees with the
 499 Reynolds at $V_{Tunnel} = 5.5$ m/s; that is, $Re_A = 193$ (Soria-Salinas, et al., 2020b).

500 Figure 6 shows as an example the retrieval output when applied to ATS rod 3 under front
 501 winds in Test 1 ($\alpha = 0^\circ$).



503 **Figure 6:** (Left) Comparison of the evolution of the average convective heat transfer coefficient
 504 estimated from \bar{m} , \bar{h}_{Wopt} (assuming the Whitaker approach for the \bar{N}_u modelling), with
 505 respect to the theoretical coefficient associated to the free stream wind speeds, $\bar{h}_{Whitaker}$, on
 506 the ATS rod 3. The estimated fluid temperature, T_{f3} , as well as the temperatures from the CU
 507 Pt1000, T_{CU} , and the tunnel, T_{Probe} (which is in a distant position in the tunnel), are also in-
 508 cluded. (Right) Comparison of the wind speed prediction performed by the ATS rod 3 with
 509 the free stream wind speed values. The errors associated to the wind speed reference estima-
 510 tion of ± 0.3 m/s have been included.

511 As can be seen on Figure 6 (left), the average convective heat transfer coefficient estimated
 512 from \bar{m} , \bar{h}_{Wopt} , fairly match the theoretical predictions of this coefficient, $\bar{h}_{Whitaker}$, when
 513 the wind speed measured by the tunnel probe, V_{Tunnel} , is used for its calculation (instead of
 514 \bar{m}). Also, the overheating of the airflow because of the friction of the flow with the inner tun-
 515 nel walls can be observed on the temperature evolution of the tunnel temperature probe,
 516 T_{Probe} . Figure 6 (right) shows the comparison between the wind speed retrieval and the free
 517 stream wind speed values tested. Here, the tunnel wind speed reference error of ± 0.3 m/s has
 518 been included. Each wind speed value tested was kept constant for 10 min, and both the refer-
 519 ence and retrieval wind speed values were averaged for this period.

520 In summary, the wind tunnel campaign performed in the AWTS allowed us to assess the ca-
 521 pabilities and main limitations of the HABIT ATS wind retrieval for frontal and lateral winds.
 522 The formers are expected to be retrieved with a maximum error of 0.81 m/s from ATS 1 and
 523 ATS 3 for horizontal wind speeds $V \in [0,6]$ m/s, assuming some perturbation from the Exo-
 524 Mars SP. The latter, with a maximum error of 0.29 m/s from ATS 2 for horizontal wind
 525 speeds $V \in [0,6]$ m/s; the platform is not expected to interfere in the airflow around ATS rod
 526 2 in this case. Future wind tunnel studies are planned to investigate the time response sensitiv-

527 ity to varying winds, the role of the temperature and the possibility to extract information
528 about the angle of incidence of the wind.

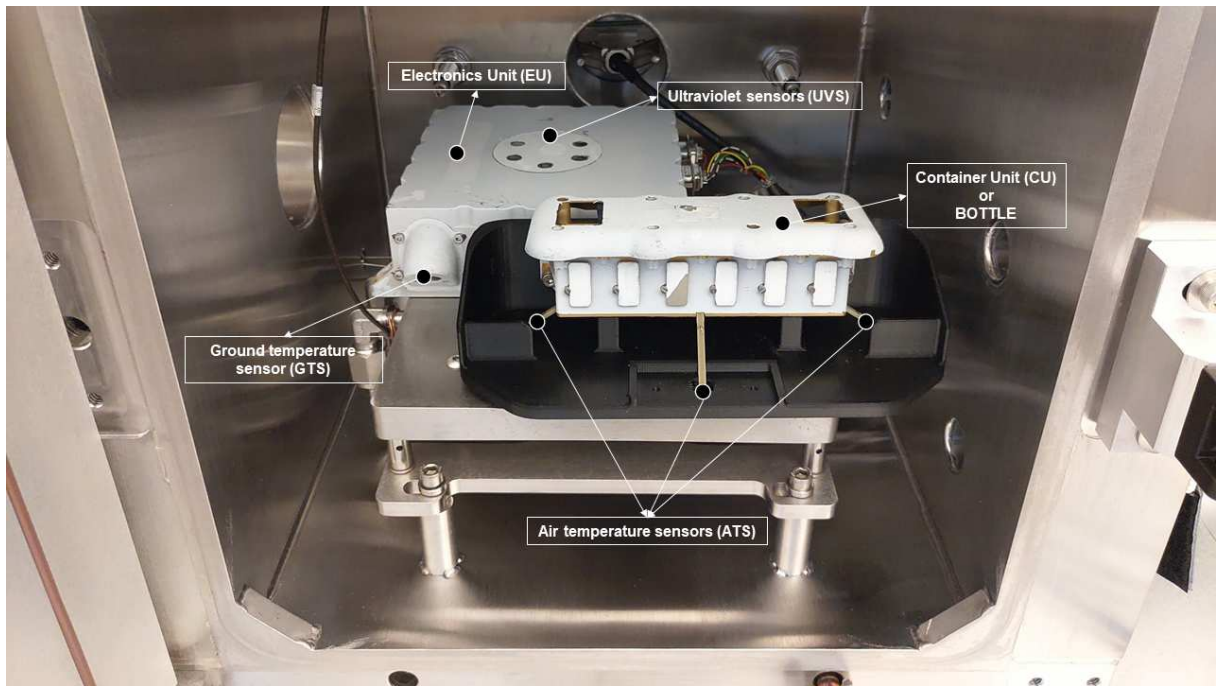
529

530 **4. Ground Temperature Sensor (GTS)**

531

532 The HABIT Ground Temperature Sensor (GTS) monitors the temperature at a single spot of
533 the surface at the landing site. The GTS is a light weight, low power, and low-cost pyrometer
534 that will measure the soil kinematic temperature of the martian surface during the nominal
535 mission lifetime of one Earth year. It benefits from a simple design with no moving parts and
536 its fixed pointing to the martian surface at an angle of 45° . The sensor acquires its heritage
537 from the Rover Environmental Monitoring Station (REMS) Ground Temperature Sensor
538 (GTS), an instrument aboard the NASA's Mars Science Laboratory (Gomez-Elvira et al.,
539 2012; Gomez-Elvira et al. 2014). Figure 7 shows the Engineering Qualification Model (EQM)
540 of HABIT, including GTS as a part of the Electronics Unit (EU). In this section, we describe
541 the design of GTS, the energy balance model and the associated calibration with the Engi-
542 neering Qualification Model (EQM) and the Flight Model (FM).

543



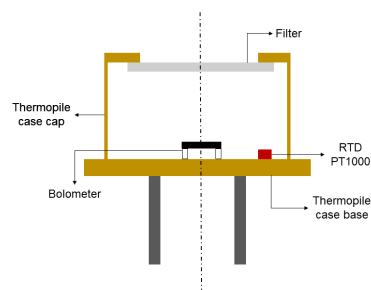
544

545 **Figure 7:** Image of the HABIT instrument Engineering Qualification Model (EQM) in the
 546 SpaceQ chamber highlighting the Container Unit (CU), Electronic Unit (EU) with Ground
 547 Temperature Sensor (GTS) and other sensors.

548 **4.1 Design and operation**

549 The core of the GTS is an 8-14 μ m wavelength band sensitive infrared thermopile (Figure 8),
 550 which depending on the total energy influx from its field of view, converts this to a corre-
 551 sponding voltage output.

552



553

554 **Figure 8:** Schematic representation of a cross-section of a thermopile and its components.

555 The sensing area of the thermopile that receives the energy from its field-of-view and con-
 556 verts it to a voltage is the bolometer. The response depends on the difference in temperature
 557 with the thermopile case base (cb) as measured with the help of a Pt1000 temperature sensor.
 558 The incoming energy passes through an 8-14 μ m band-pass filter to measure temperatures in a
 559 range specific to the Mars. The blackbody temperature for a peak wavelength can be calculat-
 560 ed using the Wien's displacement law as shown in equation (7).

$$561 \quad \lambda_{max} = \frac{2898}{T} (\mu m) \quad (7)$$

562 The thermopile with its 8-14 μ m band-pass filter is sensitive to measure a temperature range
 563 between -66.17 °C and 89.07 °C.

564

565 **4.2 Measurements and calibration**

566 The GTS measures the thermopile case base (cb) temperature and the thermopile output as
 567 ADC counts. HABIT uses a 12-bit ADC (4096 step resolution) with a reference voltage (V_{ref})
 568 of 4.096 V. The amplifier gain (A_v) is set to 201 in order to read the output signal, since the
 569 output voltage produced by the thermopile is marginally smaller than the reference voltage.
 570 The thermopile output voltage ($V_{thermopile}$) can be calculated using equation (8).

571

$$572 \quad V_{thermopile} = \frac{1}{A_v} * \left(\frac{ADC \ counts * V_{ref}}{ADC \ resolution} \right) = ADC \ Counts * \left(\frac{100}{201} \right) (\mu V) \quad (8)$$

573

574 The resultant voltage output is a function of the difference in temperature between the ther-
 575 mopile case base (cb) and the bolometer (s) as expressed in equation (9).

576

$$577 \quad V_{thermopile} = f(T_{cb})(T_s - T_{cb}) \quad (9)$$

578

579 where, T_{cb} is the temperature of the thermopile case base and T_s is the temperature of the bo-
580 lometer.

581

582 For the IPHT TS-100 thermopile used in the GTS, the relationship is provided by the manu-
583 facturer as in equation (10).

584

$$585 \quad f(T_{cb}) = -88.86 * 10^{-4} + 3.057 * 10^{-4} * T_{cb} - 0.01526 * 10^{-4} * T_{cb}^2 + 3.366 * 10^{-9} * T_{cb}^3 - 2.788 * 10^{-12} * T_{cb}^4$$

586 (10)

587

588 For a set of 100 thermocouples connected in series in the thermopile, equation (10) is modi-
589 fied to equation (11)

$$590 \quad V_{\text{thermopile}} = 100 * f(T_{cb}) * (T_s - T_{cb}). \quad (11)$$

591

592 **4.2.1 Energy balance model**

593 In a non-contact measuring instrumentation like the thermopile, the energy flux contributing
594 to the temperature measurement could be of conductive, convective and radiative nature as
595 shown in Figure 9.

596

597 The energy balance equation (Gómez-Elvira et al., 2012, Sebastian et al. 2010) is expressed in
598 equation (12):

599

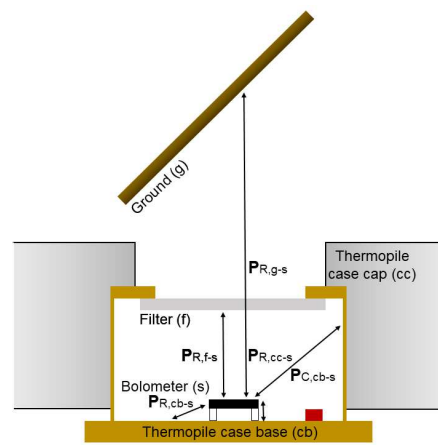
$$600 \quad P_{R,g-s} + P_{R,f-s} + P_{R,cc-s} + P_{R,cb-s} + P_{C,cb-s} = 0 \quad (12)$$

601

602 where, P – Energy

603 R – Radiation

604 C – Conduction
 605 g – Ground
 606 f – Filter
 607 cc – Thermopile case cap
 608 cb – Thermopile case base
 609 s – Bolometer



611
 612 **Figure 9:** Graphical representation of the conductive, convective and radiative energy fluxes
 613 associated within the thermopile and with the external environment. The black arrows show
 614 the interaction between the different components of the thermopile with the subscripts depict-
 615 ing if the nature of the energy flux is conductive, convective or radiative in nature.

616
 617 Considering constants K_1 , K_2 , and K_3 , each representing the relative weight of different ener-
 618 gy transfer processes, and assuming filter temperature, T_f and the thermopile case cap tem-
 619 perature, T_{cc} are the same as the thermopile case base temperature, T_{cb} , since there are in con-
 620 tact with each other, the simplified energy balance equation obtained is as shown in equation
 621 (13).

622

$$623 \quad K_1(\Phi_g^I - \Phi_s^I) + K_1(\Phi_{cc}^O - \Phi_s^O) + K_2(\Phi_{cc}^T - \Phi_s^T) + K_3(T_{cb} - T_s) = 0 \quad (13)$$

624

625 where, Φ - Heat flux

626 I – Heat flux of radiation passing through filter (wavelength restricted)

627 O - Heat flux radiation not passing through filter

628 T - Heat flux of radiation in total spectrum (all wavelengths)

629

630 The different heat flux terms are calculated using Planck's law as shown in equations (14) to

631 (19).

$$632 \quad \Phi_g^I = \varepsilon_g \int_{\lambda_a}^{\lambda_b} T_f(\lambda) \frac{2hc^2}{\lambda^5 \left(e^{\frac{hc}{\lambda k_B T_g}} - 1 \right)} d\lambda \quad (14)$$

633

$$634 \quad \Phi_{cc}^T = \int_0^\infty \frac{2hc^2}{\lambda^5 \left(e^{\frac{hc}{\lambda k_B T_{cc}}} - 1 \right)} d\lambda \quad (15)$$

635

$$636 \quad \Phi_{cc}^I = \int_{\lambda_a}^{\lambda_b} T_f(\lambda) \frac{2hc^2}{\lambda^5 \left(e^{\frac{hc}{\lambda k_B T_{cc}}} - 1 \right)} d\lambda \quad (16)$$

637

$$638 \quad \Phi_s^T = \int_0^\infty \frac{2hc^2}{\lambda^5 \left(e^{\frac{hc}{\lambda k_B T_s}} - 1 \right)} d\lambda \quad (17)$$

639

$$640 \quad \Phi_s^I = \int_{\lambda_a}^{\lambda_b} T_f(\lambda) \frac{2hc^2}{\lambda^5 \left(e^{\frac{hc}{\lambda k_B T_s}} - 1 \right)} d\lambda \quad (18)$$

641

$$642 \quad \Phi_{cc}^O = \Phi_{cc}^T - \Phi_{cc}^I \quad \Phi_s^O = \Phi_s^T - \Phi_s^I \quad (19)$$

643

644 where, α_s – Bolometer absorptivity = 1

645 ϵ_g – Ground emissivity

646 ϵ_{cc} – Thermopile can cap emissivity = 1

647 ϵ_{cb} – Thermopile can base emissivity = 1

648 ϵ_s – Bolometer emissivity = 1

649 $T_f(\lambda)$ – Filter transmittance = 0.754

650 $\rho_f(\lambda)$ – Filter reflectance = 0

651 T_g – Ground temperature

652 T_f – Filter temperature

653 T_{cc} – Thermopile case cap temperature

654 T_{cb} – Thermopile case base temperature

655 T_s – Bolometer temperature

656 $T_f = T_{cc} = T_{cb}$

657 h = Planck's constant

658 c = Speed of light = 3×10^8 m/s

659 k_B = Boltzmann constant

660 λ_a = Minimum limit of the wavelength band of the thermopile filter

661 λ_b = Maximum limit of the wavelength band of the thermopile filter

662

663 **4.2.2 Thermopile output voltage vs temperature**

664 The purpose of calibration of the GTS was to determine the calibration constants in the ener-

665 gy balance equation and represent the output voltage produced by the thermopile to the

666 brightness temperature of the target area in its field-of-view. We used a blackbody calibration

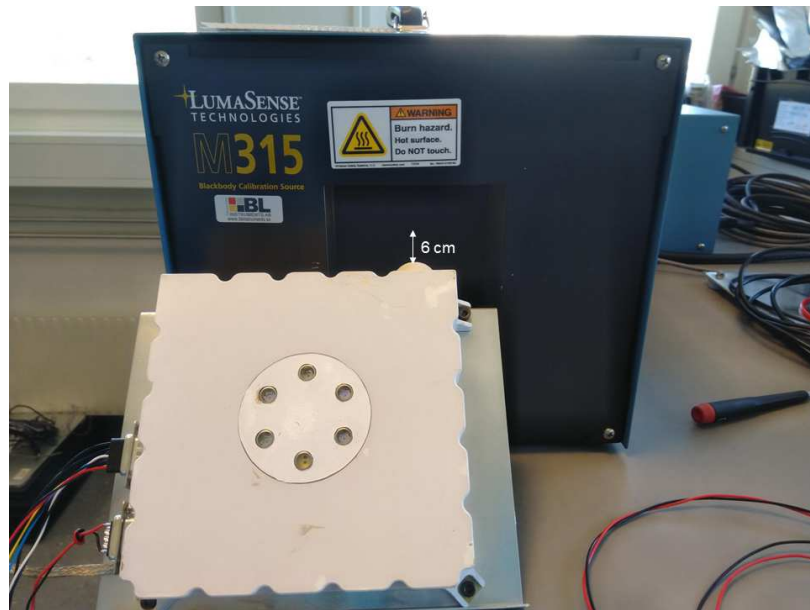
667 source with emissivity (ϵ) equal to 1 to establish this relationship. The calibration constants

668 and the resultant energy balance equation obtained were used for the accurate conversion of

669 the thermopile output voltage to the target temperature assuming the emissivity of the martian
670 surface from literature. The calibration procedure of the GTS consisted of three different
671 blackbody targets: 1) Blackbody calibration source and controller (LumaSense Technologies
672 M315X4-HT, Exit port: 101 mm X 101 mm) for ambient laboratory temperature experiments;
673 2) Peltier controlled thermographic paint coated aluminum plate for colder temperatures in-
674 side the climate chamber; and 3) Thermographic paint coated aluminum plate inside the
675 SpaceQ Mars simulation chamber (an environment simulating facility designed and built for
676 HABIT validation and calibration) at martian temperature, pressure and carbon-dioxide at-
677 mosphere (Vakkada Ramachandran et al., 2019). A MELEXIS MLX90614 infrared ther-
678 mometer (with a temperature measurement range of $-70\text{ }^{\circ}\text{C}$ to $380\text{ }^{\circ}\text{C}$ and a measurement ac-
679 curacy and resolution of $0.5\text{ }^{\circ}\text{C}$ and $0.02\text{ }^{\circ}\text{C}$ respectively) was used as a reference for this ap-
680 plication.

681
682 This experiment was conducted in three parts: i) under laboratory ambient, with blackbody
683 temperatures between $25\text{ }^{\circ}\text{C}$ and $73\text{ }^{\circ}\text{C}$ using LumaSense Technologies blackbody calibration
684 source (see Figure 10); and ii) under martian temperatures in a climate chamber, with black-
685 body temperatures between $10\text{ }^{\circ}\text{C}$ and $-50\text{ }^{\circ}\text{C}$ using the peltier controlled blackbody source
686 ($100\text{ mm} \times 100\text{ mm}$ aluminium target coated with LabIR thermographic paint, normal emis-
687 sivity of 0.92 for $8\text{-}14\text{ }\mu\text{m}$ band, see Figure 11); iii) under martian conditions of temperature,
688 pressure and carbon-dioxide atmosphere in a simulation chamber, with blackbody tempera-
689 tures between $10\text{ }^{\circ}\text{C}$ and $-50\text{ }^{\circ}\text{C}$ with $10\text{ }^{\circ}\text{C}$ interval using the thermographic paint coated alu-
690 minium plate with paint properties same as (ii), see Figure 12, attached to the liquid nitrogen
691 cooled working table of the chamber. All the experiments were conducted in an ISO Class 7
692 clean room with HABIT EQM.

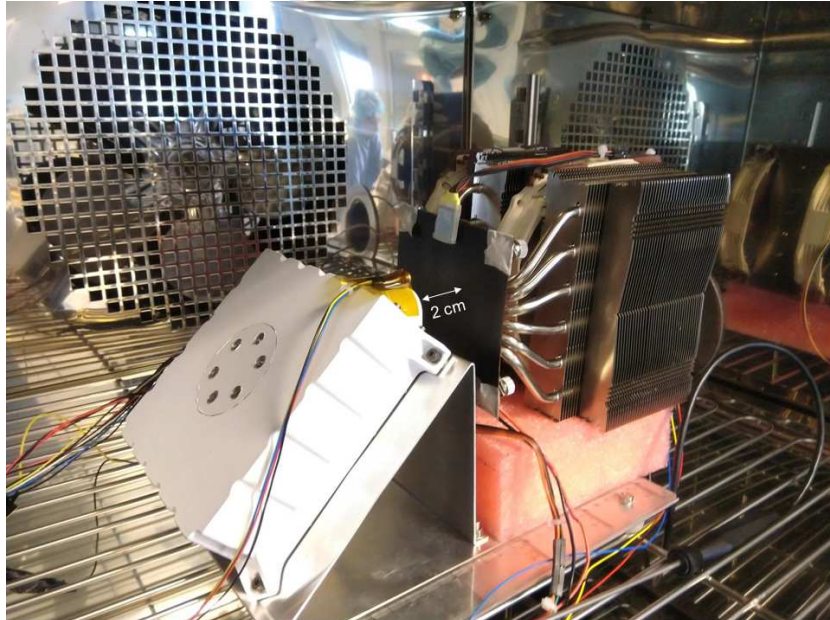
693



694

695 **Figure 10:** Experimental setup of the GTS mounted against the LumaSense Technologies
696 blackbody calibration source where the distance between the blackbody source and the GTS
697 is maintained at 6 cm.

698

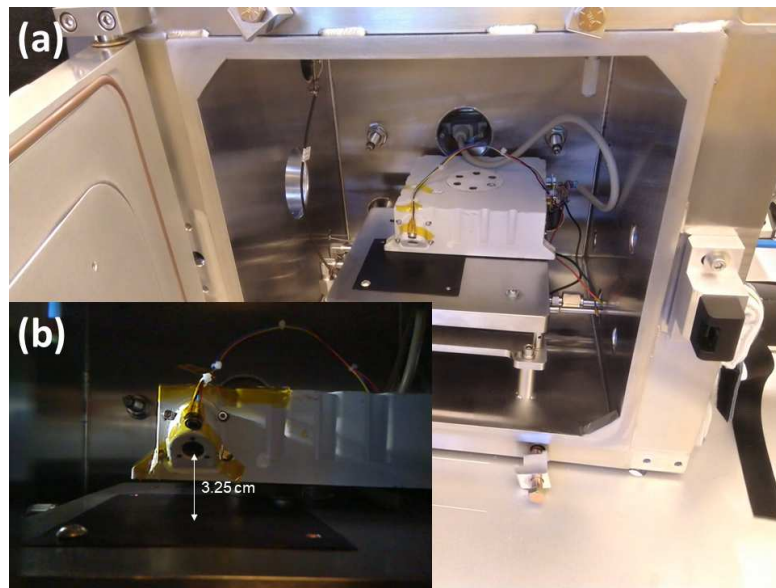


699

700 **Figure 11:** Experimental setup of the GTS against the Peltier-controlled thermographic paint
701 coated aluminium plate inside the climate chamber where the distance between the blackbody
702 source and the GTS is maintained at 2 cm.

703

704



705

706 **Figure 12:** Experimental setup of the GTS against the thermographic paint coated aluminium
707 plate inside the Mars simulation chamber where the GTS is maintained at a slant distance of
708 3.25 cm and pointed at an angle of 45° , representative of the operational scenario on Mars.

709

710 All the experiments were performed by maintaining a blackbody temperature between 73°C
711 and -50°C , decreasing the temperature at 10°C intervals while the ambient temperature was
712 maintained relatively constant. For the experiment under martian conditions, an insulating
713 spacer was used to isolate the Electronics Unit (EU) from the cold working table. The re-
714 sponse of the thermopile was recorded for 100 data points at a frequency of 1000 ms.

715

716 The different distance between the GTS and the blackbody calibration source for each of the
717 experimental setup (6 cm for LumaSense Technologies blackbody calibration source, 2 cm
718 for peltier controlled for thermographic paint coated aluminum plate, and 3.25 cm for ther-
719 mographic paint coated aluminum plate inside the SpaceQ Mars simulation chamber) was
720 maintained such that the oval footprint of the thermopile's field-of-view was covering only
721 the calibration target of the blackbody. Hence, the difference in distances does not affect the

722 accurate retrieval of the target temperature but rather provide an average brightness tempera-
 723 ture from the area of coverage in the field-of-view of the thermopile. During HABIT deploy-
 724 ment on Mars the distance between the GTS and the martian surface will be of 1.5 m. Since
 725 the temperature is computed from the overall radiation within the field-of-view, the same cal-
 726 ibration can be used without any modification for its use on Mars.

727

728 4.2.3 Ground temperature calculation

729 From the energy balance equation (12) and adding radiation from blackbody reflections Φ_{re-}
 730 flection, we obtain equation (20):

$$731 \quad K_1(\Phi_g^I + \Phi_{reflection} - \Phi_s^I) + K_1(\Phi_{cc}^O - \Phi_s^O) + K_2(\Phi_{cc}^T - \Phi_s^T) + K_3(T_{cb} - T_s) = 0 \quad (20)$$

732

733 For this calibration, all the data points from ambient laboratory tests with the Lumasense
 734 blackbody calibration source, the martian temperature tests in a climate chamber and the mar-
 735 tian condition tests inside the SpaceQ Mars simulation chamber were used.

736

737 Solving this equation for pairs of blackbody temperatures,

$$738 \quad K_1(\Phi_{g1}^I + \Phi_{reflection} - \Phi_{s1}^I) + K_1(\Phi_{cc1}^O - \Phi_{s1}^O) + K_2(\Phi_{cc1}^T - \Phi_{s1}^T) + K_3(T_{cb1} - T_{s1}) = 0$$

$$739 \quad K_1(\Phi_{g2}^I + \Phi_{reflection} - \Phi_{s2}^I) + K_1(\Phi_{cc2}^O - \Phi_{s2}^O) + K_2(\Phi_{cc2}^T - \Phi_{s2}^T) + K_3(T_{cb2} - T_{s2}) = 0$$

740

$$741 \quad K_1(\Phi_{g1}^I - \Phi_{g2}^I - \Phi_{s1}^I + \Phi_{s2}^I) + K_1(\Phi_{cc1}^O - \Phi_{cc2}^O - \Phi_{s1}^O + \Phi_{s2}^O) + K_2(\Phi_{cc1}^T - \Phi_{cc2}^T -$$

$$742 \quad \Phi_{s1}^T + \Phi_{s2}^T) + K_3(T_{cb1} - T_{cb2} - T_{s1} + T_{s2}) = 0 \quad (21)$$

743

744 $K_1(=1)$, K_2 , and K_3 are achieved from equation (21) using linear least square algorithm.

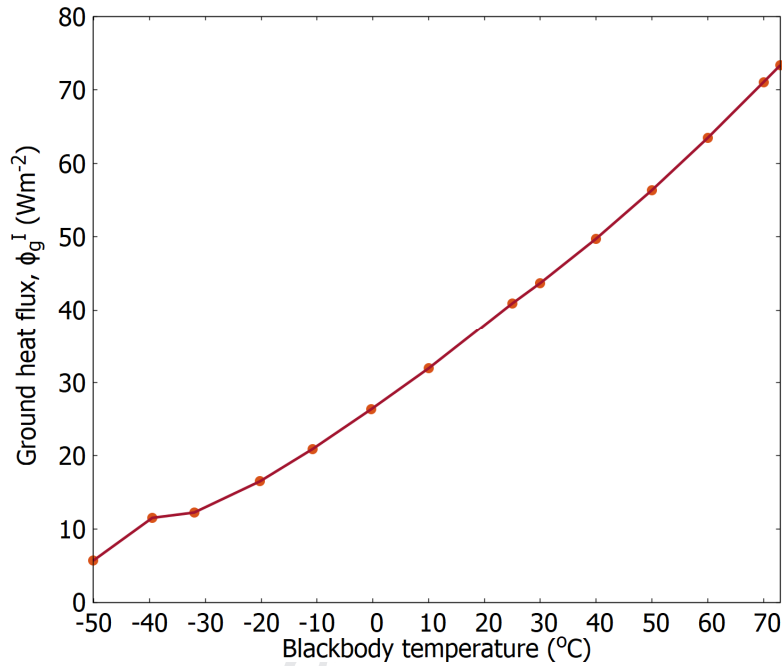
$$745 \quad [K_1 \ K_2 \ K_3] = [1 \ 113.4336 \ -7462.499]$$

746

747 Using the constants K_1 , K_2 , and K_3 in equation (13), the ground heat flux term, Φ_g^I can be
 748 computed, as shown in Figure 13, for the measured blackbody temperatures.

749

750



751

752 **Figure 13:** Computed ground heat flux values for the corresponding blackbody target temper-
 753 atures based on equation (7).

754

755 The target temperature, T_g can then be calculated from the resultant ground heat flux using a
 756 3rd order polynomial least square fit as expressed in equation (22).

$$757 \quad T_g = 9.235e^{-05} \cdot \Phi_g^I{}^3 - 0.0103 \cdot \Phi_g^I{}^2 + 1.8493 \cdot \Phi_g^I + 285.3986 \quad (22)$$

758

759 The retrieved target temperature is the brightness temperature considering the blackbody with
 760 an emissivity, $\epsilon = 1$. For martian surfaces where $\epsilon \neq 1$, the assumed emissivity from the litera-
 761 ture can substituted to recalculate the Φ_g^I term in equation (14) to retrieve the soil kinematic
 762 temperature on Mars.

763

764 **5.3 GTS performance**

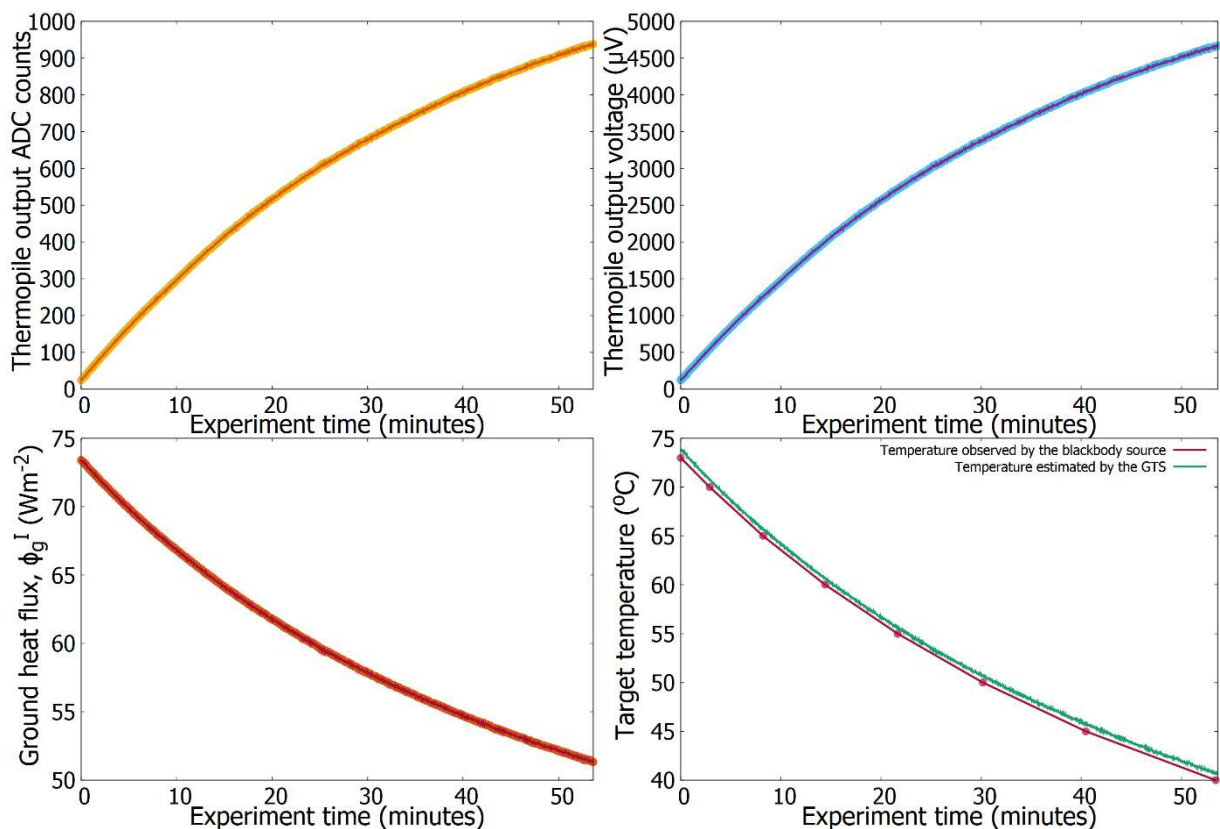
765 The purpose of this experiment was to verify the overall performance of the GTS thermopile
 766 response after obtaining the calibration constants from the tests described in section 5.2.2.

767 This experiment used the same setup of the LumaSense Technologies blackbody source and
 768 the HABIT EQM but was performed as a continuous single sweep in the blackbody tempera-

769 tures from 73 °C to 40 °C including the transitions between temperatures. The experiment
 770 was also conducted in an ISO Class 7 clean room with GTS thermopile at ambient tempera-

771 ture and maintaining 6 cm between the blackbody source and the GTS. The results are shown
 772 in Figure 14.

773



774

775 **Figure 14:** GTS performance tests. The retrieved GTS temperatures obtained using the cali-

776 bration function are shown in d).

777

778 We have found that with the current calibration, the error in temperature retrieval is within 0.8
779 °C. With more calibration data points for intermediate temperatures and more stability, the
780 energy balance model can be improved and thereby the error can be minimized.

781

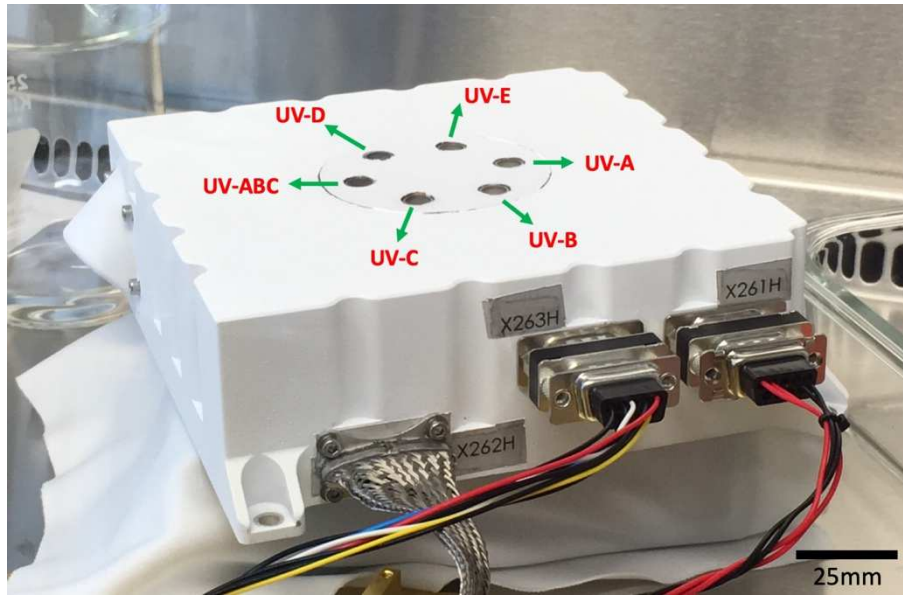
782 **5. Ultraviolet Sensor (UVS)**

783

784 In this section, we present the Ultraviolet Sensor (UVS) design and the calibration tests which
785 have been conducted on the FM, EQM and a UVS engineering calibration prototype. The
786 UVS is located on the top surface of the EU of HABIT, mounted on the Surface Platform
787 deck. It consists of 6 SiC photodiodes designated as UV-A, UV-B, UV-C, UV-D, UV-E and
788 UV-ABC, see Figure 15. Their measurements cover the UV range of 200-400 nm, see Table
789 1. Table 1 shows the spectral bandwidths of the photodiodes, the nominal gain values of the
790 HABIT electronics and the maximum measurable UV irradiance. The UVS has a solid angle
791 field of view of $\pm 30^\circ$ with an unobstructed view of the sky, similarly to the UVS of the
792 REMS sensor on the Curiosity rover (Gomez-Elvira et al., 2012). The UVS of HABIT, along
793 with the REMS-UVS, will enable simultaneous measurements of UV radiation on the martian
794 surface at two different locations on the planet. Each photodiode is housed in a TO-39 pack-
795 age and has an active sensing area of 0.965 mm^2 . Each photodiode is placed within a sur-
796 rounding magnetic ring to deflect airborne, in-falling, magnetic dust particles and mitigate the
797 obstruction of the windows of the sensor by dust. The UVS includes a thermistor to monitor
798 its temperature.

799

800



801

802 **Figure 15: UV photodiodes on the top of the HABIT electronic unit. The location of each**
 803 **photodiode type is marked with arrows.**

804 **Table 1: Spectral bands, gain and maximum measurable UV irradiance by the HABIT UVS photodiodes.**

805

Channel	Spectral band (nm)	HABIT-UVS Gain (G_i) (mV/nA)	Maximum measurable UV irradiance (W/m^2)
UV-A	335-395	2	28.30
UV-B	280-325	5	8.94
UV-C	220-275	10	3.14
UV-ABC	210-380	0.649	59.46
UV-D	245-290	10	2.93
UV-E	310-335	5	7.38

806

807 The UV Photodiodes are passive sensors activated by the incident UV radiation that excites
 808 the SiC sensor area. Upon excitation, the photodiodes produce a current which is converted,
 809 by the HABIT electronics, to voltage values. The maximum allowed output of the UV photo-
 810 diodes is divided into 4096 ADC levels (12 bits), each one represents a 1mV (with a maxi-
 811 mum output of 4.096 V). This gives a sensing resolution limit of 0.024% of the maximum
 812 measurable irradiance. The voltage output V_i (mV) obtained from the photodiodes is a func-
 813 tion of 4 parameters namely;

- 814 (i) Incident spectral radiation $I_i(\lambda)$ (nW/cm²)
- 815 (ii) Sensor spectral responsivity $S(\lambda)$ (A/W)
- 816 (iii) Sensor area A_i (cm²)
- 817 (iv) HABIT electronics gain G_i (mV/nA)

818

819 Equation (23) shows the integrated spectral radiance over the bandwidth of the UV photodi-
 820 ode, weighted by the spectral responsivity of the sensor and multiplied by the sensing area of
 821 the detector. Equation (24) is the voltage that the electronics reads with the gain of the circuit:

822

$$823 \text{ Current}_i = A_i \int_{\lambda_{ai}}^{\lambda_{bi}} S(\lambda) I(\lambda) d\lambda \cong R_i^{avg} A_i \int_{\lambda_{ai}}^{\lambda_{bi}} I(\lambda) d\lambda = A_i R_i^{avg} I_i^{avg} \quad (23)$$

824

$$825 V_i = G_i A_i R_i^{avg} I_i^{avg} \quad (24)$$

826

827 We can calibrate R_i^{avg} by measuring under a controlled calibrated broadband incident irradi-
 828 ance, the voltage response, knowing the $I_i^{avg} = \int_{\lambda_{ai}}^{\lambda_{bi}} I(\lambda) d\lambda$ incident within the range of re-
 829 sponse of the sensor

$$830 R_i^{avg} = \frac{V_i}{A_i G_i I_i^{avg}} \quad (25)$$

831 or alternatively by calibrating the photodiode spectrally with narrow wavelength excitation
832 beams, as it is done by the manufacturers, and then calculating a weighted average. During
833 operations on Mars, the incident irradiance can be retrieved by converting the output voltage
834 recorded by the UVS electronics, knowing the gain, sensing area and weighted sensor respon-
835 sivity within the given wavelength range.

836

$$837 \quad I_i = \frac{V_i}{A_i G_i R_i^{avg}} \quad (26)$$

838

839

840 The calibration process of UV photodiodes with a monochromator yields their spectral re-
841 sponsivity $S(\lambda)$ at normal incidence and ambient temperatures. Other tests are focused on es-
842 timating the linearity and the dependence on the angle of incidence of the direct solar beam.
843 The angular tests have been performed with the FM and EQM HABIT EU at ambient condi-
844 tions, whereas the temperature dependence tests are performed in a thermally controllable
845 chamber. The calibration procedures are described next.

846

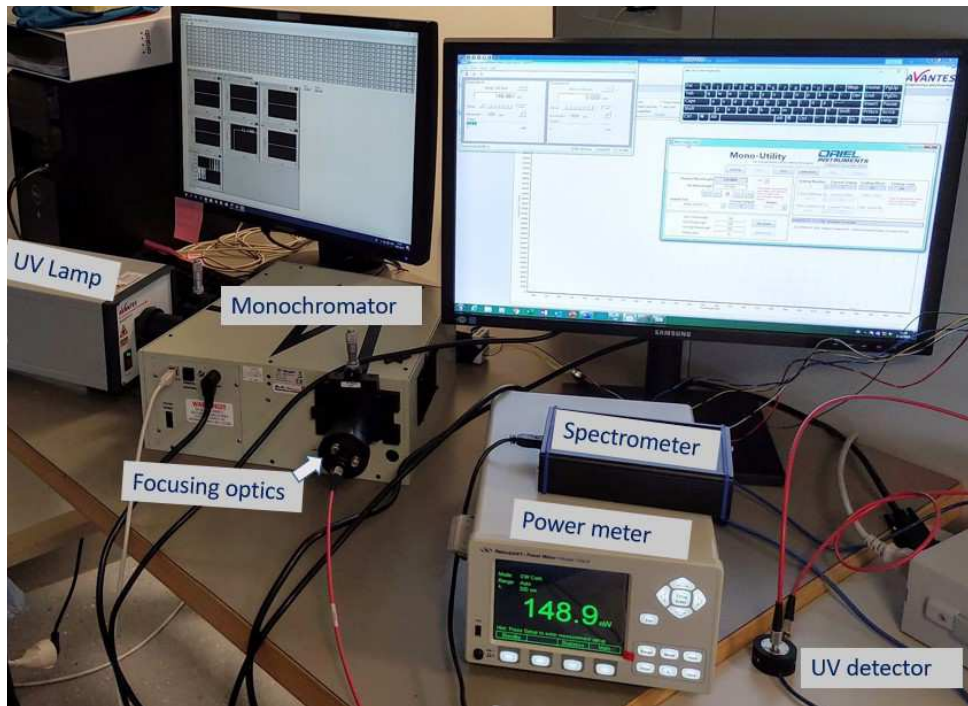
847 *Spectral calibration:*

848 The spectral responsivity of the UV photodiodes is obtained by measuring the response of
849 each photodiode to a narrow wavelength excitation beam of $\sim 5\text{nm}$, at normal incidence and
850 ambient temperature. The UVS spectral calibration setup is shown in Figure 16. It consists of
851 a Deuterium UV lamp (Avantes), focusing optics, UV-VIS monochromator (Newport), filter
852 wheel (Newport) with 3 ND filters, fiber optic cables with beam splitter, a power meter
853 (Newport) along with 818-UV/DB photodetector to measure the strength of UV radiation and
854 a UV/VIS/NIR spectrometer (Avantes). Figure 17 shows the relative spectral responsivity of

855 the UV-A, UV-B, UV-C, UV-D and UV-ABC photodiodes from the spectral calibration tests,
856 the UV-E curve is shown from the data provided by the manufacturer.

857

858

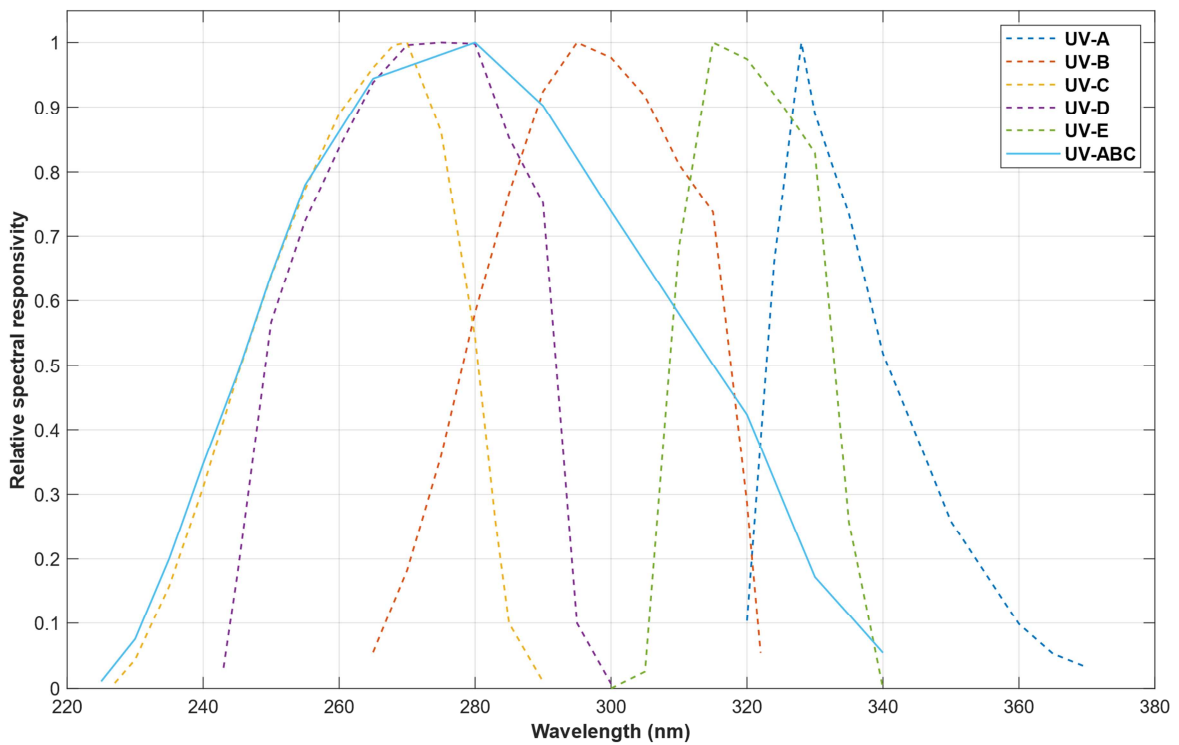


859

860 **Figure 16: UVS calibration workbench consisting of the required instruments for the**
861 **calibration of UV photodiodes.**

862

863



864

865 **Figure 17: Relative spectral responsivity of the HABIT UV photodiodes.**

866

867 *Linearity test:*

868

869 The linearity tests have been performed with the FM at ambient conditions in a cleanroom.

870 The neutral density (ND) filters on the filter wheel housing with optical densities (OD) 0.04,

871 0.1 and 0.5 are used while maintaining all other parameters constant, namely the focusing,

872 bandwidth, and wavelength. Figure 18 shows the normalised linearity of the photodiodes UV-

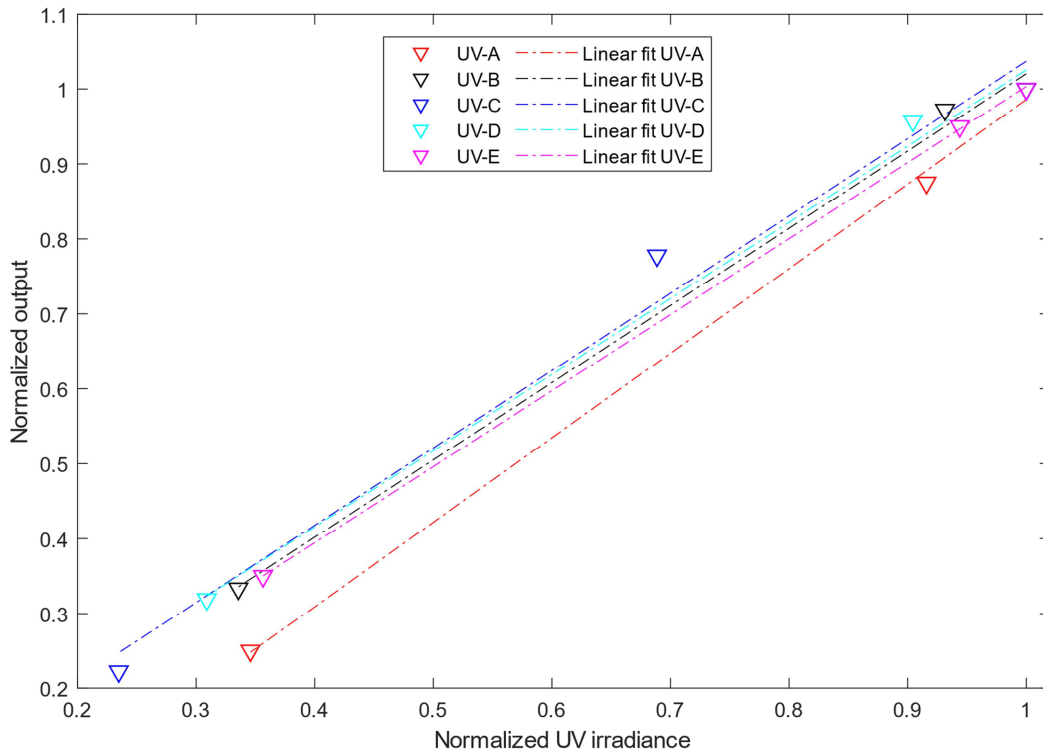
873 A, UV-B, UV-C, UV-D and UV-E. The photodiodes UV-A, UV-B and UV-E show better

874 than 2.5% linearity for the described test conditions, while UVC and UVD show linearity bet-

875 ter than 11.38% and 3.06% respectively. The large errors in linearity shown by the UV-C and

876 UV-D channels compared to the rest are induced due to their sensitivity to the ozone absorp-

877 tions in the air column under Earth laboratory conditions. The linearity calculation errors for
 878 these two photodiodes can be improved in future tests under neutral atmosphere conditions.



879

880 **Figure 18: Normalized linearity graph of UVS photodiodes.** The X-axis shows the normal-
 881 ised UV irradiance power and Y-axis shows the normalised output of the photodiodes and the
 882 dotted lines represent the linear fit curves for the response of the photodiodes.

883

884

885 *R-test:*

886

887 The R-tests are conducted to derive the average responsivity, R_i^{avg} of the UVS photodiodes a
 888 constant value which is wavelength-independent and can be used to obtain in real-time a
 889 quick estimate of the incident irradiance without prior knowledge of the spectral shape of the
 890 incident irradiance. Among the six UVS photodiodes, the UV-A channel has minimum sensi-
 891 tivity to ozone absorption and atmospheric effects from the air column. Next, we describe for

892 illustration the procedure of the R test for the UVA and UVB photodiodes with an engineer-
 893 ing calibration prototype. The sensing area is illuminated with the full-strength signal from
 894 the Deuterium UV lamp using the fibre optic cable and COL-UV/VIS collimating lens at 10
 895 mm distance from the photodiode. The output current from the photodiode is recorded using
 896 HM8112-3 precision multi-meter and the power of the incident UV radiation is recorded us-
 897 ing the power meter 1936-R connected with the 818-UV/DB low power UV enhanced silicon
 898 photodetector. The incident UV power is measured at the wavelengths of the UVA or UVB
 899 spectral band and the mean power is substituted for I_i^{avg} in Equation (1) to calculate the aver-
 900 age spectral responsivity, resulting in $R_A^{avg} = 0.0596 \frac{A}{W}$ and $R_B^{avg} = 0.2438 \frac{A}{W}$. At the time
 901 of submission of this work, the averaged spectral responsivity for the other photodiodes is still
 902 pending calibration. We include for reference an estimate of the order of range of R using the
 903 Full Width at Half Maximum (FWHM) of the spectral response as provided by the manufac-
 904 turer, see Table 2. These values shall be updated with future calibrations of the EQM when
 905 the model returns from IKI after the tests of the ExoMars program. The REMS UV channel-
 906 averaged responses are included for comparison.

907

908 **Table 2:** Reference R^{avg} values of the HABIT-UVS photodiodes.

909

Channel	R_i^{avg} (A/W) HABIT 25°C	R_i^{avg} (A/W) REMS 25°C
UV-A	0.0596	0.0529
UV-B	0.2428	0.1946
UV-C	0.111 (FWHM)	0.0716
UV-D	0.1293 (FWHM)	0.0602
UV-E	0.103 (FWHM)	0.0907
UV-ABC	0.1276 (FWHM)	0.1008

910

911 *Temperature dependence test:*

912

913 The spectral response of the SiC UV photodiodes increases with temperature, approximately
914 1% per degree, due to small variations of the quantum efficiency of SiC detectors and possi-
915 bly also due to changes in the spectral response of the filters. This makes the output shift to-
916 wards longer wavelengths beyond 280 nm. Thus, photodiodes UV-A, UV-E and partly UV-B
917 whose spectral response is higher than 280 nm are more sensitive to the temperature changes
918 (Gomez-Elvira et al., 2012). The responsivity of the electronics can also vary with tempera-
919 ture. The UVS is within the Electronic Unit (EU), on the Surface Platform deck, and it is pro-
920 tected with a heater that is activated when its temperature is below -33°C and disabled if the
921 temperature of the surface platform rises above -30°C , we, therefore, limit the tests to the
922 range -40°C to 25°C (ambient conditions to compare with the rest of calibrations).

923

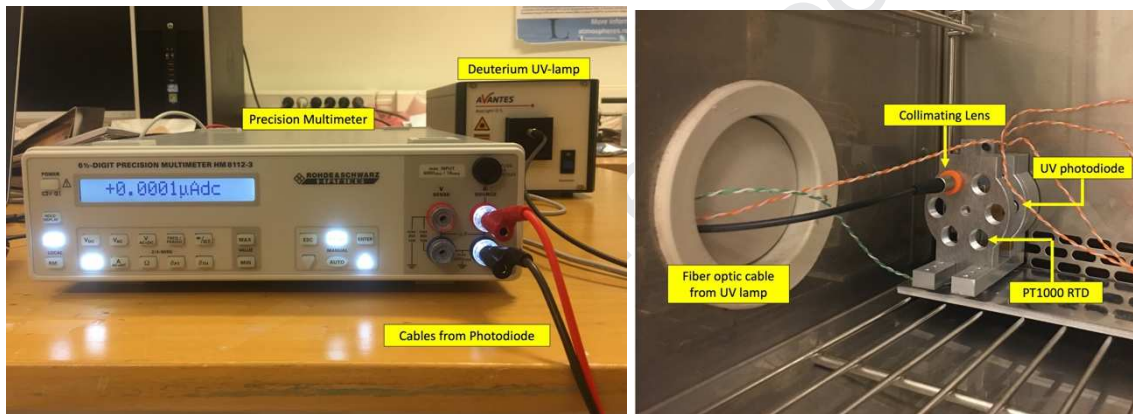
924 Figure 19 shows the temperature dependence test workbench along with the engineering pro-
925 totype inside the thermal chamber with the photodiodes and the collimating lens in normal
926 incidence orientation. This test uses an HM8112-3 precision multi-meter from Rohde and
927 Schwarz that measures the current across the UV photodiodes, a deuterium UV lamp from
928 Avantes and a temperature measurement unit made using Arduino, MAX31865 and PT1000
929 RTD. The UVS engineering calibration prototype houses the same UV photodiodes used in
930 the HABIT flight model. A Heraeus Votsch HT4010 climate chamber is used. A single-mode
931 fibre optic cable of core diameter $800\ \mu\text{m}$ was used to couple the deuterium UV lamp and the
932 collimating lens thereby utilising the full spectrum UV radiation from the UV lamp. The dis-
933 tance between the collimating lens and the UV photodiodes is kept constant at 10 mm during
934 the tests. Sufficient warm-up time is provided for the UV lamp to obtain a stable output be-
935 fore proceeding with the tests. The temperature measurements are obtained from the PT1000

936 RTD sensor mounted close to the UV-photodiodes as shown in Figure 19. The data from the
 937 multi-meter and the PT1000 temperature sensor is logged at a frequency of 1 Hz. Figure 20
 938 shows, an illustration, the relative change of the response with respect to the nominal ambient
 939 response for UV-ABC and UV-C. Once on Mars, for a given temperature of the photodiodes,
 940 the temperature correction factor can be interpolated from these data.

941

942

943

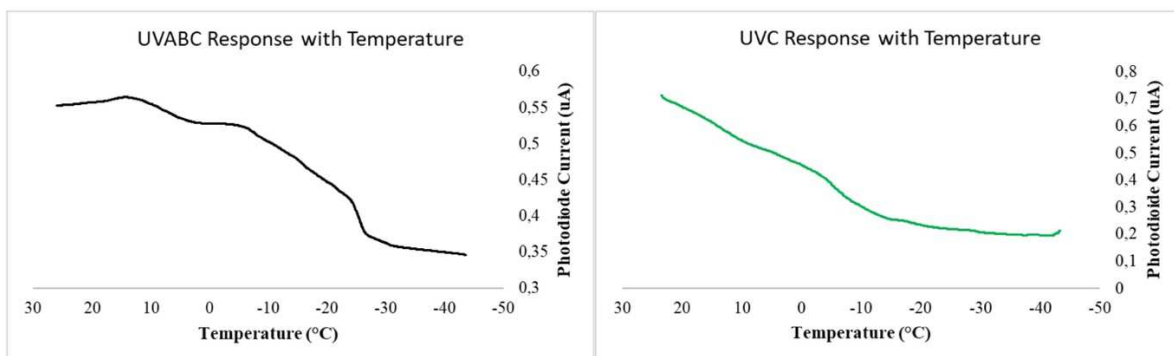


944

945 **Figure 19: (left) Temperature dependence workbench (right) UVS engineering proto-**
 946 **type inside the climate chamber.**

947

948



949

950

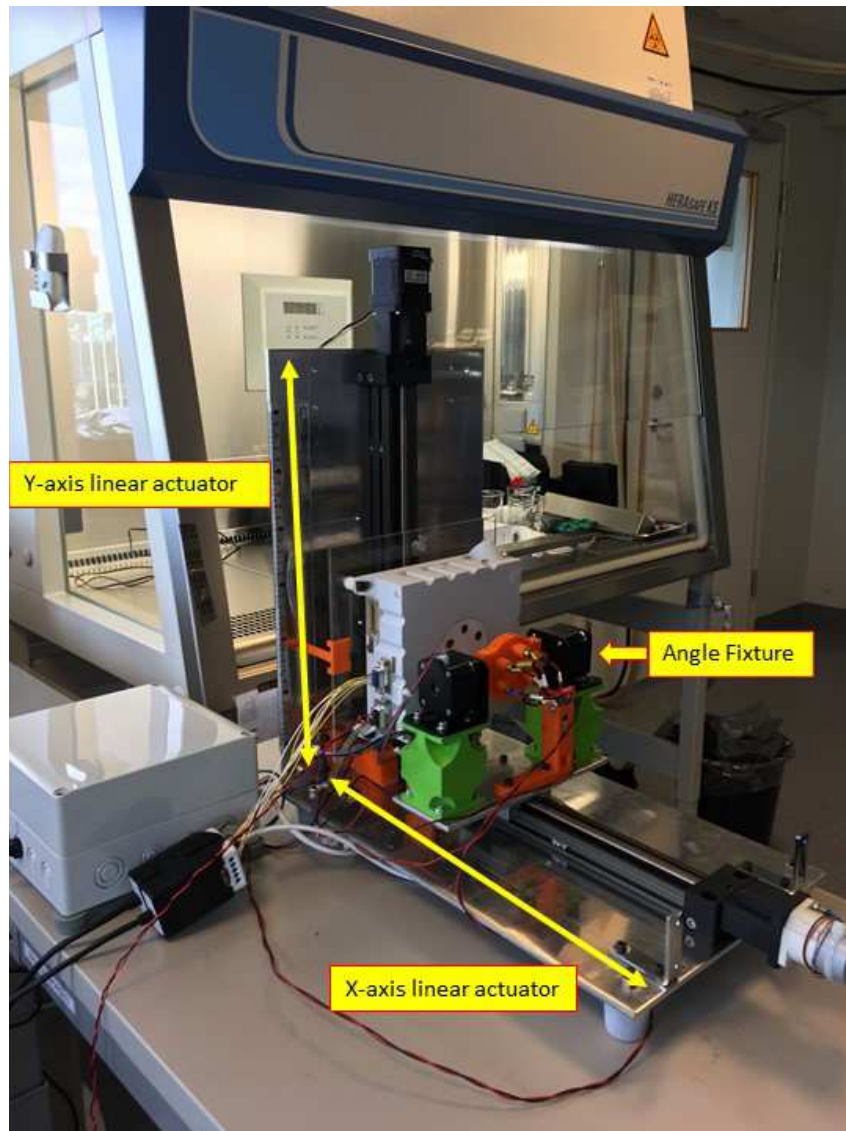
Figure 20: Variation of the UV photodiode current response with temperature.

951 *Incidence angle dependence test:*

952

953 The direct beam of UV radiation from the Sun reaches the photodiode with an angle of inci-
954 dence that varies as the Sun moves in the sky. It depends on the hour of the day and the sea-
955 son, and of course on the landing site and surface platform inclination. This information can
956 be obtained as ancillary data during the operation on Mars. The REMS UVS photodiodes fol-
957 low a $\cos^a(\theta)$ response, as long as the direct beam is within the nominal field of view of +/-
958 30° (Zorzano et al., 2009; Gómez-Elvira et al., 2012). This exponent is next calibrated for
959 HABIT. To modify precisely the angle of incidence, a dedicated fixture was designed, see
960 Figure 21. It consists of a twin linear actuator, a screw-driven mechanism and a micro-stepper
961 motor with an open-loop control in the x-y axis.

962



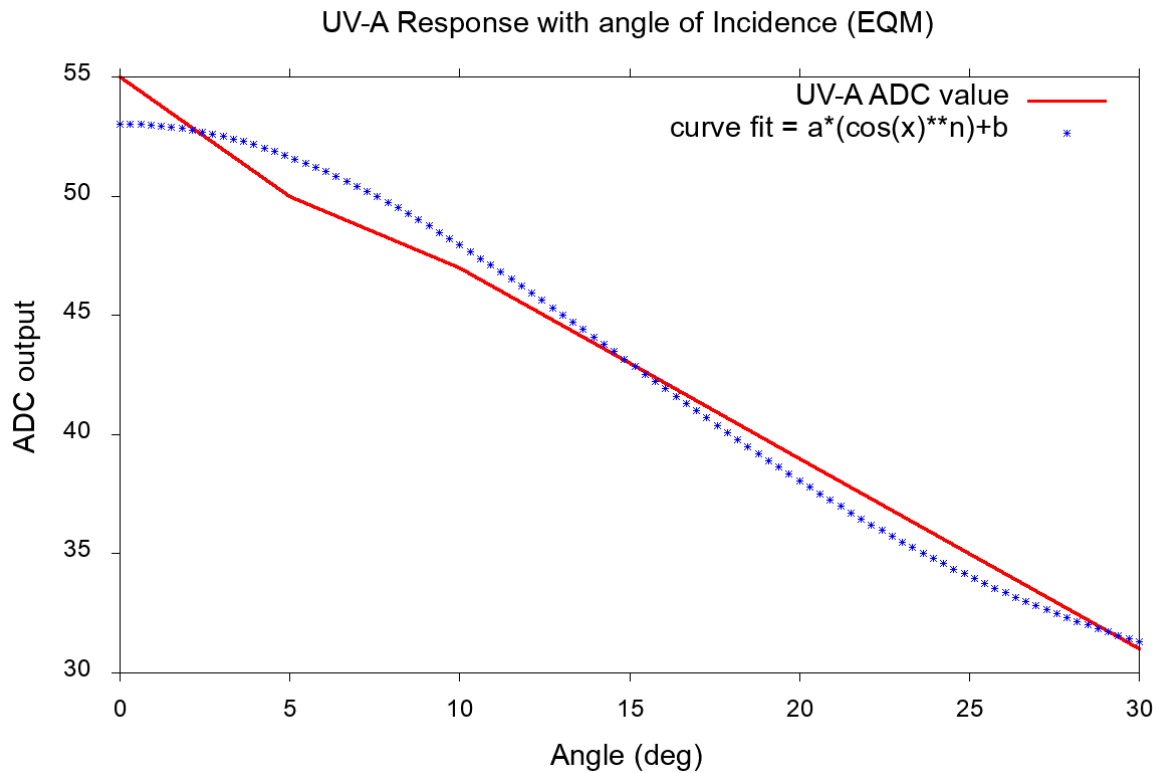
963

964 **Figure 21: Mechanical fixture for incidence angle dependence tests on which the EU of**
 965 **HABIT EQM is fixed.**

966 The angle of incidence test was performed at ambient conditions for 0 to 30° incidence angle,
 967 at increments of 5°. A 340 nm 0.33 mW UV LED 341 nm from Thor labs was utilised as the
 968 UV source during the incidence angle test. The distance between the UV photodiode and the
 969 UV LED was accurately maintained at 6 cm. Figure 22 shows an example of the calibration
 970 curve, and the fit to $a \cos^n(\theta) + b$ for the UVA photodiode. The parameters of the fit for the
 971 UVA sensor are: $n_{UVA}=15$, $a_{UVA}=24$, $b_{UVA}=29$.

972

973



974

975

Figure 22: UV-A photodiode cosine angle dependence response.

976

977

978

6. Brine Observation Transition To Liquid Experiment

979

(BOTTLE)

980

981

6.1 Design and operation of BOTTLE

982

The Brine Observation Transition To Liquid Experiment (BOTTLE) shall monitor the for-

983

mation and stability of liquid brines on Mars. It consists of 6 cells, shown in Figure 23. Four

984

of them are exposed to the Mars environment and contain deliquescent salts that have been

985

found on Mars (calcium-chloride CaCl_2 , ferric-sulphate $\text{Fe}_2(\text{SO}_4)_3$, magnesium-perchlorate

986 $\text{Mg}(\text{ClO}_4)_2$, and sodium-perchlorate NaClO_4). The other 2 cells are empty will be used to
987 monitor the air electrical conductivity and collect atmospheric dust.

988
989 The salts are exposed to the martian environment through a High Efficiency Particulate Air
990 (HEPA) filter (to comply with planetary protection protocols). The deliquescence process of
991 salts will be monitored by observing the changes in the electrical conductivity (EC) of each
992 cell with the electrode pairs at three different levels (denoted as low, mid and high). Gold
993 electrodes are used for better resistance to corrosion (Nazariou et al., 2019). This technique
994 has been commonly used for the detection of liquid or frozen water in soils (McKay et al.,
995 2003; Davis et al., 2010). Previous studies have suggested electrical conductivity as a good
996 method to monitor the deliquescence process of salts (Yang et al., 2006; Heinz et al., 2016).
997 The electrical conductivity measurements correlated well with the deliquescence rates of mix-
998 tures of martian analogue soil and perchlorates or chlorides. The electrical conductivity meth-
999 od provides a time series and may be a good indicator of the brine formation process, with
1000 higher sensitivity than either Raman spectroscopy or estimates based on deliquescence rela-
1001 tive humidity (Heinz et al., 2016). The salts of BOTTLE have the property of absorbing water
1002 from the martian atmosphere, forming different hydrates, and brines, depending on the tem-
1003 perature and relative humidity; and we expect to identify the phase changes from dry salt to
1004 its hydrate forms, to liquid brine or to frozen brine.

1005
1006 BOTTLE is equipped with a PT1000 temperature sensor to serve as reference and a 12 W
1007 heater (at Pulse Width Modulation with 100% duty cycle) to allow recyclability of salts. The
1008 purpose of the heater will be to: 1) dehydrate the liquid brine to anhydrous salt and restart the
1009 experiment, and 2) provide inflight calibration for detecting any degradation in the hardware
1010 and correct the offsets. The experiments on BOTTLE will show the absorption and release of

1011 water and will serve to demonstrate the ISRU potential of the instrument to collect water for
1012 the future exploration of Mars.

1013

1014 **6.2 Measurements and calibration of BOTTLE EQM and FM**

1015 BOTTLE measures the current flow, I and the voltage drop, V across low, mid and high elec-
1016 trode pairs (see figure 23) of each of the six cells, along with the CU temperature. Conse-
1017 quently, the conductance, G in Siemens is calculated with equation (27).

1018

$$1019 \quad G = \frac{I}{V} \text{ (S)} \quad (27)$$

1020

1021 BOTTLE uses a 2.048V bias voltage and have two mode of conductance measurements. The
1022 high conductance mode uses a 70 mV electric pulse and a 100 Ω current sense resistor while
1023 the low conductance mode uses a 700 mV electric pulse to generate enough current for meas-
1024 urements and a 10 k Ω resistor. At the beginning of every data acquisition session, BOTTLE
1025 measures in high conductance mode by default. The mode is switched to low, when the meas-
1026 urement is out of range. The first 8 samples are used as a filter to make a smooth transition
1027 between the modes and thus these 8 measurements must be discarded for interpretation.

1028

1029 Laboratory experiments for evaluating the coefficients of the calibration function were per-
1030 formed for 100 data points at a frequency of 1000 ms. The measured conductance of the sam-
1031 ple is translated to electrical conductivity by multiplying with the geometrical cell constant of
1032 the electrode pair, K_{cell} as in equation (28).

1033

$$1034 \quad \sigma_{measured} = K_{cell} \times G \text{ (}\mu\text{Scm}^{-1}\text{)} \quad (28)$$

1035

1036 The geometrical cell constant of the electrode pair depends on its physical dimensions. There
1037 are three electrode pairs each in four middle cells and two electrode pairs each in the two end
1038 cells summing to a total of 16 electrode pairs and consequently 16 geometrical cell constants
1039 are to be calculated with equation (29).

1040

$$1041 \quad K_{cell} = \frac{\text{Distance between electrodes (in cm)}}{\text{Surface area of electrodes (in cm}^2\text{)}} = \frac{d}{A} \quad (29)$$

1042

1043 where, $A_{low} = 1.6 \text{ cm} \times 0.4 \text{ cm} = 0.64 \text{ cm}^2$

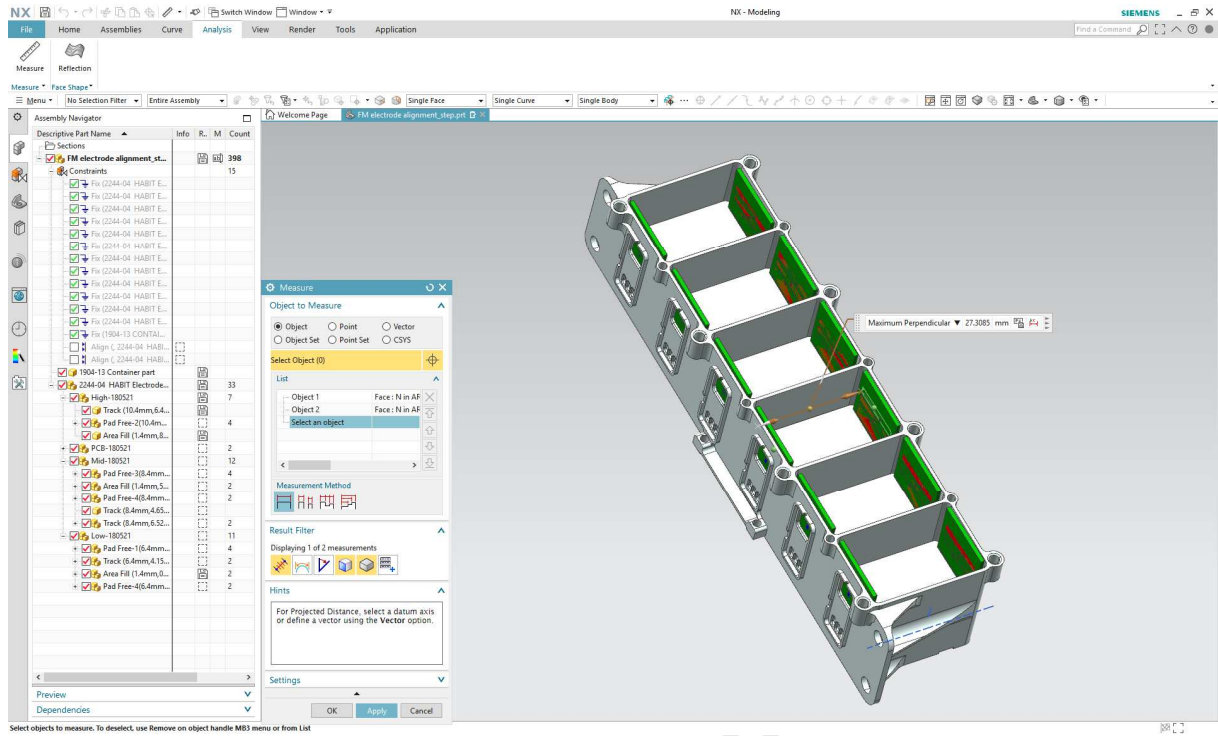
1044 $A_{mid} = 1.6 \text{ cm} \times 0.2 \text{ cm} = 0.32 \text{ cm}^2$

1045 $A_{high} = 1.6 \text{ cm} \times 0.2 \text{ cm} = 0.32 \text{ cm}^2$

1046

1047 The distance between the electrode pairs are measured using an optical measurement system
1048 (Mitutoyo MF 176) to verify the alignment of the electrodes and to obtain an accurate geo-
1049 metrical cell constant of the electrode pairs. The minimum and maximum distances between
1050 each of the electrode pairs are averaged. A screenshot of the procedure used to measure the
1051 distances in the CAD model constructed with the optical measurements is shown in Fig. 23
1052 and the computed cell constants are provided in Table 3.

1053



1054

1055 **Figure 23:** CAD model of BOTTLE flight model generated using the measurements from the
 1056 optical measurement system.

1057

1058 **Table 3:** Computed geometrical cell constants of all the electrode pairs.

Electrode designation	d (cm)	A (cm ²)	K _{cell} (cm ⁻¹)
Cell-1, Low	2.7124	0.64	4.2381
Cell-1, Mid	2.6985	0.32	8.433
Cell-2, Low	2.7361	0.64	4.2752
Cell-2, Mid	2.7335	0.32	8.5421
Cell-2, High	2.7313	0.32	8.5354
Cell-3, Low	2.7412	0.64	4.2832
Cell-3, Mid	2.7352	0.32	8.5476
Cell-3, High	2.7312	0.32	8.5351
Cell-4, Low	2.7363	0.64	4.2755
Cell-4, Mid	2.7322	0.32	8.5382
Cell-4, High	2.7289	0.32	8.5279

Cell-5, Low	2.7402	0.64	4.2816
Cell-5, Mid	2.7312	0.32	8.5349
Cell-5, High	2.7246	0.32	8.5144
Cell-6, Low	2.7294	0.64	4.2646
Cell-6, Mid	2.7226	0.32	8.5082

1059

1060

1061 **6.2.1 Zero offset or dry point**

1062 The purpose of this experiment is to obtain the zero or dry point for determining the calibra-
 1063 tion function coefficients. This test was performed with the BOTTLE flight model, inside a
 1064 vacuum chamber at a temperature between 26.27 °C and 26.32 °C and measuring the empty
 1065 cells with only the natural noise frequency of the instrument electronics. The measured elec-
 1066 trical conductivity in the vacuum conditions is used as a reference for the absolute zero meas-
 1067 urement $0.00364 \mu\text{Scm}^{-1}$, averaged across all the low electrodes.

1068

1069 **6.2.2 Calibration function and coefficients**

1070 The calibration function was determined with several calibration standards with known elec-
 1071 trical conductivity values. During this calibration procedure, we used the conductivity stand-
 1072 ards: $84 \mu\text{Scm}^{-1}$, $1413 \mu\text{Scm}^{-1}$, $5000 \mu\text{Scm}^{-1}$, $12880 \mu\text{Scm}^{-1}$, $80000 \mu\text{Scm}^{-1}$ and $111800 \mu\text{Scm}^{-1}$
 1073 ¹ as shown in Figures 24 and 25. As we will show later in the experiments of the SpaceQ, our
 1074 EC range of interest is from 0 to $1000 \mu\text{Scm}^{-1}$, as the maximum EC we have observed with a
 1075 deliquescent forming brine under martian conditions is of the order of $100 \mu\text{Scm}^{-1}$. We in-
 1076 crease the region of interest by one order of magnitude.

1077

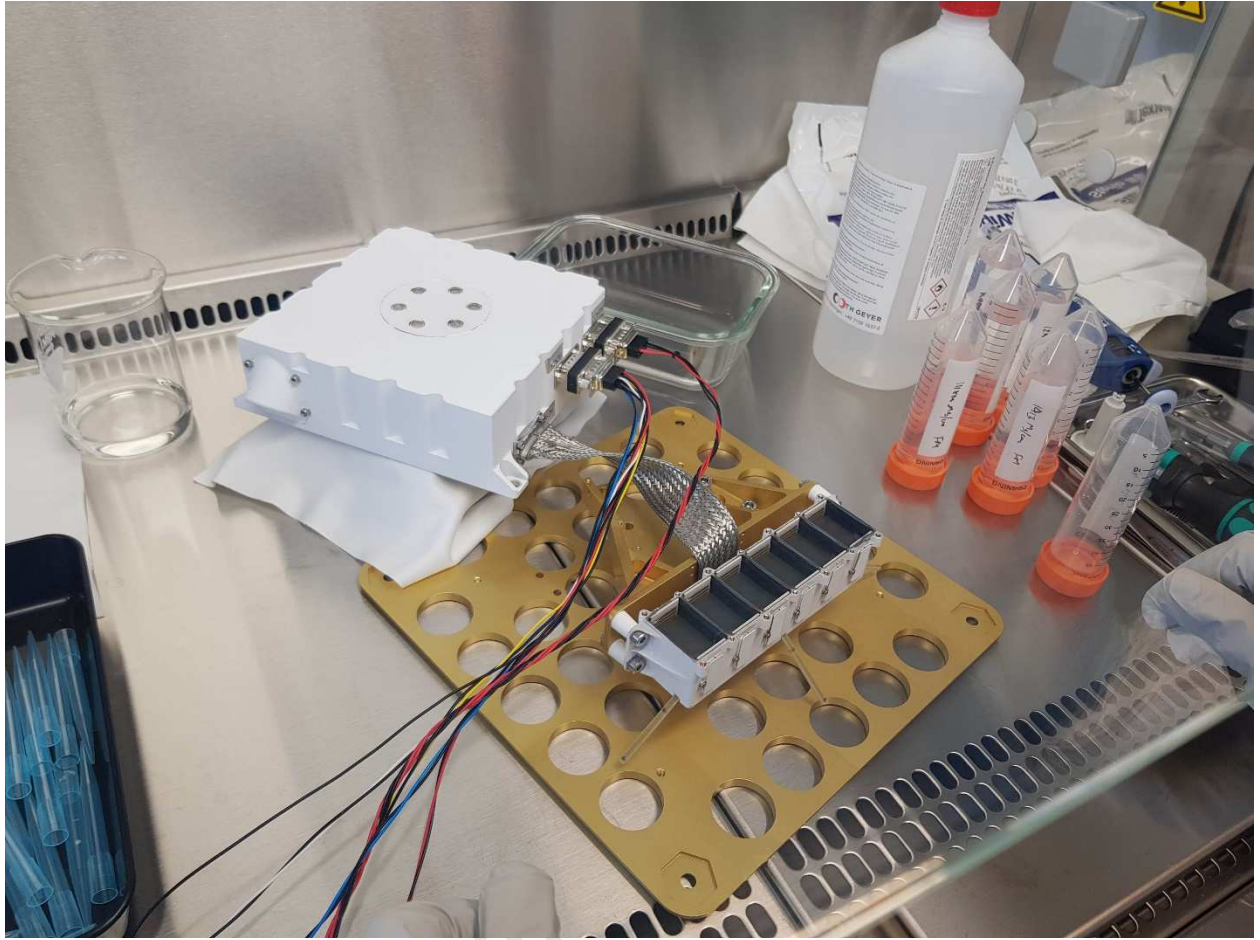


1078

1079 **Figure 24:** Procedure to fill the calibration standards in BOTTLE cells in a clean laminar

1080 fume cabinet at ambient temperature.

1081



1082

1083 **Figure 25:** A view of the BOTTLE flight model in the laminar fume cabinet with calibration
 1084 standards filled in the cells.

1085 The calibration function was determined from these calibration standard measurements with a
 1086 least square regression, by fitting a 2nd order polynomial curve as in equation (30). For a bet-
 1087 ter interpretation of the BOTTLE measurements, the specific choice of the two calibration
 1088 standards will be chosen depending on the required accuracy within a certain measurement
 1089 range (two-point calibration).

1090

1091
$$\sigma_{actual} = a_2 \sigma_{measured}^2 + a_1 \sigma_{measured} + a_0 (\mu S cm^{-1}) \quad (30)$$

1092

1093 We obtain 16 calibration function coefficients, where each corresponds to an electrode pair.

1094 The advantage of having multiple calibration points is to use them depending on the desired

1095 measurement range and to fine-tune the resolution of measurements within a specified range.

1096 By default, we use the calibration function derived with zero point, $84 \mu\text{Scm}^{-1}$ and 1413

1097 μScm^{-1} (two-point calibration). But, if the measured EC is over $1413 \mu\text{Scm}^{-1}$, additional cali-

1098 bration points ($5000 \mu\text{Scm}^{-1}$) can be included to extend the measurement range of the instru-

1099 ment. Also, in order to measure the $2000 \mu\text{Scm}^{-1}$ accurately, a new set of calibration function

1100 coefficients can be computed only using the $1413 \mu\text{Scm}^{-1}$ and $5000 \mu\text{Scm}^{-1}$ points. Figure 26

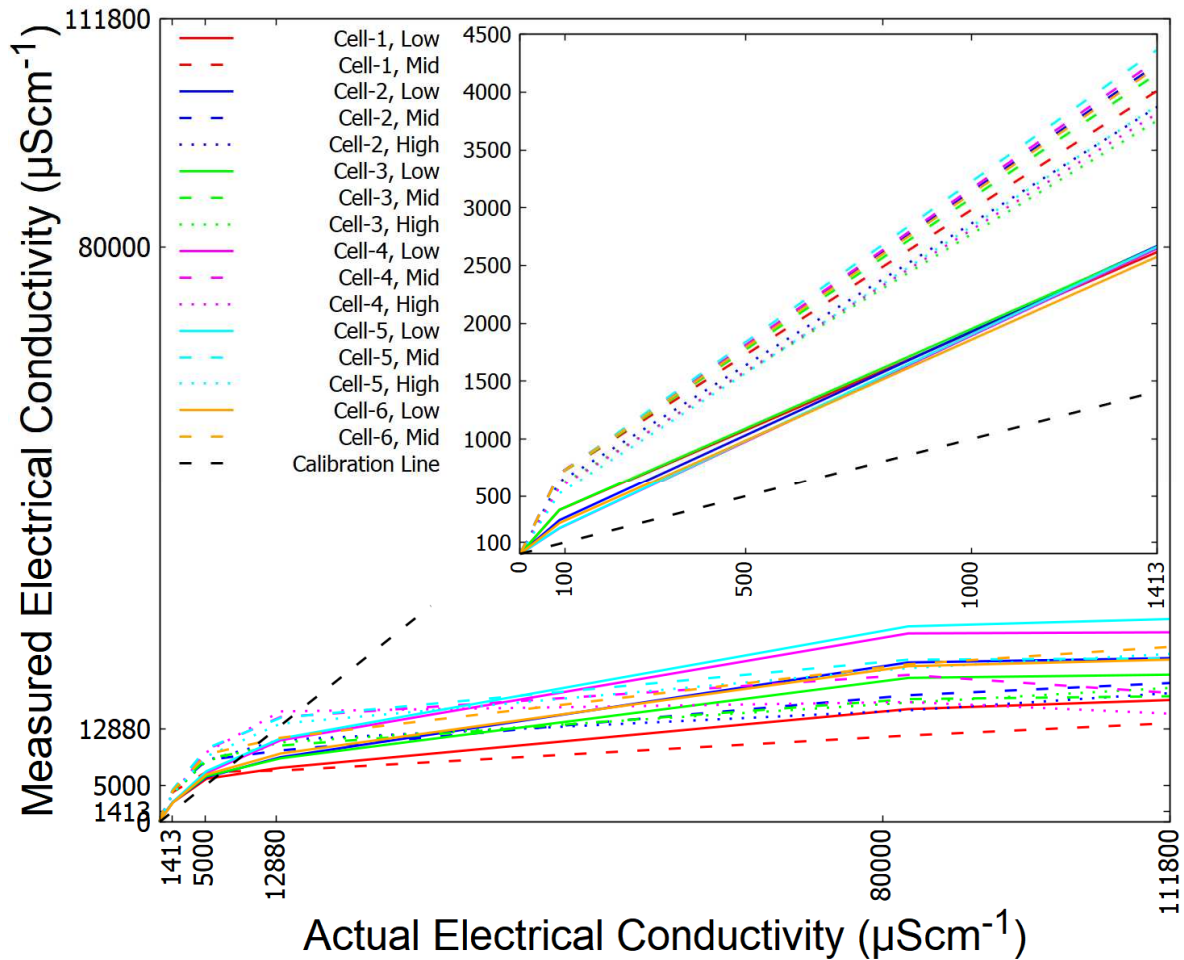
1101 shows the full extent of the calibration points with the 0 to $1413 \mu\text{Scm}^{-1}$ plot shown in the in-

1102 set.

1103

1104

1105



1106

1107 **Figure 26:** A comparison of the measured electrical conductivity by BOTTLE flight model
 1108 with the actual electrical conductivity of the calibration standards, where the standard solu-
 1109 tions were covering until the high electrode. Similarly, EC measurements were carried out for
 1110 low and mid electrode coverages (not shown in this figure). Lower conductivities show better
 1111 accuracy with the calibration standards while the higher conductivities deviate from the cali-
 1112 bration line (black). (inset) A comparison of the measured electrical conductivity by BOTTLE
 1113 flight model with the actual electrical conductivity of the calibration standards in the lower
 1114 electrical conductivity range between 0 and 1413 μScm^{-1} .

1115 We compute the calibration function coefficients by covering the three levels of electrode for
 1116 each conductivity standard. Therefore, we have three sets each of 16 calibration function co-
 1117 efficients (total of 48) for low, mid and high electrode coverage. While converting the raw

1118 measured electrical conductivity to the actual values, one of these set of coefficients will be
 1119 utilized depending on the coverage of salts in the low, mid or high electrode. In its initial con-
 1120 figuration during FM delivery, the salts covered until low electrode and thus the 16 calibra-
 1121 tion function coefficients corresponding to low electrode coverage will be initially used. As the
 1122 experiment progresses and the brine level increases, the use of a different set of coefficients
 1123 (mid or high) may be recommended. The information about the coefficients used will be
 1124 flagged in the HABIT PSA data bundle. Table 4 shows the 48 calibration function coeffi-
 1125 cients.

1126

1127

1128

1129

1130 **Table 4: Calibration function coefficients for each electrode pair with low, mid and high**
 1131 **electrode coverages.**

Electrode designation	Electrode coverage								
	Low			Mid			High		
	a_2	a_1	a_0	a_2	a_1	a_0	a_2	a_1	a_0
Cell-1, Low	0.000287	0.333303	-0.01185	0.000170	0.216503	-0.00770	0.000134	0.178402	-0.00634
Cell-2, Low	0.000272	0.299276	-0.00790	0.000127	0.289159	-0.00764	9.315e-05	0.271738	-0.00718
Cell-3, Low	0.000283	0.329932	-0.01238	0.000161	0.218174	-0.00819	0.000128	0.179918	-0.00675
Cell-4, Low	0.000308	0.285631	-0.00642	0.000162	0.214088	-0.00481	5.559e-05	0.380646	-0.00855
Cell-5, Low	0.000283	0.295427	-0.01459	0.000161	0.214926	-0.01061	5.529e-05	0.377817	-0.01866
Cell-6, Low	0.000294	0.312369	-0.01462	0.000172	0.207752	-0.00973	9.218e-05	0.302299	-0.01415
Cell-1, Mid	0	0	0	8.917e-05	0.111435	-0.00285	6.612e-05	0.078964	-0.00202
Cell-2, Mid	0	0	0	7.842e-05	0.118968	-0.00778	5.692e-05	0.084491	-0.00552
Cell-3, Mid	0	0	0	8.426e-05	0.115198	-0.00244	5.966e-05	0.084016	-0.00178
Cell-4, Mid	0	0	0	7.381e-05	0.123625	-0.01022	5.539e-05	0.086063	-0.00712

Cell-5, Mid	0	0	0	7.175e-05	0.116869	-0.00487	5.207e-05	0.089149	-0.00371
Cell-6, Mid	0	0	0	7.782e-05	0.117713	-0.00508	5.758e-05	0.084977	-0.00367
Cell-2, High	0	0	0	0	0	0	6.716e-05	0.09683	-0.00480
Cell-3, High	0	0	0	0	0	0	7.004e-05	0.105298	-0.00788
Cell-4, High	0	0	0	0	0	0	6.535e-05	0.113147	-0.00632
Cell-5, High	0	0	0	0	0	0	5.676e-05	0.136389	-0.01211

1132

1133 6.3 Validation

1134 This experiment is designed to validate the evaluated calibration function. Using zero, 84

1135 μScm^{-1} and 1413 μScm^{-1} calibration points to determine the calibration function coefficients,

1136 we applied it back to the measurements with the 1413 μScm^{-1} calibration standard solution.

1137 Figures 27 to 29 shows the calibrated electrical conductivity values of 1413 μScm^{-1} after ap-

1138 plying the calibration function. For this experiment the mean EC error with respect to the

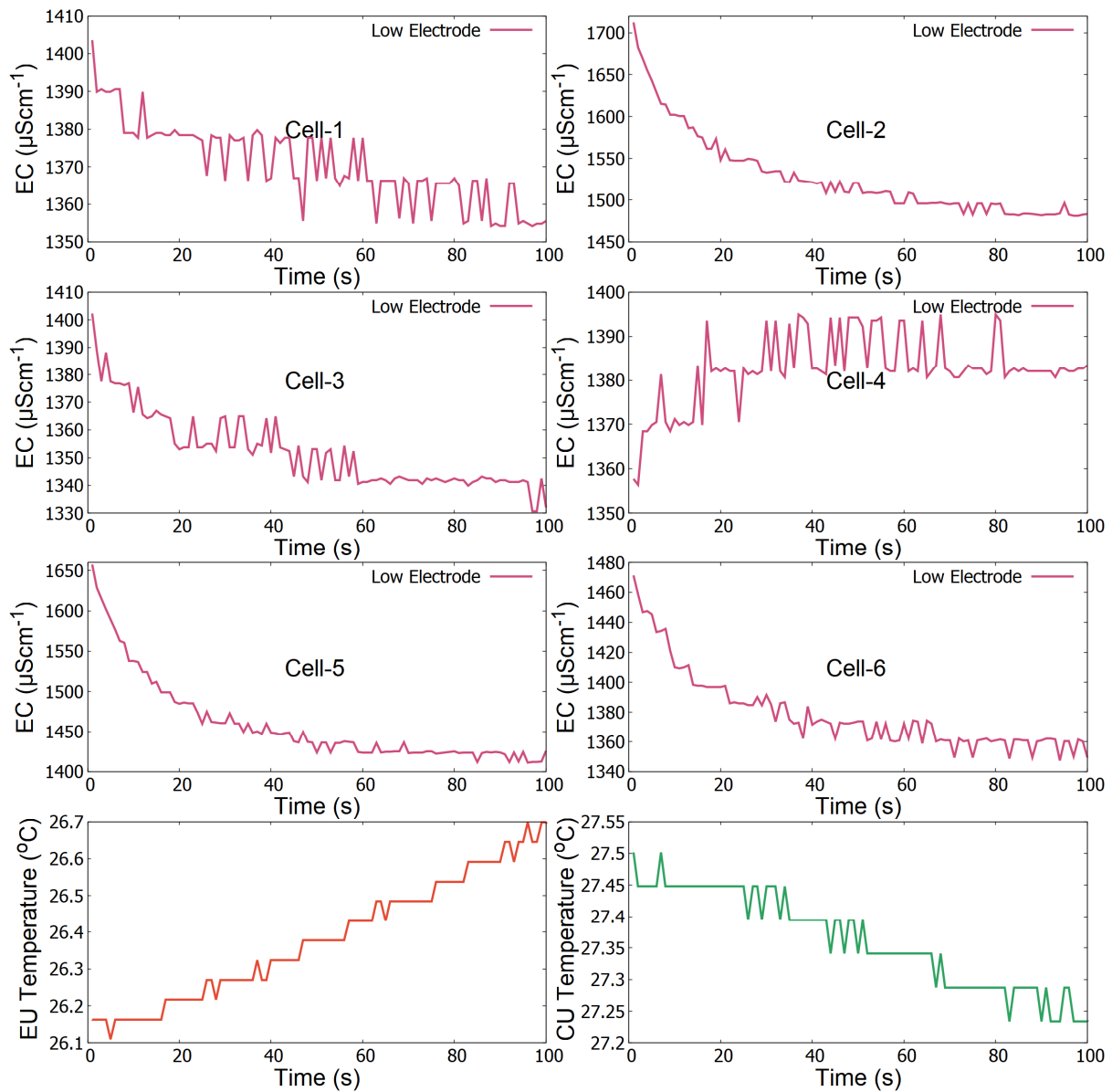
1139 standard was computed to be +70/-90 μScm^{-1} (averaged between low, mid and high electrode

1140 measurements).

1141

1142

1143



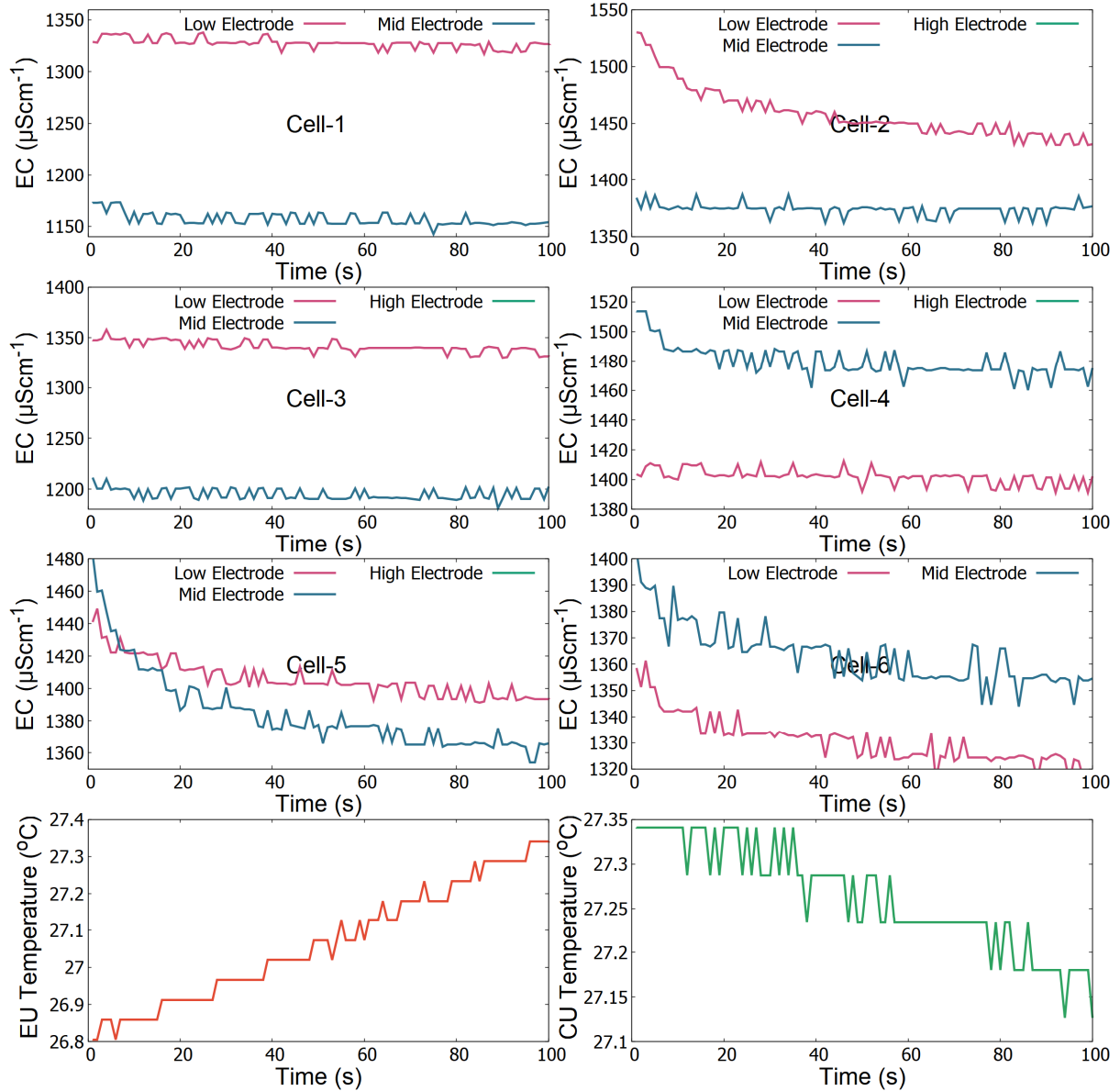
1144

1145 **Figure 27:** Measurements of the $1413 \mu\text{Scm}^{-1}$ calibration standard covering the low electrode.

1146 The actual electrical conductivity was computed using the calibration function coefficients

1147 corresponding to the low electrode coverage. Only the low electrode EC measurements are

1148 shown for clarity.



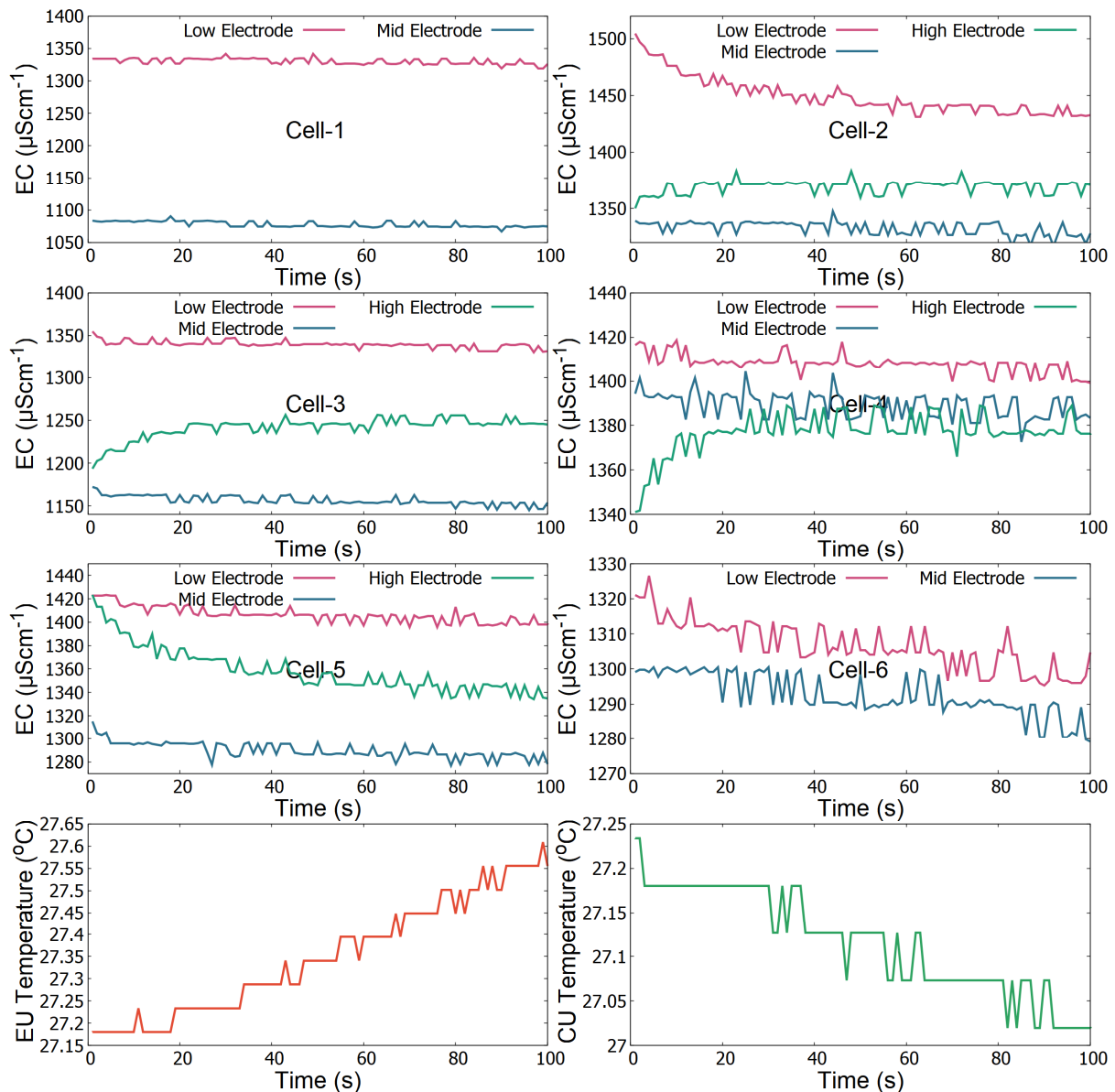
1149

1150 **Figure 28:** Measurements of the $1413 \mu\text{Scm}^{-1}$ calibration standard covering the mid electrode.

1151 The actual electrical conductivity was computed using the calibration function coefficients

1152 corresponding to the mid electrode coverage. Only the low and mid electrode EC measure-

1153 ments are shown for clarity.



1154

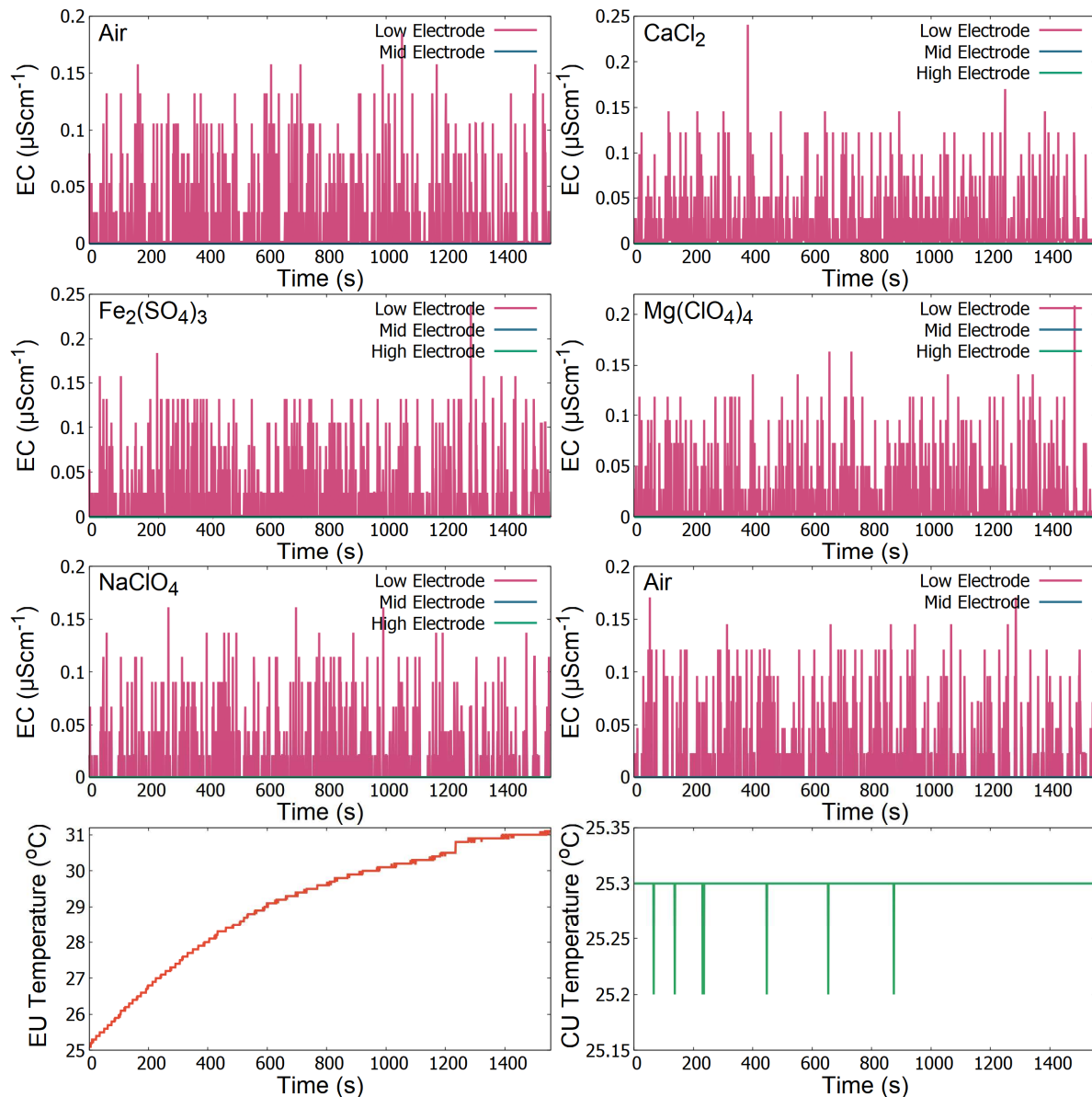
1155 **Figure 29:** Measurements of the $1413 \mu\text{Scm}^{-1}$ calibration standard covering the high elec-
 1156 trode. The actual electrical conductivity was computed using the calibration function
 1157 coefficients corresponding to the high electrode coverage. EC measurements are
 1158 shown for all electrodes.

1159 6.5 Clean-room measurement of dry salts in FM

1160 Figure 30 shows an example of BOTTLE measurements performed with the FM in the clean-
 1161 room of Thales Alenia Space-France (TAS-F) during integration in the Surface Platform. The
 1162 EC values are of the order of $\sim 0.1 \mu\text{Scm}^{-1}$, for all the containers, including the ones that only

1163 have sealed ambient air. This demonstrates the proper sealing of the containers, and the dry
1164 state of the salts. When the EC of the salts is equal (or lower) than that of the pure air, that
1165 shows that no hydration has taken place, and the dominant component of the EC is the one of
1166 the air voids between the salt grains. On the other hand, the EC of the air depends on the
1167 amount of water, which in turn depends on the pressure (Earth or martian) and the relative
1168 humidity. The reported value of $0.1 \mu\text{Scm}^{-1}$ is the one we expect under Earth environmental
1169 conditions. However, as mentioned above this value is as low as $0.00364 \mu\text{Scm}^{-1}$ for vacuum
1170 tests in the thermal vacuum facility. And as we shall show later on, this value goes to values
1171 as low as $10^{-8} \mu\text{Scm}^{-1}$ when the experiments are run at martian pressures, and any remaining
1172 water molecule is flushed with a dry CO_2 atmosphere.

1173



1174

1175 **Figure 30:** BOTTLE measurement of 1) electric conductivity of the empty containers with
 1176 Earth ambient air, and the containers with salts and Earth ambient air. These data were meas-
 1177 ured on 2019/12/06 in the clean-room of TAS-F, during integration on the Surface Platform.

1178

1179 Just before launch, the sealing lid will be replaced with the BOTTLE lid with the HEPA filter
 1180 that will allow the atmospheric moisture to interact with the salts. We will continue to monitor
 1181 the EC values during the three health check campaigns that are foreseen to take place during
 1182 the cruise phase, where this value is expected to drop to the minimum level as the containers

1183 will be exposed to vacuum. The environmental conditions during the cruise will have no im-
1184 plications on the operation of HABIT on Mars. In the uneventful situation (most of the terres-
1185 trial water will be lost during the depressurization at launch) that SAP still holds frozen water,
1186 the water will be released once that the mission lands on Mars, because HABIT will be reset,
1187 i.e., the heaters will be switched on until the minimum value of conductivity is reached. Fu-
1188 ture work in the Mars simulation chamber will include a proper characterization of the chang-
1189 es of the martian air EC with varying RH% under martian CO₂ atmospheric conditions and
1190 with martian surface pressures.

1191

1192

1193

1194 **7. BOTTLE products**

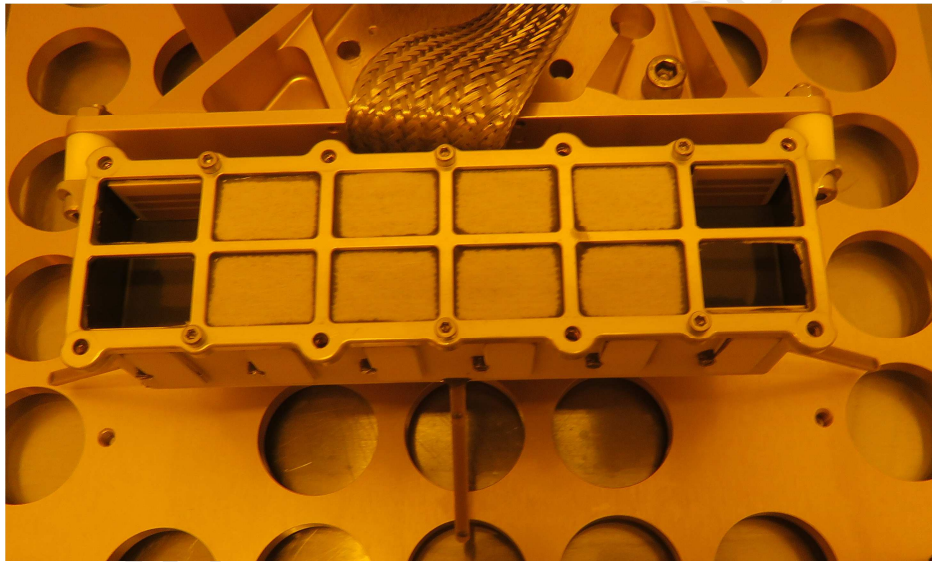
1195

1196 All the products were purchased from Sigma Aldrich and are summarized in Table 5. The
1197 four deliquescent salts that are contained in the cells of BOTTLE in the HABIT FM have
1198 been found on Mars, and their phase diagrams allow to hold liquid conditions under the ex-
1199 pected temperature range of Oxia Planum. Additionally, these salts have been chosen because
1200 their response to the Earth environment is such that it allows for long storage under clean
1201 room conditions without initiating deliquescence into liquid. For each salt, the lower tempera-
1202 ture which can hold liquid state is indicated in Table 6. The salts are mixed in a 2:1 ratio with
1203 a Super Absorbent Polymer (SAP) hydrogel. SAP is made of cross-linked networks of hydro-
1204 philic polymer, which on Earth are capable of absorbing from liquid water, 400 times approx-
1205 imately or more of its weight. The SAP used in HABIT is the Poly (Acrylamide-co-Acrylic
1206 Acid) with a grain size ranging from 200-1000 μm . The containers are separated from the at-

1207 mosphere through a HEPA filter, to comply with Planetary Protection requirements. The
1208 HEPA filter allows the free flow of molecules of the air while impeding the possible flow of
1209 micro-organisms and particulate of size greater than 0.3 μm . This is commonly used in clean
1210 room facilities and is also implemented in the REMS pressure sensor chimney, as a filter for
1211 the air towards the inner part of the rover. Figure 31 shows the HEPA filter assembly of the
1212 Flight Model HABIT- BOTTLE experiment.

1213

1214



1215

1216 **Figure 31:** HEPA Filter Assembly of HABIT glued and screwed to the BOTTLE structure
1217 using six M2 bolts.

1218

1219 The SAP when mixed with a salt sample, as shown in Figure 32, supports the normal
1220 deliquescence of the salt and induces the formation a viscous jelly state, as observed in Figure
1221 33, which inhibits the capillarity of the salts along the walls of the container. Additionally, the
1222 mixture with SAP allows to measure the electrical conductivity and the repeated absorption
1223 and release of water vapour with the atmosphere as in a porous regolith or soil matrix, mim-
1224 icking the behaviour of the martian regolith. A set of tests (not shown) have been applied to

1225 these mixtures to demonstrate that they retain their water absorption properties, after applica-
 1226 tion of the Planetary Protection Protocols of Dry Heat Microbial-bioburden Reduction
 1227 (DHMR) and the depressurization caused during the launch and cruise phase (Mathanlal et.
 1228 al, 2020). Additionally, the swelling ratio of sterilized and depressurized SAP was compared
 1229 and demonstrated to be equal to the unperturbed commercial SAP (not shown). This demon-
 1230 strates the capability of SAP to be used for water storage in space applications, in particular at
 1231 martian pressure conditions and exposed to diurnal and seasonal thermal changes (Mathanlal
 1232 et. al, 2020).

1233
 1234 **Table 5: BOTTLE products**

Cell #	Product formula	Product name	Total dry weight (g)
1	Empty air/dust	Air at martian pressures and collected mineral dust	-
2	CaCl ₂ : C ₆ H ₈ KNO ₃	Calcium chloride + SAP	2.25
3	Fe ₂ (SO ₄) ₃ : C ₆ H ₈ KNO ₃	Ferric Sulphate +SAP	2.25
4	Mg(ClO ₄) ₂ : C ₆ H ₈ KNO ₃	Magnesium Perchlorate + SAP	2.25
5	NaClO ₄ :	Sodium Perchlorate + SAP	2.25

1235

	$C_6H_8KNO_3$		
6	Empty air/dust	Air at martian pressures and collected mineral dust	-

1236

1237

Table 6: Eutectic temperatures and relative humidity's of the chosen salts.

1238

Salt	Te (K)	RH (%)	Reference
$NaClO_4$	236	53	Hennings et al. (2013a,b).
$CaCl_2$	226	60	Toner et. al., 2014
$Mg(ClO_4)_2$	206	52	Stillman and Grimm (2011)
$Fe_2(SO_4)_3$	246.35	40	Hennings et al., 2013a

1239



1240

1241 **Figure 32:** Illustration of the state of the products after DHMR and depressurization, as they

1242 are put on the BOTTLE containers. From left to right, mixtures of SAP with Calcium Chlo-

1243 ride (container 2), Ferric Sulphate (container 5), Magnesium Perchlorate (container 3) and

1244 Sodium Perchlorate (container 4). These containers were left outdoors exposed to the ambient
1245 temperature and humidity conditions to determine their water absorption capability.

1246



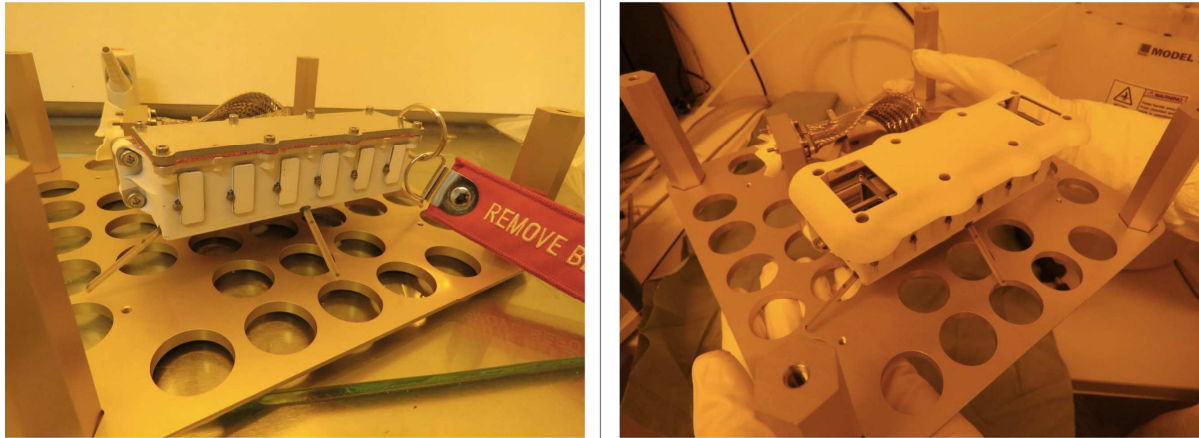
1247

1248 **Figure 33:** After three days of Earth outdoor exposure, the products have absorbed water
1249 from atmosphere and are in a jelly state. For this specific day, the environmental conditions
1250 were such that supported deliquescence of Calcium Chloride and this, mixed with SAP,
1251 formed a hydrogel. The other salts appear in hydrated state.

1252

1253 In order to avoid deliquescence of the BOTTLE products during its storage on Earth, and to
1254 allow for all the stages of assembly, including the upside-down configuration, a solid alumin-
1255 ium lid with a specially designed gasket made of HT870 silicone has been used in HABIT for
1256 all the phases of storage, integration, testing and transportation. The solid lid fixture designed
1257 will be replaced with the actual roof of HABIT before flight. Figure 34 (left) shows the solid
1258 aluminium lid with the specially designed gasket and the Figure 34 (right) shows the actual
1259 roof of HABIT which is the flight ready configuration.

1260



1261
1262 **Figure 34:** (left) HABIT BOTTLE fitted with the solid aluminium lid and HT870 gasket dur-
1263 ing the storage phase and (right) HABIT BOTTLE fitted with the roof, which is the flight
1264 ready configuration of the HABIT BOTTLE.

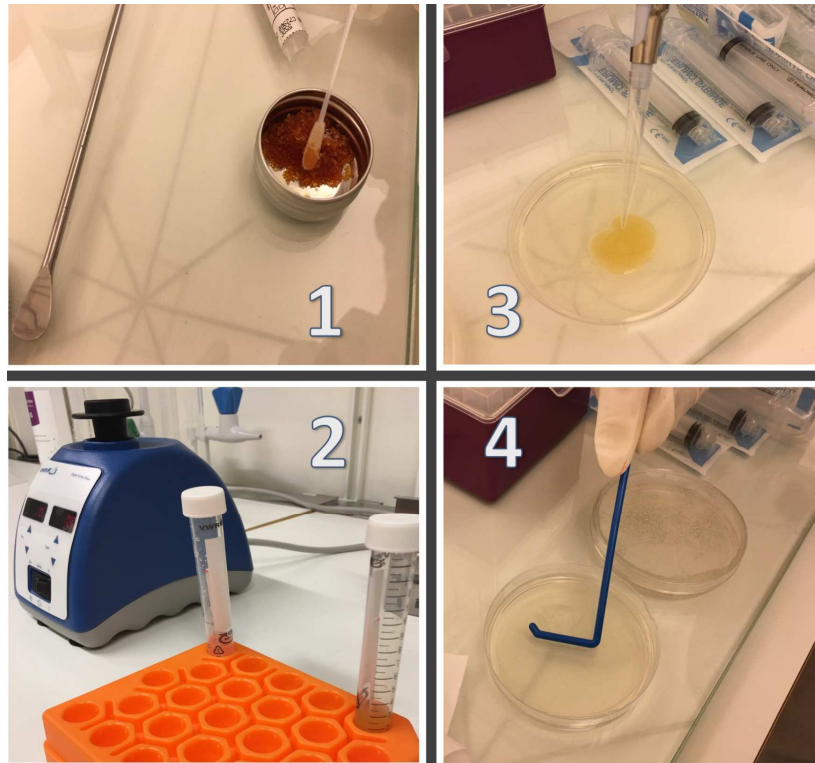
1265

1266 **Planetary Protection Tests**

1267 The ExoMars 2022 mission, landing at Oxia Planum, is designated under Category IVb, fol-
1268 lowing the indications of the European Space Agency (ESA) Planetary Protection Office
1269 (PPO), as per the Committee on Space Research (COSPAR) regulation. Category IVb. de-
1270 fines missions with lander systems designed to investigate the extant of martian life without
1271 contacting a martian special region. The protocol demands a stringent surface bioburden level
1272 of ≤ 30 spores. Though the BOTTLE contents are isolated from the exterior environment
1273 through the HEPA filter, which prohibits any flow of micro-organism, and any potential bio-
1274 logical contamination would be thus considered as encapsulated bioburden, we have applied
1275 bioburden reduction procedures to the products of BOTTLE. All the four salt+SAP mixtures
1276 have been subjected to Dry Heat Microbial Reduction (DHMR) at 125 °C for 10 hours, to re-
1277 duce the magnitude of any potential bioburden by 4 to 6 orders magnitude, according to the
1278 ECSS-Q-ST-70-57C protocol. To verify the clean level, the salt+SAP mixtures have then
1279 been subjected to the standard swab assay procedure as per the ECSS-Q-ST-70-55C protocol.

1280 Figure 35 shows an overview of the swab assay procedure of the Ferric sulphate salt+SAP
1281 mixture for determining the microbial colony count after subjecting the salt+SAP mixture to
1282 10 hours of DHMR at 125 °C. The procedure has been performed in a custom made double
1283 walled cleanroom tent with a HEPA filtration system (Mathanlal et. al, 2020). Controls were
1284 also taken into account during the process to determine the cleanliness of the procedure. As a
1285 main certification process, a set of swabs were sent to the Department of Aerospace Medicine,
1286 Deutsches Zentrum für Luft- und Raumfahrt (DLR), Germany to assay and determine the col-
1287 ony counts. The products of the swabs were then incubated at 32 °C for three days and the
1288 colony count was analysed at 24-hour, 48-hour and 72-hour mark. The test results from DLR
1289 further confirmed the cleanliness of the salts with zero colony counts. Additionally, the
1290 HABIT hardware was also sterilized and tested, with probes showing also with compliance
1291 with the PP requirements.

1292



1293

1294 **Figure 35:** 1). Swabbing the surface of the Ferric Sulphate salt+SAP mixture with nylon
1295 tipped FLOQSwab 2). The swabs are then suspended in 2.5ml of sterile Phosphate Buffer Sa-
1296 line (PBS) 3). After the vortex and thermal shock procedure, 0.5ml of the sample pipetted on
1297 to R2A petri dish. 4). Spreading the sample with sterile spreader prior to incubation.

1298

1299 Since the products in BOTTLE will probably form brines at relatively mild temperatures, the
1300 planetary protection measurements are especially important here. In parallel, as double blind
1301 assay, for the products of BOTTLE we performed studies with an equivalent set of swabs and
1302 incubated them after spreading them over R2A petri dishes. It could be found that there was
1303 no colony observed in the salt+SAP mixture after 72 hours. The BOTTLE contents have thus
1304 met the Category IVb. planetary protection requirements and the organics (SAP) are con-
1305 tained in a HEPA-encapsulated environment.

1306

1307

1308 **8. HABIT operations and data archiving**

1309

1310 **8.1 HABIT operation modes**

1311 HABIT has two sampling modes: i) the nominal mode, that measures every second within
1312 the first 5 minutes of each hour; ii) the extended mode, during which the instrument will
1313 measure data for every second during a maximum time-lapse of 3 hours. After that, it will
1314 switch to the nominal mode again.

1315 HABIT expects to receive a command every day from the Surface Platform BIP (acronym for
1316 the Onboard Information and Memory unit, in Russian) with an updated schedule, specifying
1317 for instance when the extended acquisition starts or if the heaters of the CU should be
1318 switched on, for how long and with how much intensity. If no command is sent, it will apply
1319 by default the last schedule. Additionally, during the cruise phase to Mars, HABIT will be
1320 turned on at least twice for autonomous functional tests. The duration of these tests can be
1321 limited to a few minutes without the need to go on regular or extended mode (just three pack-
1322 ets are needed for health-check tests: the initial data packet, a self-check packet and a sensor
1323 data packet, which can be received within one minute). When HABIT is powered up, it ini-
1324 tializes its own time by receiving it from the surface platform's (SP) interface and memory
1325 unit (BIP), which transmits to HABIT the schedule for operations for the next sol or sols.
1326 HABIT has a safety procedure to identify when the received command contains an error, for
1327 instance if a parameter is set out of range it is set by default to its maximum allowed value.
1328 The transmitted data files include self-check flags (binary variables that point for instance to
1329 out of range values). Self-check routines of the brines and temperature electronics take place
1330 at the beginning of every session using its own resistors. The current of each cell is also veri-
1331 fied with self-check resistor to detect if the calibration is drifting or if some electronic func-
1332 tion is not working properly.

1333

1334 8.2 HABIT scientific data outputs and units, for every second of acquisition

1335 HABIT will provide the following measurement: air temperature 1 (K), air tempera-
1336 ture 2 (K), air temperature 3 (K), wind 1 (m/s), wind 2 (m/s), wind 3 (m/s), ground tempera-
1337 ture (K), UVA (W/m^2), UVB (W/m^2), UVC (W/m^2), UVD (W/m^2), UVE (W/m^2), UVABC
1338 (W/m^2), Electrical Conductivity (S/m) level 1 and 2 of air-cells 1 and 6, Electrical Conduc-
1339 tivity (S/m) level 1, 2 and 3 of cells 2 to 5 (deliquescent salts, absorbing atmospheric water).

1340

1341 8.3 HABIT data budget

1342 HABIT stores all data packets (science data, engineering and scheduled operations) within its
1343 non-volatile (NAND) 512MB flash memory. The expected amount of data acquired per sol
1344 (assuming an extended acquisition of 3 hours) is 764 packets, each packet is 2032 bytes size
1345 resulting in a total of 1.5 MB/sol. The data are saved in HABIT even when the power is
1346 turned off. BIP, once per sol, will request all stored data (science and engineering data) to
1347 HABIT and send it to the Surface Platform's onboard computer in binary format, from here
1348 the data are sent to Earth through the satellite relay. Upon successful transmission to the sur-
1349 face platform's BIP, HABIT's memory pointer will go to the next register, keeping the trans-
1350 mitted data in the previous one. This register is easily accessible again in case it is needed to
1351 send the data yet to BIP. HABIT will only delete data from its memory once it is full, starting
1352 from the oldest to the most recent one. HABIT will be able to store up to 330 sols in its inter-
1353 nal memory; hence, in case there is any miscommunication between BIP and HABIT, it will
1354 be able to work autonomously for many sols.

1355

1356 8.3 HABIT power budget

1357 The power consumption of HABIT can be calculated with the following equation, the first
 1358 term is the power required for the operation of the EU, 2.1 W, the second term is the power
 1359 used when the heater of the CU is activated. The resistance of the CU heater is $R_{CUheater} = 131$
 1360 Ohm. The CU heater operates with pulse width modulation (PWM) and its amplitude can be
 1361 modulated by modifying the intensity with the duty cycle (D) factor, that can be commanded
 1362 and varies between 0, when no heating is activated on the CU, and 1, when a dedicated dehy-
 1363 dration cycle is programmed for the BOTTLE containers. The last term is the power required
 1364 to heat the EU when the temperature drops below -33°C . The EU heater can be switched on
 1365 by the autonomous start-up logic or by the EU logic. The EU logic will enable the EU heater
 1366 when the EU temperature is below $-33^{\circ}\text{C} \pm 1^{\circ}\text{C}$ and disable if the temperature rises above -
 1367 $30^{\circ}\text{C} \pm 1^{\circ}\text{C}$.

1368

where $D = 0$ to 1.00

$R_{CUheater} = 131 \text{ Ohm}$

$R_{EUheater} = 249 \text{ Ohm}$

$U_{in} = 28 \text{ V}$

$P_{EU} = 2.1 \text{ W}$

$$P_{Avg} = P_{EU} + \frac{(U_{in} * D)^2}{R_{CUheater}} + \frac{U_{in}^2}{R_{EUheater}}$$

1369 Thus, the power consumption of HABIT is 2.1 W, but when the heaters are activated to their
 1370 maximal power it can reach 13.1 W.

1371

1372 **8.4 HABIT data structure**

1373 The data obtained with HABIT will be archived at the IKI/Roscosmos Archive and will be
 1374 mirrored at the Planetary Science Archive (PSA) (<https://www.cosmos.esa.int/web/psa/psa->

1375 introduction) of the European Space Agency (ESA) using the Planetary Data System (PDS)
1376 version 4 (PDS4) format standard created by NASA. This archiving protocol is used in all
1377 current NASA and ESA space missions along with other international agencies that are mem-
1378 bers of the International Planetary Data Alliance (IPDA) (<https://planetarydata.org/>). HABIT
1379 produces four levels of data:

1380

- 1381 - Raw: Received directly from IKI/Roscosmos in binary format, there is no telemetry on
1382 them, just the surface platform timestamp and HABIT data. The size of this file is ap-
1383 proximately 350 KB.
- 1384 - Partially processed: During operations, right after receiving the raw data, HABIT bi-
1385 nary data is converted into a column-based comma-separated ASCII data file without
1386 any calibration. The size of this file is approximately 10 MB containing 39 columns.
- 1387 - Calibrated: Using the partially processed data and applying the instrument sensors cal-
1388 ibration functions, we create the calibrated version of the data. The size of this file is
1389 approximately 20 MB containing 50 columns.
- 1390 - Derived: To create the final scientific version of the data we must combine ancillary
1391 data that HABIT does not measure by itself and the calibrated data that we calculated
1392 previously. This level needs external information which is stored in the ancillary file,
1393 such as solar zenith angle (SZA), or ambient pressure (P). The size of this file is 24
1394 MB containing 65 columns. The HABIT processed data that will be stored at PSA will
1395 also include quality flags check for UV dust, UV shadow, UV angle blindness, noise
1396 for electronic conductivity and ATS errors.

1397

1398 The downlinked data will be processed daily by the Ground Software HABIT Team, using the
1399 calibration functions of the sensors and electronics of HABIT. This transformation shall be

1400 implemented on a daily basis, during operations. Then following the ExoMars data policy, the
 1401 files shall be delivered as a bundle, including the XML product labels, to be archived at the
 1402 PSA. Every HABIT bundle will include all the data, ancillary information and other files re-
 1403 lated to the instrument. In Figure 36 we show an example of a bundle made with the PDS4
 1404 standard.
 1405

```

| bundle_em20_sp_habit.xml
| README.txt
|
|---calibration
|   collection_calibration.csv
|   collection_calibration.xml
|   README.txt
|   README.xml
|
|---data_raw
|   collection_data_raw_inventory.csv
|   collection_data_raw_inventory.xml
|
|   ---SOL_00001_00020
|       collection_sol_00001_00020_raw_inventory.csv
|       collection_sol_00001_00020_raw_inventory.xml
|       SOL_00001.TAB
|       SOL_00001.xml
|
|---data_partially_processed
|   collection_data_partially_processed_inventory.csv
|   collection_data_partially_processed_inventory.xml
|
|   ---SOL_00001_00020
|       collection_sol_00001_00020_pp_inventory.csv
|       collection_sol_00001_00020_pp_inventory.xml
|       SOL_00001.TAB
|       SOL_00001.xml
|
|---data_calibrated
|   collection_data_calibrated_inventory.csv
|   collection_data_calibrated_inventory.xml
|
|   ---SOL_00001_00020
|       collection_sol_00001_00020_calibrated_inventory.csv
|       collection_sol_00001_00020_calibrated_inventory.xml
|       SOL_00001.TAB
|       SOL_00001.xml
|
|---data_derived
|   collection_data_derived_inventory.csv
|   collection_data_derived_inventory.xml
|
|   ---SOL_00001_00020
|       collection_sol_00001_00020_derived_inventory.csv
|       collection_sol_00001_00020_derived_inventory.xml
|       SOL_00001.TAB
|       SOL_00001.xml
|
|---miscellaneous_ancillary
|   collection_ancillary_inventory.csv
|   collection_ancillary_inventory.xml
|   ancillary_data.TAB
|   ancillary_data.xml
|
|---xml_schema
|   PDS4_PDS_1C00.sch
|   PDS4_PDS_1C00.xml
|   PDS4_PDS_1C00.xsd

```

1406

1407

Figure 36: Example of HABIT PDS4 Bundle

1408

1409

1410

1411 **9. Operation on Mars, simulations on a Mars environmental** 1412 **chamber**

1413

1414 To validate HABIT operations on Mars we have developed the SpaceQ chamber, see Figure
1415 37 (Vakkada-Ramachandran et al., 2019). This chamber is designed to operate instrumenta-
1416 tion in representative space conditions (vacuum, Mars atmosphere, Earth stratosphere, lunar
1417 environment, etc.) and to validate their performance by acquiring real-time data in a simulated
1418 environment. It is also designed to test and qualify the behaviour of certain components when
1419 exposed to thermal vacuum, outgassing, baking, low temperatures and dry heat microbial re-
1420 duction procedures.

1421

1422 One of the main functionalities of the SpaceQ chamber is to recreate locally certain conditions
1423 of the Mars surface for research purposes, in this case to operate HABIT as on Mars. The
1424 chamber can mimic the diurnal variation of temperature and the water cycle within a thin at-
1425 mosphere of CO₂ as on Mars. A unique feature of this chamber is that it includes connectors
1426 for USB and DB25 to read the data from the instrumentation while being tested inside.

1427

1428 The SpaceQ chamber is a versatile facility which can operate with a wide range of tempera-
1429 tures from 193 K to 423 K and pressure from ambient to $< 10^{-5}$ mbar. It is a cubical chamber
1430 made of stainless steel with internal volume of 27 l having multiple viewports, feedthroughs
1431 and flanges. In order to get the chamber down to vacuum we use rotary pump which pumps
1432 down the air to 10^{-3} mbar, if we need to achieve higher orders of vacuum, we then use a turbo
1433 molecular pump to reach 10^{-5} mbar. The pressure inside is monitored by a cold cathode/pirani
1434 combination gauge which is connected to the display unit and records the values in real time.

1435 There are two gas inlets one used to inject CO₂ gas and other to inject water using a stainless-
 1436 steel syringe connected via a ball valve to the Swagelok on the chamber. The instrument
 1437 placed inside the chamber can be communicated in real time with the USB and DB 25 pins
 1438 fitted on the chamber ports. The temperature of the chamber is maintained in two ways i)
 1439 cooling: the working table is fitted with a feedthrough to circulate liquid nitrogen in order
 1440 cool it down to 193 K ii) heating: the chamber wall from outside is fitted with the heating
 1441 jackets which raise the temperatures up to 423 K. A Vaisala HMT 334 probe which is used
 1442 for vacuum applications to monitor RH/T of the air inside the chamber. The chamber also has
 1443 a provision to do visible/near infra-red VNIR spectroscopy as the chamber walls are fitted
 1444 with the feedthrough to connect the probes of the spectrometer and obtain the spectra of the
 1445 samples inside. The SpaceQ chamber has been used to simulate the launch phase, cruise
 1446 phase, and operation of HABIT on the surface of Mars and will be further used to investigate
 1447 the different responses of the BOTTLE unit of HABIT depending on the angle of inclination,
 1448 thermal cycles, etc.

1449

1450

1451

Table 7: Specifications of the SpaceQ chamber

Parameter	Characteristics
Internal volume	27 litres
Operating temperature	-80°C to + 150 °C
Operating pressure	ambient to < 10 ⁻⁵ mbar
UV lamp	115-400 nm
Data output connectors	USB and DB 25
Vaisala RH/T sensor	0-100% and -70°C to +180°C
Gas/fluid inlet	CO ₂ and water

Spectrometers

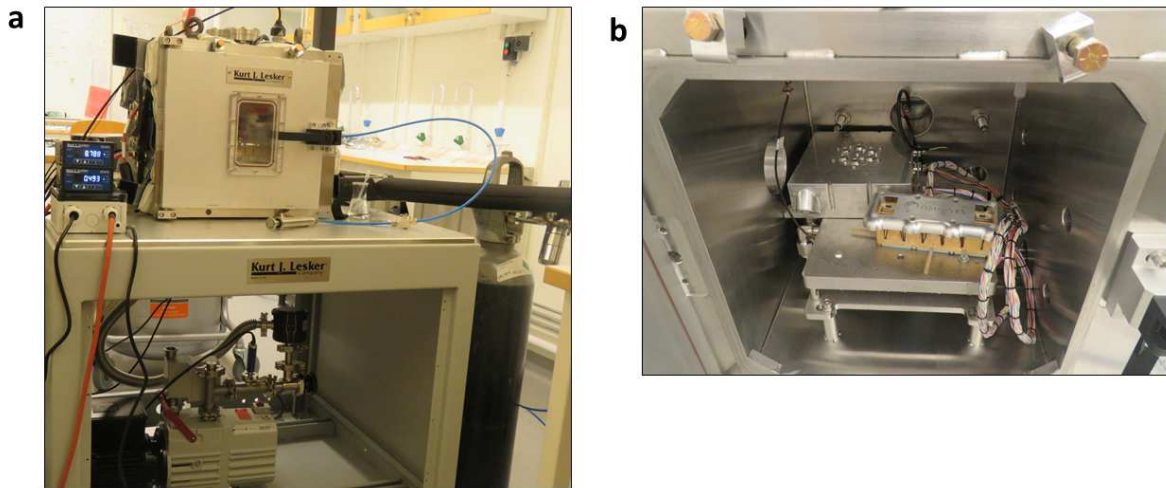
VNIR 200-1100 nm

and 1000 nm-2500 nm

1452

1453

1454



1455

1456

Figure 37: SpaceQ chamber facility: (a) SpaceQ chamber with the rotary pump, syringe used to inject water, CO₂ cylinder and the pressure display (b) HABIT EM placed on the working table and integrated with the DB 25 pin for operation. This table is refrigerated with liquid N₂.

1461

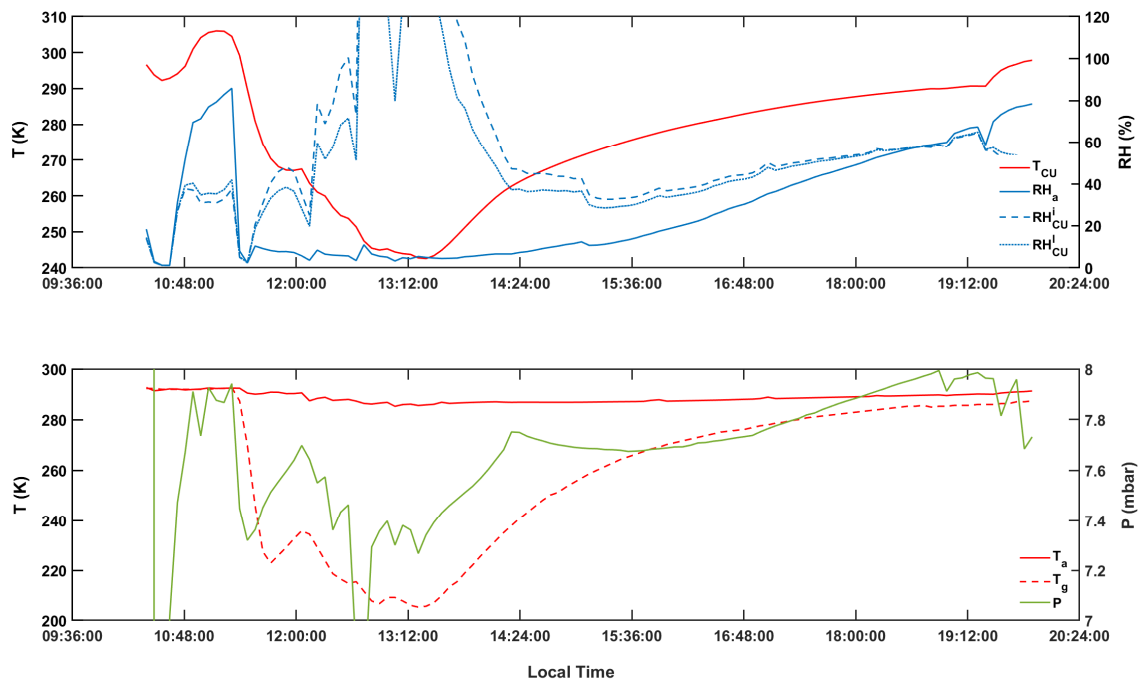
1462

1463 9.1 Mars Sol simulation

1464 We simulated a plausible martian day-night cycle with the HABIT Engineering Model (EM)
1465 inside the SpaceQ chamber while controlling the temperature, relative humidity and carbon-
1466 dioxide atmosphere identical to the ExoMars 2022 landing site, Oxia Planum. We evacuated
1467 out the ambient air from the chamber using a rotary vacuum pump to attain about pressures in
1468 orders of 10^{-3} mbar followed by a gentle increase in pressure with carbon-dioxide atmosphere

1469 and maintained it between 6 mbar. We then injected water using a vacuum grade syringe in
1470 increments of 0.5 ml at a time, to increase the air relative humidity gradually over 80%. It re-
1471 quired up to 20 water injection cycles to achieve the desired initial air relative humidity of 86
1472 %. The working table where HABIT was accommodated was then cooled by injecting LN₂
1473 until BOTTLE temperature (denoted as TCU in Figure 38) reached 243 K (203 K in working
1474 table). As the working table cools down, frost starts to form on the table and the relative hu-
1475 midity of the air is reduced. Based on REMS observations on board the Curiosity rover, this is
1476 just what is expected to happen on Mars, where the ground will generally be much cooler than
1477 the surface platform and the air above. The cooling was then cut off for the chamber to warm
1478 up again to ambient temperature of about 293 K. As the table temperature increases water is
1479 released to the air, which produces an increase in the RH%. Once the table temperature reach-
1480 es ambient temperature the experiment is stopped. The progression of the martian simulated
1481 environmental conditions during the experiment is shown in Figure 38. We also calculated the
1482 relative humidity for water with respect to ice (RH_{CU}^i) and liquid (RH_{CU}^l) phases for the tem-
1483 peratures of the container unit to estimate the relative humidity to which the salts are exposed.
1484
1485 The BOTTLE EM was initially filled with salt-SAP mixtures in the same configuration and
1486 same amounts as in the flight model, with the only modification of using a salt also in Cell-6
1487 instead of leaving it empty. In the process as the table temperature crosses the triple point of
1488 water the frost turns into liquid droplets which then evaporate, and the relative humidity of the
1489 air starts to raise. The increment in EC of the salt containers indicated that some salts were
1490 absorbing water through hydration or deliquescence.

1491



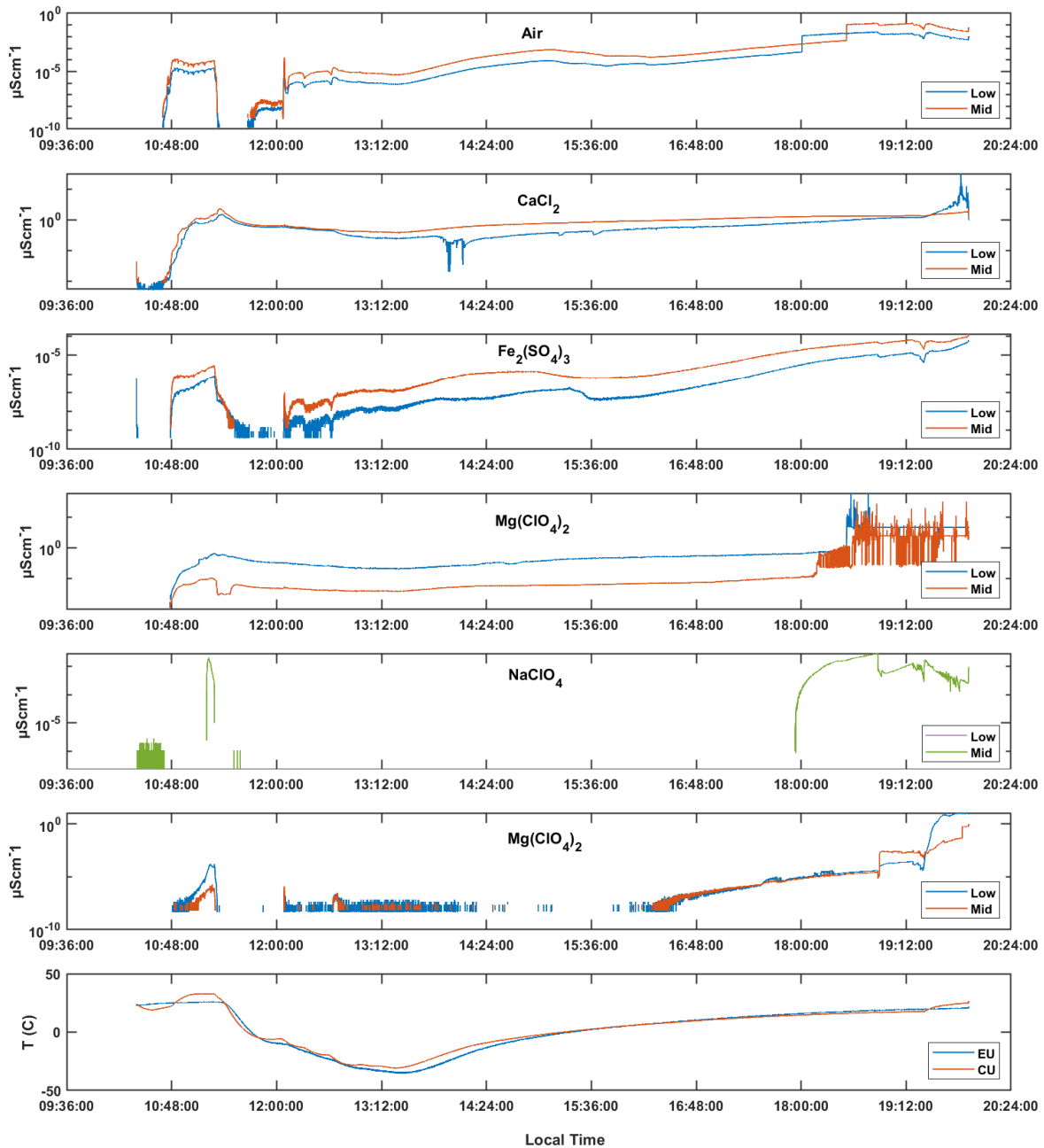
1492

1493 **Figure 38:** Simulated environment inside the SpaceQ chamber. Different variables recorded
 1494 such as temperature of both air (T_a) and BOTTLE temperature (T_{CU}), pressure (P), relative
 1495 humidity of air (RH_a) and relative humidity at which BOTTLE salts are exposed, calculated
 1496 with respect to ice (RH_{CU}^i) and liquid (RH_{CU}^l).

1497

1498 Figure 39 shows the electrical conductivity measurements of the salt-SAP mixtures during the
 1499 martian day-night simulation. We could observe two distinguishable EC characteristics corre-
 1500 sponding to the water capture, one at the beginning and one towards the end of the simulation.
 1501 When the relative humidity of the air was increased to 86% as the water was injected, we see
 1502 a spike in the electrical conductivity values. This behaviour lasted for a short time until the
 1503 moment when the relative humidity reduced due to cooling and the table acted as a sink,
 1504 where frost was formed. Cell-1 which was empty also showed this behaviour probing that the
 1505 EC measurements are sensitive to the air relative humidity. As the temperature of the chamber
 1506 increased, water was released from the cold table, and the air relative humidity increased too.
 1507 This allowed favourable conditions for the salt-SAP mixtures to capture water which resulted

1508 in a gradual increase in the EC values. Both of these EC increases were complemented with a
 1509 visible increase in the BOTTLE (CU) temperature –which is not observed in the EU- due to
 1510 the exothermic nature of the deliquescence process.



1511
 1512 **Figure 39:** Electrical conductivity measurement (in log scale) of the salts in BOTTLE Engi-
 1513 neering Model during the Mars Sol simulation in the SpaceQ chamber. For this experiment
 1514 the salts were distributed as, Cell-1: Empty, Cell-2: CaCl₂, Cell-3: Fe₂(SO₄)₃, Cell-4: Mg
 1515 (ClO₄)₂, Cell-5: NaClO₄, Cell-6: Mg (ClO₄)₂. This experiment simulates HABIT for the study

1516 during its launch and cruise phase with a more representative diurnal variation on Oxia
1517 Planum, Mars.

1518

1519 10. Summary

1520

1521 This manuscript describes the instrument HABIT, its operation, and calibration and retrieval
1522 procedures that shall be applied to convert the raw data into scientific meaningful variables.
1523 HABIT operation has been analysed under laboratory conditions, and dedicated chambers, in
1524 order to simulate the different environments where it will be operating. HABIT has been inte-
1525 grated in the ExoMars Surface Platform *Kazachok*, and once in operations on Mars the data
1526 will be archived and shared publicly through the Planetary Science Archive (ESA).

1527

1528 HABIT is composed by 2 modules: EnvPack and BOTTLE. The EnvPack module has herit-
1529 age from previous missions operating on the surface of Mars, and the sensors environmental
1530 observations will be directly comparable with those provided by other landed missions on
1531 Mars. In HABIT, the implementation of the ATS, GTS and UVS has been adapted to the re-
1532 quirements of the Surface Platform but their observations can be directly compared with those
1533 acquired by REMS, on board the Curiosity rover. HABIT can provide information of the lo-
1534 cal temperature with $\pm 0.2^{\circ}\text{C}$ accuracy, local winds with ± 0.3 m/s, surface brightness tem-
1535 perature with $\pm 0.8^{\circ}\text{C}$, incident UV irradiance with 10% error of its absolute value in the
1536 UVA, UVB, UVC ranges, as well as in the total UVABC range, and two additionally wave-
1537 bands, dedicated to ozone absorption. The BOTTLE experiment is unique in its kind. The po-
1538 tential transient or permanent existence of brines on Mars, has been controversial and this ex-
1539 periment will provide ground-truth to this hypothesis, by monitoring their response in a con-
1540 trolled environment. Our experiments in a Mars simulation chamber demonstrate that the salts

1541 can absorb water within a few hours of exposure to increased ambient humidity, and that this
1542 water can be indirectly monitored by the changes of electric conductivity. Our tests illustrate
1543 how BOTTLE can monitor the hydration state of a set of four salts, which have been found on
1544 Mars, calcium chloride, ferric sulphate, magnesium perchlorate and sodium perchlorate by
1545 measuring their electric conductivity (EC). The EC of the air and the dry salts within the
1546 Flight Model under Earth ambient, clean room conditions has been shown to be of the order
1547 of $0.1 \mu\text{Scm}^{-1}$ whereas for vacuum conditions it is of the order of $0.00364 \mu\text{Scm}^{-1}$ at ambient
1548 temperatures. We have simulated the operation of HABIT, within an environmental chamber,
1549 under martian conditions similar to those expected at the ExoMars landing site, in Oxia
1550 Planum. For dry CO_2 conditions, at low temperatures, the measured air EC can be as low as
1551 $10^{-8} \mu\text{Scm}^{-1}$, however it increases with RH%. For this specific experiment, after an increase
1552 from 0 to 60% RH, within a few hours, the EC of the air increased up to $10^{-1} \mu\text{Scm}^{-1}$, while
1553 magnesium perchlorate hydrated and reached values of $10 \mu\text{Scm}^{-1}$ and calcium chloride deli-
1554 quesced forming a liquid state with EC of $10^2 \mu\text{Scm}^{-1}$. The EC of the other two salts was
1555 comparable to the one of the air, indicating that the EC of the air within the voids between salt
1556 grains is the dominant component and none, or a negligible amount of hydration has taken
1557 place. This comparison between the conductivity of the salts and that of the air, shall always
1558 be done on Mars, since all the experiments are exposed to the same temperature and ambient
1559 humidity. The products of BOTTLE have been treated with Dry Heat Microbial Reduction
1560 sterilization protocols and have passed the ESA planetary protection controls showing no via-
1561 ble cells. Furthermore, our studies demonstrate that the swelling ratio of thermally heated and
1562 depressurized SAP was equal to the one of unperturbed commercial SAP suggesting that SAP
1563 can be used for water storage in space applications, in particular at martian pressure condi-
1564 tions and exposed to diurnal and seasonal thermal changes (Mathanlal et. al, 2020).

1565

1566 In addition to these tests under laboratory conditions, to demonstrate the operation under non-
1567 controlled planetary environments, two engineering prototypes of HABIT have been tested at
1568 two international field-campaigns, one in the Himalaya region, a cold and arid environment,
1569 (Pandey et al., 2019) and another one in a subsurface environment, with very constant dry
1570 conditions (Cockell et al., 2018).

1571

1572 The temperature range of HABIT operation, about -30°C , is coincident with the reported lim-
1573 iting temperatures for cell metabolism and reproduction activity, and thus detecting brine
1574 formation at these temperatures will have strong implications for the habitability of present
1575 day Mars and also on the future planetary protection considerations that should be applied to
1576 sites where brines may be formed.

1577

1578 The observations provided by HABIT may have synergies with other instruments of the rover
1579 and SP (<http://exploration.esa.int/mars/56933-exomars-2020-surface-platform/>). The meteoro-
1580 logical station of the SP, METEO-SP, includes Pressure (P) and Relative Humidity (RH) sen-
1581 sors whose measurement shall be complementary to those of HABIT. Additionally, the at-
1582 mospheric composition observations of FAST-SP, a spectrometer, and MGAP-SP, a GCMS,
1583 may be complemented with the UV measurements of HABIT, as on Mars the atmospheric
1584 concentration of certain species changes through seasons due to UV-induced photochemistry.
1585 Regarding the interaction of the subsurface and near surface with the atmosphere, any season-
1586 al or diurnal variation of the observations of ADRON (which monitors the subsurface water)
1587 should be compared with the observations of BOTTLE of the electric conductivity changes of
1588 the air and the state of hydrate, frozen brine or liquid brine of the salts in the containers. The
1589 near surface winds and temperatures may be compared with the winds of the anemometer and
1590 the air temperature of the METEO-SP payload, which is mounted at a higher level on the

1591 mast. The near surface heat-flux depends on the vertical thermal gradients and winds and
1592 monitoring this at two points near the surface is critical to provide this information at the
1593 boundary layer. The ground temperatures may be compared with the radio thermometer PAT-
1594 M for soil temperatures, and the diurnal profile should be used to model the ground thermal
1595 inertia which in turn may be correlated with the presence of subsurface ices, that may be po-
1596 tentially detected by ADRON or the radar of WISDOM (in the rover). The atmospheric dust
1597 cycle may be observed in the UV by HABIT, and compared with the observations of the
1598 PanCam instrument, the camera of the rover, and the studies of dust cycle may be comple-
1599 mented with the observations of the Dust Suite instrument of the platform which determines
1600 size, impact and the atmospheric charge. Furthermore, the studies of hydration and surface-
1601 atmosphere water interchange may have implications on the hydration state of surface and
1602 subsurface minerals. This may be monitored in coordinated campaigns with ISEM and
1603 Mi_MASS on the rover, which monitor the IR spectra of minerals. The EC measurements of
1604 the air, within the protected environment of the BOTTLE unit, can be compared with those
1605 provided by the Dust Suite of the open air on the Surface Platform. The near surface dust ob-
1606 servations of MicroMED about the dust cycle and the typical particle size can be compared
1607 with the studies of the total column of the atmosphere provided by HABIT combining the
1608 measured spectral dependence of the absorbed UV radiation with radiative transfer models.
1609 The relative humidity changes observed by the METEO package can be compared with the
1610 long-term EC variations of the air. The UV 200-400 nm observations shall be compared with
1611 those provided by the 250-400 nm measurements of SIS on the Surface Platform. Addition-
1612 ly, the comparison of HABIT observations with those provided by MEDA on the NASA Per-
1613 severance Rover 2020, and REMS on the NASA Curiosity Rover, will provide three simulta-
1614 neous and similar data sets of environmental observations at different latitudes and longitudes

1615 on Mars that may be used as ground truth validation for orbiter observations and to compare
1616 with global and meso-scalar atmospheric modelling and radiative transfer models.

1617

1618

1619

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1641

1642 **Author Disclosure Statement**

1643

1644 No competing financial interests exist.

1645

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HiGhlights

- The HABIT instrument will characterise the present-day habitability at the ExoMars 2020 landing site
- HABIT will investigate the existence of liquid water on Mars in the form of brines, for how long, and the water-regolith cycle.
- HABIT will investigate the conductivity of atmospheric dust on Mars
- The module BOTTLE will be the first European In-situ Resource Utilization instrument on Mars. It will produce liquid water for future Mars exploration

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Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

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