

1     **A rapid monitoring method for inorganic arsenic in rice flour using**  
2     **reversed phase-high performance liquid chromatography-inductively**  
3     **coupled plasma mass spectrometry**  
4

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12

13    **Abstract**

14    A new rapid monitoring method by means of high performance liquid  
15    chromatography-inductively coupled plasma mass spectrometry (HPLC-ICP-MS) following the  
16    heat-assisted extraction was developed for measurement of total inorganic arsenic species in rice  
17    flour. As(III) and As(V) eluted at the same retention time and completely separated from  
18    organoarsenic species by an isocratic elution program on a reversed phase column. Therefore,  
19    neither ambiguous oxidation of arsenite to arsenate nor the integration of two peaks were  
20    necessary to determine directly the target analyte inorganic arsenic. Rapid injection allowed  
21    measuring 3 replicates within 6 min and this combined with a quantitative extraction of all  
22    arsenic species from rice flour by a 15 minute HNO<sub>3</sub>-H<sub>2</sub>O<sub>2</sub> extraction makes this the fastest  
23    laboratory based method for inorganic arsenic in rice flour.  
24

25    *Keywords: Arsenic speciation; HPLC-ICP-MS; Inorganic arsenic determination; Rice flour;*  
26    *Risk assessment; Extraction.*  
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31    **Highlights**

- 32     • A new HPLC-ICP-MS method was developed for the rapid monitoring of inorganic As  
33     • Inorganic As can be analyzed directly since As(III) and As(V) co-eluted  
34     • Inorganic As can be separated from all organoarsenic compounds in rice  
35     • Three replicates can be analyzed within 6 minutes  
36     • Complete extraction of As species from rice flour was achieved within 15 minutes  
37

## 38 1. Introduction

39 Knowledge of chemical species of elements present in biological and environmental  
40 samples is important for an understanding of the toxicity, metabolism and transport properties of  
41 the elements.[1-2] Therefore, determination of specific chemical forms of elements  
42 (speciation analysis) has become an increasingly active-research field in recent years.[3]  
43 Arsenic (As) is found in various chemical species in environmental and food samples. In  
44 general, naturally occurring organoarsenic compounds are considered to be of non-toxicity, but  
45 inorganic arsenic in the form of arsenite (As(III)) or arsenate (As(V)) is a class I carcinogen.

46 The Codex Alimentarius Commission, which provides the guideline for allowable  
47 concentrations of toxic elements and compounds in foodstuffs, approved a tolerance level of  
48 inorganic arsenic as  $0.2 \text{ mg kg}^{-1}$  in polished rice in 2014. A tolerance level for inorganic  
49 arsenic in unpolished rice is still under discussion but rice for baby and toddler food has an even  
50 lower maximum limit for inorganic arsenic ( $0.1 \text{ mg kg}^{-1}$ ).[4-6] Therefore, it is now apparent  
51 that a rapid determination of total inorganic arsenic species present in rice flour is necessary for  
52 monitoring test and/or surveillance.[6]

53 Prior to measurements, the arsenic species in rice flour samples are generally extracted by  
54 the techniques employing heated  $\text{HNO}_3$  solutions.[7-10] Although most analytical procedures  
55 for arsenic in any foodstuffs are composed with similar operating steps, there are some specific  
56 conditions for the determination of inorganic arsenic in rice flour samples. The Codex  
57 Alimentarius Commission recommends the analytical methods of the arsenic species in rice  
58 flour, involving extraction with diluted acids (*e.g.*  $0.15 \text{ M HNO}_3$  or 1 to 2 (v/v)%  $\text{HNO}_3$ ,  
59 Extraction:  $100 \text{ }^\circ\text{C}$  for 2 h) and subsequent determination of As by HPLC-ICP-MS using a  
60 reversed phase (ODS) column.[11] On the other hand, the US Food and Drug Administration  
61 (FDA) recommends  $0.28 \text{ M HNO}_3$  for arsenic extraction followed by IC-ICP-MS using an  
62 anion-exchange column (IC) for the determination.[12]

63 An ODS column is well able to separate many arsenic species; it can separate the two  
64 inorganic arsenic species, six methylated arsenic species and four arsenosugars within 10 to 15  
65 minutes, however the peaks of As(III) and MMA tend to be very close each other. In a similar  
66 way, the peaks of DMA and arsenobetaine (AsB) and those of trimethylarsine oxide (TMAO)  
67 and tetramethylarsonium ion (TeMA) almost overlap one another.[13-15] When a large  
68 number of measurements are carried out by using one column, the column packing is likely to  
69 deteriorate and these pairs of peaks would not be separated eventually. In addition, a complex  
70 component of eluent containing ion-pair reagents and organic contents can cause a problem at  
71 the ICP-MS interface with long-term analysis.

72 Each of these techniques has their own advantages and disadvantages. Anionic arsenic  
73 species can be separated by an IC column with a simple component eluent and the separation

74 property depends upon the component and pH of the eluent.[16-22] An IC column is useful to  
75 separate arsenite [As(III)], arsenate [As(V)], monomethylarsonic acid (MMA), dimethylarsinic  
76 acid (DMA), and some arsenosugars respectively, but it cannot separate some methylated  
77 arsenic compounds under a single set of operating conditions.

78 The main arsenic species in agricultural samples, in particular rice flour, are As(III),  
79 As(V), MMA and DMA.[23-24] TeMA however was found in Chinese rice as well.[25]  
80 Hence, only arsenic in the forms of As(III), As(V), MMA, DMA and TeMA are of interest when  
81 rice grains are going to be analyzed.

82 The Codex Alimentarius Commission regulation refers to the inorganic arsenic, i.e. the  
83 sum of As(III) and As(V). Nevertheless, most monitoring test methods for inorganic arsenic in  
84 rice flour separate As(III) and As(V) and determine them individually which makes the  
85 integration of two peaks necessary for the quantification of the target analyte inorganic arsenic.  
86 Another approach is to oxidize all As(III) to As(V) and subsequently measure all inorganic  
87 arsenic as As(V) using an anion column. A peak appearing in the void where As(III) would  
88 elute means that either the cationic species TeMA is in the rice grain extract or the oxidation of  
89 As(III) was not complete. Since As(III) and TeMA elute at the same time[25], TeMA should  
90 be separated from As(III) and As(V) to avoid ambiguity and so far no method has shown to be  
91 capable for showing both species separately.

92 One of the most important purposes of food analyses is to ensure the food safety based on  
93 the appropriate regulations with rapidity and efficiency. Therefore, a rapid screening method  
94 with high sample throughput is required. In recent year, a screening and a rapid measurement  
95 methods for inorganic arsenic in rice flour were reported; the former was a field test kit method  
96 based on the Gutzeit methodology and the latter a fast separation-detection method using  
97 IC-ICP-MS/MS.[26-27]

98 In this study a rapid monitoring method of inorganic arsenic in rice flour with  
99 HPLC-ICP-MS using a reversed phase column with a simple component eluent was developed.  
100 Chromatographic conditions, in particular eluent component and pH were optimized to  
101 determine As(III) and As(V) at the same time, and completely separate them from other  
102 organoarsenic compounds such as DMA, MMA and TeMA.

103

## 104 **2. Experimental**

### 105 *2.1 Instrumentations*

106 An ICP-MS (7500c, Agilent, Tokyo, Japan) equipped with a Micromist nebulizer (100  $\mu$ L  
107 type) and a Scott spray chamber (2 °C) was used. Typical operating parameters for the  
108 ICP-MS were as follows: incident rf power was 1600 W, outer Ar gas flow rate 15 L min<sup>-1</sup>,  
109 intermediate Ar gas flow rate 0.9 L min<sup>-1</sup>, carrier Ar gas flow rate 0.8 L min<sup>-1</sup> and make-up Ar

110 gas flow rate  $0.4 \text{ mL min}^{-1}$ . The ICP-MS was usually operated using He as the collision cell  
111 gas ( $3 \text{ mL min}^{-1}$ ) to reduce some polyatomic molecular interferences. An HPLC was used for  
112 separation of arsenic species. The exit of the HPLC column was directly connected to the  
113 nebulizer of the ICP-MS with PEEK tubing (HPLC-ICP-MS). The typical operation  
114 conditions and its performances are shown in Table S1.

115 Fifteen types of columns were investigated in the experiment (Tables S2 and S3). Five  
116 types of columns categorized in group I provided satisfactory performances for rapid  
117 monitoring of inorganic arsenic compounds, but ten other types of columns categorized in group  
118 II were unsuitable for the purpose. The columns of group I are as follows; A Shim-pack  
119 VP-C8  $C_8$  column (hereafter referred to as the  $C_8$ : particle size of the filler  $5 \mu\text{m}$ , ID  $4.6 \text{ mm} \times$   
120  $250 \text{ mm}$ , end-capped type, Shimadzu Co., Kyoto, Japan), a CAPCELL PAK  $C_{18}$  MG column  
121 (hereafter referred to as the  $C_{18}$ : particle size of the filler  $5 \mu\text{m}$ , ID  $4.6 \text{ mm} \times 250 \text{ mm}$ ,  
122 polymer-coated type, Shiseido Co., Ltd., Tokyo, Japan), a CAPCELL PAK  $C_{18}$  ACR (hereafter  
123 referred to as the  $C_{18}$  ACR: particle size of the filler  $5 \mu\text{m}$ , ID  $4.6 \text{ mm} \times 250 \text{ mm}$ ,  
124 polymer-coated type, Shiseido), a Sunrise  $C_{28}$  column (hereafter referred to as the  $C_{28}$ : particle  
125 size of the filler  $5 \mu\text{m}$ , ID  $4.6 \text{ mm} \times 250 \text{ mm}$ , TMS end-capped type, ChromaNik Technologies  
126 Inc., Osaka, Japan), and a DEVELOSIL  $C_{30}$ -UG-5  $C_{30}$  column (hereafter referred to as the  $C_{30}$ :  
127 particle size of the filler  $5 \mu\text{m}$ , ID  $4.6 \text{ mm} \times 250 \text{ mm}$ , end-capped type, Nomura Chemical Co.  
128 Ltd., Aichi, Japan) were used.

129 The columns of group II were also tested under multiple elution conditions of eight  
130 separation modes. Details are given in the Tables S2 and S3.

131 A heating block system (Digi PREP, SCP Science Inc., Quebec, Canada) was used for  
132 heat assisted extraction.

133

## 134 2.2 Calibration standards and reagents

135 The Japan Calibration Service System (JCSS) arsenic standard solution (*ca.*  $1000 \text{ mg L}^{-1}$ ,  
136 Kanto Chemical Co., Inc., Tokyo, Japan) was used as the source of the calibration standard  
137 solution. It is produced by dissolving  $\text{As}_2\text{O}_3$  powder into  $\text{HNO}_3$  solution and contains only  
138 As(III). It did not actually contain detectable levels of any other arsenic species including  
139 As(V) and was thus used as the As(III) source standard solution.

140 The certified reference materials of As(V) (NMIJ CRM 7912-a), the dimethylarsinic acid  
141 (DMA) (NMIJ CRM 7913-a) and the arsenobetaine (AsB) (NMIJ CRM 7901-a) supplied by the  
142 National Metrology Institute of Japan / National Institute of Advanced Industrial Science and  
143 Technology (NMIJ/AIST, Tsukuba, Japan) were used as source standard solutions.

144 Source standard solutions of the other organoarsenic species such as monomethylarsonic  
145 acid (MMA), trimethylarsine oxide (TMAO), tetramethylarsonium chloride (TeMA), and

146 arsenocholine bromide (AsC) were prepared from commercially available reagents  
147 (Tri-Chemical Laboratories Inc., Yamanashi, Japan), after their purity was evaluated (moisture,  
148 elemental analysis of C, H, O, Br and Cl, and arsenic impurities). Each compound was  
149 dissolved in water to prepare an in-house standard solution containing *ca.* 1000 mg As kg<sup>-1</sup>.

150 The acids and ammonia solutions used were of ultrapur<sup>®</sup> grade (Kanto), and the organic  
151 solvents used were of HPLC grade (Kanto). Ammonium dihydrogen phosphate and  
152 diammonium hydrogen phosphate were of ultrapure grade (Kanto). Sodium 1-butanefulfonate  
153 (Tokyo Chemical Industry Co., Ltd., Tokyo, Japan), malonic acid (Wako Pure Chemical  
154 Industries, Ltd., Osaka, Japan), and tetramethylammonium hydroxide (TMAH, Tama Chemicals  
155 Co., Ltd., Kanagawa, Japan) were obtained as indicated. Ultra-pure water was generated with  
156 a Milli Q-Labo filter (Nippon Millipore Ltd, Tokyo, Japan) and was used throughout.

157

### 158 *2.3 Extraction procedure for arsenic species*

159 Arsenic species in the rice flour samples were extracted by a heat-assisted technique using  
160 acid solvent. A portion of rice flour sample (*ca.* 0.5 g) was accurately weighed into a 10 mL  
161 glass tube and 2 g of extracting solvent were added. Four kinds of extracting solvents were  
162 used here respectively; 0.15 mol L<sup>-1</sup> HNO<sub>3</sub>, 0.28 mol L<sup>-1</sup> HNO<sub>3</sub>, 0.30 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub> and 0.20 mol  
163 L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>+0.10 mol L<sup>-1</sup> HNO<sub>3</sub>. The tube was capped and placed in a dry heating block system  
164 at 100 °C for a period from 15 min to 2 h. After cooling to room temperature, 8 g of water was  
165 added (total liquid phase: 10 g). The tube was centrifuged at 4000 rpm for 5 min, and then the  
166 liquid phase was then passed through a 0.45 μm syringe-type polyvinylidene difluoride (PVDF)  
167 membrane filter. The filtrate was analyzed by HPLC-ICP-MS.

168

### 169 *2.4 Certified Reference Materials (CRMs)*

170 NMIJ brown rice flour CRMs (NMIJ CRM 7532-a and NMIJ CRM 7533-a) and NIST  
171 rice flour SRM (SRM 1568b) were analyzed to valid the newly developed rapid monitoring  
172 method.

173

## 174 **3. Results and Discussion**

### 175 *3.1 Conditions for simultaneous determination of inorganic arsenic species*

176 Besides the columns categorized in group I, the columns in group II were also examined  
177 with various types of eluent and at various pHs shown in supplementary material 1. However,  
178 unsatisfactory results were obtained for simultaneous determination of inorganic arsenic species.  
179 Therefore, the following discussion is limited to the experiments carried out using the columns  
180 in group I.

181 For chromatographic analysis of water-soluble arsenic species, pH of the eluent is one of

182 the most effective factors to control the separation of the compounds. The separation of  
183 As(III), As(V), MMA, DMA and TeMA using the columns of group I were investigated using  
184 the eluents containing 0.001 to 0.01 mol L<sup>-1</sup> HNO<sub>3</sub> / 0.05 % methanol. With increasing HNO<sub>3</sub>  
185 concentration (that is, lower pH), As(III), MMA, DMA and TeMA were eluted faster. The  
186 retention times of each species with the eluent of 0.01 mol L<sup>-1</sup> HNO<sub>3</sub> / 0.05 % methanol are  
187 shown in **Figure 1**, where the chromatography was carried out with four different types of  
188 columns. Increasing carbon chain length of the chromatographic support, the retention times  
189 of As(III) and As(V) closed in each other, and they completely matched when C<sub>28</sub> and C<sub>30</sub> ODS  
190 columns were used. However, when HNO<sub>3</sub> eluent was applied, the peaks tended to broaden  
191 and peaks of inorganic arsenic overlap to that of organoarsenic species using C<sub>28</sub> and C<sub>30</sub>  
192 columns. On the other hand, As(III) and As(V) always eluted faster than the organoarsenic  
193 species regardless of the carbon chain length of the columns applied. This is not surprising  
194 since the interaction of these polar inorganic arsenic species would show less interaction with  
195 the non-polar reverse phase. Tetramethylarsonium, the only cationic species under these  
196 conditions eluted last among five compounds with the C<sub>8</sub> column, but faster than DMA with the  
197 C<sub>18</sub>, C<sub>28</sub> and C<sub>30</sub> columns. This indicates that the C<sub>8</sub> column may have some ion exchange  
198 properties. This suggestion is also supported by the fact that the anionic As(V) elutes before  
199 the neutral As(III). In addition, the specific surface area of filler of C<sub>8</sub> is the biggest; therefore  
200 the retention time of the first eluting species was slightly longer. In contrast, only C<sub>18</sub> was the  
201 polymer-coated type column, and its specific surface area of filler was the smallest in the  
202 columns; hence, the retention times of As species were faster than the other columns. However,  
203 MMA and DMA were always separated even when using any columns in the group I.

204 Buffers and chelating agents are effective for suppressing peak tailings.[28] A series of  
205 eluents containing 1 to 5 mmol L<sup>-1</sup> of diammonium hydrogen phosphate / 0.05 % methanol was  
206 tested; the pH of the eluents was adjusted to 2.0 with HNO<sub>3</sub>. The eluents containing  
207 diammonium hydrogen phosphate reduced the tailing of the peaks, and improved the separation  
208 capacity. However, it also made As(III) and As(V) peaks split slightly from each other using  
209 C<sub>28</sub> and C<sub>30</sub> columns. On the other hand, it was found that the eluent with lower pH was  
210 effective to elute As(III) and As(V) coincidentally. The ODS columns employed here are  
211 usually used in pH range from 2 to 10, because the Si-C bond of the functional group will be  
212 deteriorated at pH of lower than 2. Therefore, a column with high resistance to acidic  
213 conditions was likely to prove valuable for simultaneous elution of As(III) and As(V).

214 A polymer-coated C<sub>18</sub> ACR column, which is stable at pH range of 1 to 10, was  
215 investigated using the diammonium hydrogen phosphate buffer containing HNO<sub>3</sub> (pH 1.5~3.0)  
216 / 0.05 % methanol. Although the length of the carbon chains of chromatographic support was  
217 shorter than C<sub>28</sub> and C<sub>30</sub> columns, the sharp peaks were observed and the retention times of

218 As(III) and As(V) coincided completely (Table 1), whereas the MMA, DMA and TeMA were  
219 observed at the different retention times. TeMA eluted before all other possible arsenic species  
220 (As(III) and As(V) and DMA, MMA) in rice grains using C<sub>18</sub> ARC column with the  
221 diammonium hydrogen phosphate buffer containing HNO<sub>3</sub> (pH 1.5~2.0) / 0.05 % methanol.  
222 As(III) and As(V) were determined simultaneously as total inorganic arsenic, and the  
223 organoarsenic species were separated from the inorganic arsenic. This means that there is only  
224 one integration necessary which shortens the SOP and may prevent errors in the routine lab.  
225 Although previous methods have detected all inorganic arsenic in one peak, but a complete  
226 oxidation of As(III) to As(V) was necessary prior to the analysis. The completeness of the  
227 oxidation could often not been checked since As(III) was co-eluting with TeMA [25]. Our new  
228 method prevents this ambiguity. In addition, the concentration of diammonium hydrogen  
229 phosphate was not critical for the chromatographic separation under the operating conditions  
230 tested here.

231 As an optimum experimental setups, the C<sub>18</sub> ACR column was selected and the eluent  
232 containing 1 mmol L<sup>-1</sup> diammonium hydrogen phosphate (pH 2.0) / 0.05 % methanol was  
233 flowed at 0.75 mL min<sup>-1</sup>. Under these conditions, the inorganic arsenic was separated from  
234 MMA, DMA and TeMA and determined within approximately 3.5 min.

235

### 236 *3.2 Monitoring time*

237 With the object of developing a rapid monitoring method, the flow rate of the eluent was  
238 investigated to achieve quick and effective separation of the inorganic arsenic from organic  
239 arsenic species, using a C<sub>18</sub> ACR column (particle size 5 μm, ID 4.6 mm × 250 mm). The  
240 inorganic arsenic were well separated from the organoarsenic species at the flow rate range from  
241 0.75 to 1.0 mL min<sup>-1</sup>, although the separation capacity slightly deteriorated at the flow rate more  
242 than 1.1 mL min<sup>-1</sup>, because the separation factor decrease with increasing the flow rate.  
243 Therefore, taking into consideration of chromatographic resolution and measurement rapidity  
244 for monitoring, the flow rate of the eluent was set at 1.0 mL min<sup>-1</sup>, and then the measurement  
245 time of HPLC-ICP-MS was 3 min. Each chromatogram of As(III), As(V), MMA, DMA and  
246 TeMA at a flow rate of 1.0 mL min<sup>-1</sup> are shown in Figure S2. Since the actual separation of  
247 inorganic arsenic from the other organoarsenicals takes only 1 minute with 2 minutes of void,  
248 samples could be injected every minute. This was demonstrated by injecting 3 replicates of one  
249 extract which was determined within 6 minutes (see Figure S3). This is a considerable  
250 improvement in analysis time.

251

### 252 *3.3 Organoarsenic compounds*

253 Agricultural samples occasionally contain very little amounts of TMAO, TeMA, AsB, and

254 AsC, although MMA and DMA were major organoarsenic compounds in rice flour samples.  
255 Therefore, the chromatographic characteristics of TMAO, AsB, and AsC under the operating  
256 conditions established above were investigated. When C<sub>18</sub> ACR column was used, AsB was  
257 eluted at the same retention time as MMA, DMA, TMAO and AsC were eluted at the same  
258 retention time as TeMA. They were clearly separated from the inorganic arsenic. As the  
259 results, the organic arsenic compounds, such as MMA, DMA, TMAO, TeMA, AsB and AsC  
260 were well separated from inorganic arsenic.

261

### 262 *3.4 Extraction of arsenic species from rice flour samples*

263 A heat assisted technique was applied to the extraction of arsenic species from rice flour  
264 samples. The extraction efficiency was investigated using the following four extraction  
265 solutions; 0.15 mol L<sup>-1</sup> HNO<sub>3</sub>, 0.28 mol L<sup>-1</sup> HNO<sub>3</sub>, 0.30 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>, and 0.20 mol L<sup>-1</sup>  
266 H<sub>2</sub>O<sub>2</sub>+0.10 mol L<sup>-1</sup> HNO<sub>3</sub>. [29-30] The certified reference material of NMIJ CRM 7532-a was  
267 analyzed. The extracted solutions were measured by HPLC-ICP-MS with the C<sub>18</sub> ODS  
268 column using the eluent containing an ion pair reagent. [28-34] As(III), As(V) and DMA were  
269 detected after extraction using 0.15 mol L<sup>-1</sup> HNO<sub>3</sub> and 0.28 mol L<sup>-1</sup> HNO<sub>3</sub> (Figure 2). On the  
270 other hand, As(V) and DMA, but not As(III), were detected in the extraction solution of 0.30  
271 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub> and 0.20 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>+0.10 mol L<sup>-1</sup> HNO<sub>3</sub> (Figure 3), because almost all As(III)  
272 was oxidized to As(V) during the extraction process. [30] All the extraction efficiencies were  
273 approximately 100 %, since the inorganic and total arsenic concentrations were in good  
274 agreement with the certified values (Figure 4). The authors reported in the previous report that  
275 almost 100 % of the arsenic species were extracted by the heat assisted technique with acidic  
276 solvents. [30]

277 By the proposed method, the total inorganic arsenic is determined HPLC-ICP-MS  
278 regardless of As(III) and As(V), since they are eluted at the same retention time using C<sub>18</sub> ACR  
279 column in HPLC. The conversion of As(III) into As(V) is not strictly necessary in the method.  
280 However, there are some reports that the detection performances of As(III) and As(V) by  
281 ICP-OES and ICP-MS are slightly different each other under some operation conditions.  
282 [35-37] The single inorganic species [As(V)] measurements makes the measurement errors  
283 smaller than the two species [As(III) and As(V)] measurements and probably increases a  
284 precision of the total analysis. The presence of the oxidant H<sub>2</sub>O<sub>2</sub> in the extraction solution is  
285 required for the oxidization of As(III) during extraction process. H<sub>2</sub>O<sub>2</sub> solution by itself was  
286 able to extract the arsenic species, but the extracted solution was highly viscous, because the  
287 proteins in rice flour were not hydrolyzed. As a result, bothering and lengthy filtration was  
288 required before chromatographic analysis. The presence of a small amount of HNO<sub>3</sub> in  
289 extraction solution makes the extracted solution smooth and the filtration easy. Therefore,



290 0.20 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>+0.10 mol L<sup>-1</sup> HNO<sub>3</sub> was selected as the extraction solvent.

291 Extraction time is obviously important for a rapid monitoring test. The extraction  
292 efficiency of the heat assisted method was investigated at 100 °C using the following extraction  
293 solvents; 0.15 mol L<sup>-1</sup> HNO<sub>3</sub> and 0.20 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>+0.10 mol L<sup>-1</sup> HNO<sub>3</sub>. Results are shown in  
294 Figure 5. The extraction time 0 is non-heating extraction process, and it just stood at room  
295 temperature for 2 h. The extraction efficiency of non-heating process for inorganic arsenic was  
296 approximately 94 %. The complete extraction of inorganic arsenic was achieved in 15 min at  
297 100 °C, when 0.20 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>+0.10 mol L<sup>-1</sup> HNO<sub>3</sub> was used.

298 The working efficiency of monitoring tests for inorganic arsenic analysis is estimated  
299 based on the Codex Alimentarius survey method. The working efficiencies are summarized in  
300 the Table 2. The proposed technique will reduce approximately 40% of total measurement  
301 time, 50% of Ar consumption of ICP-MS, and 30% of eluent consumption of HPLC, comparing  
302 with CODEX Alimentarius survey method.

303

### 304 3.5 Application

305 The proposed method using C<sub>18</sub> ACR was applied to the analysis of the CRMs (Figure 6).  
306 Analytical results of the inorganic arsenic were in good agreement with the certified values  
307 (Table 3).

308 The certified values of NMIJ CRM 7532-a are as follows; inorganic arsenic 0.298 ± 0.008  
309 mg kg<sup>-1</sup> and DMA 0.0186 ± 0.0008 mg kg<sup>-1</sup> as As (the figure following ± indicates is the  
310 expanded uncertainty with  $k=2$ ;  $k$  indicates the coverage factor), and those of NMIJ CRM  
311 7533-a are; inorganic arsenic 0.530 ± 0.016 mg kg<sup>-1</sup> and DMA 0.092 ± 0.004 mg kg<sup>-1</sup> as As  
312 ( $k=2$ ). Those of NIST SRM are as follows; inorganic arsenic is 0.092 ± 0.010 mg kg<sup>-1</sup>, DMA  
313 0.180 ± 0.012 mg kg<sup>-1</sup>, and of MMA 0.0116 ± 0.0035 mg kg<sup>-1</sup> as As ( $k=2$ ).

314 The measurement precision for inorganic arsenic was around 3 %, even when the rice  
315 flour containing inorganic arsenic was less than 0.1 mg kg<sup>-1</sup> (5 ng g<sup>-1</sup> As in a measurement  
316 solution: 0.1 mg kg<sup>-1</sup> × 0.5g / 10g).

317

## 318 4. Conclusions

319 A rapid monitoring method for inorganic arsenic in rice flour was achieved with  
320 HPLC-ICP-MS using C<sub>18</sub> ACR ODS column. By the proposed method, As(III) and As(V)  
321 were completely separated from organic arsenic compounds and eluted at the same  
322 chromatographic retention times using simple component eluent, and thus they were detected as  
323 the inorganic arsenic at a time. Therefore, it was possible to measure inorganic arsenic  
324 selectively, easily and quickly. Moreover, the simple and dilute **one** component of the  
325 chromatographic eluent substantially reduces the deterioration of analytical performances of

326 columns and a detector of HCPLC-ICP-MS. Prior to HPLC-ICP-MS determination, the heat  
327 assisted extraction was applied to the extraction of arsenic compounds from rice flour. The  
328  $\text{H}_2\text{O}_2+\text{HNO}_3$  extraction solvent can extract arsenic compounds efficiently and oxidize As(III) to  
329 As(V) during the extraction process. It facilitates a reliable separation and determination of  
330 arsenic species for HPCL-ICP-MS, since it hydrolyzed proteins in extraction solution.

331 The proposed method requires a single standard solution, As(V), for inorganic arsenic  
332 determination, since the As(III) and As(V) are measured as a single peak of the total inorganic  
333 arsenic and Almost all As(III) is oxidized to As(V) during the extraction process. Low  
334 concentration levels of As(III) and As(V) in standard solutions sometimes interchange by  
335 oxidation-reduction reactions and their detection sensitivities are sometimes different in  
336 ICP-MS measurement. Those events can lead us to measurement errors when As(III) and  
337 As(V) are determined. No error caused by As(III) - As(V) interchange can occur in the  
338 proposed method.

339 The presented method was developed specifically for the rapid monitoring of inorganic  
340 arsenic in rice flour. It is also useful for monitoring inorganic arsenic in other environmental  
341 and food samples. The accurate evaluation methods might be necessary when the  
342 concentration of inorganic arsenic in a sample is very close to the tolerance values.

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#### 344 **Acknowledgment:**

345 Savarin Sinaviwat thanks the Royal Thai Scholarship for financial support.

346

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453 Table 1 Effect of carbon length of ODS column on retention times of arsenic species.

Carbon length	C <sub>8</sub>	C <sub>18</sub>	C <sub>28</sub>	C <sub>30</sub>	C <sub>18</sub> ACR
Retention Time (min)					
As(III)	4.30	3.26	3.78	3.76	2.79
(peak start)	(4.11)	(3.06)	(3.62)	(3.62)	(2.70)
As(V)	4.20	3.22	3.73	3.74	2.79
(peak start)	(4.04)	(3.04)	(3.62)	(3.60)	(2.70)
MMA	4.34	3.52	4.20	4.15	2.98
DMA	4.53	3.73	4.38	4.33	3.03
TeMA	4.79	3.68	4.22	4.18	2.69

454 Flow rate: 0.75 mL min<sup>-1</sup>.455 Eluent: 1 mmol L<sup>-1</sup> diammonium hydrogen phosphate buffer / 0.05 % methanol (pH 2.0 by456 HNO<sub>3</sub>)

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480 Table 2 Working efficiency for monitoring test of inorganic arsenic in rice flour samples.

Content	Step	Codex*	US FDA	Proposed method
ANALYTICAL TIME				
Sample pretreatment				
	Sample taken (e.g. $n=20$ )**	20 min	20 min	20 min
	Addition of extracting solvent**	10 min	10 min	10 min
	Heating time (Extraction)	120 min	90 min	15 min
	Cooling time**	30 min	30 min	30 min
	Centrifugation time**	10 min	10 min	10 min
	Filtration time ( $n=20$ )**	20 min	20 min	20 min
	Dilution ( $n=20$ )	100 min	100 min	60 min
Calibration standards				
	Preparation of standard solution** (e.g. four points, three step dilution)	60 min	60 min	60 min
Measurement				
	Runtime ( $n=24$ ) (samples + standards)	6 min × 24 = 144 min	15 min × 24 = 360 min	3 min × 24 = 72 min
	<b>Total analytical time (min)</b>	514 min	700 min	297 min
	<b>Efficiency factor (based on the Codex)</b>	1.00	1.36	0.58
Equipment				
	Ar gas consumption of ICP-MS	20 L/min × 144 min = 2880 L	20 L/min × 360min = 7200 L	20 L/min × 72 min = 1440 L
	<b>Efficiency factor (based on the Codex)</b>	1.00	2.50	0.50
	Eluent consumption of LC system	0.75 mL/min * 144 min = 108 mL	1.0 mL/min * 360 min = 360 mL	1.0 mL/min * 72 min = 72 mL
	<b>Efficiency factor (based on the Codex)</b>	1.00	3.33	0.67

\* Codex Alimentarius survey method, \*\* Common factor

Except a dry mass correction factor experiment

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485 Table 3 Inorganic arsenic (i-As) in rice flour samples obtained by the proposed method with C<sub>18</sub>486 ACR column (mg kg<sup>-1</sup>).

	Extraction 1	Extraction 2
NMIJ CRM 7532-a		
Certified value of i-As	0.298 ± 0.008 ( <i>k</i> =2)	
Result*	0.295 ± 0.002	0.296 ± 0.002
NMIJ CRM 7533-a		
Certified value of i-As	0.530 ± 0.016 ( <i>k</i> =2)	
Result*	0.522 ± 0.007	0.528 ± 0.007
NIST SRM 1568b		
Certified value of i-As	0.092 ± 0.010 ( <i>k</i> =2)	
Result*	0.093 ± 0.003	0.094 ± 0.003

Extraction time, 1: 15 min, 2: 30 min

Extraction conditions: Temperature 100 °C, Extracting solvent: 0.2 M H<sub>2</sub>O<sub>2</sub> + 0.1 M HNO<sub>3</sub>\*Mean ± SD (*n*=3)

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503 Figure Captions

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505 Figure 1 Retention times of arsenic species according to carbon length of ODS column.

506 Eluent: 0.01 M HNO<sub>3</sub> / 0.05 % methanol, Flow rate: 0.75 mL min<sup>-1</sup>.

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508 Figure 2 Chromatogram of the arsenic species in brown rice flour following extraction with  
509 solvent of 0.15 mol L<sup>-1</sup> HNO<sub>3</sub>.

510 Column: C<sub>18</sub> ODS (particle size 5 μm, 250 mm x ID 4.6 mm), Eluent: 10 mmol L<sup>-1</sup> sodium  
511 1-butanesulfonate / 4 mmol L<sup>-1</sup> malonic acid / 4 mmol L<sup>-1</sup> tetramethylammonium hydroxide /  
512 0.05 % methanol (pH 3.0), Flow rate: 0.75 mL min<sup>-1</sup>, Sample; NMIJ CRM 7532-a, Peaks from  
513 the left: 1st peak As(V), 2nd peak As(III), 3rd peak DMA.

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515 Figure 3 Chromatogram of the arsenic species in brown rice flour following extraction with  
516 solvent of 0.30 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>.

517 Column: C<sub>18</sub> ODS (particle size 5 μm, 250 mm x ID 4.6 mm), Eluent: 10 mmol L<sup>-1</sup> sodium  
518 1-butanesulfonate / 4 mmol L<sup>-1</sup> malonic acid / 4 mmol L<sup>-1</sup> tetramethylammonium hydroxide /  
519 0.05 % methanol (pH 3.0), Flow rate of 0.75 mL min<sup>-1</sup>, Sample; NMIJ CRM 7532-a, Peaks  
520 from the left: 1st peak As(V), 2nd peak DMA.

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522 Figure 4 Influence of solvents on extraction efficiencies.

523 ◆ Inorganic arsenic (i-As), □ Total arsenic(total As), Certified: Certified values of inorganic  
524 arsenic and total arsenic (*k*=1), i-As 1 and total As 1: 0.15 mol L<sup>-1</sup> HNO<sub>3</sub> extraction, i-As 2 and  
525 total As 2: 0.28 mol L<sup>-1</sup> HNO<sub>3</sub> extraction, i-As 3 and total As 3: 0.30 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub> extraction,  
526 i-As 4 and total As 4: 0.20 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub> + 0.10 mol L<sup>-1</sup> HNO<sub>3</sub> extraction, Sample: NMIJ CRM  
527 7532-a brown rice, Extraction; temperature 100 °C, time 2 h, Column: C<sub>18</sub> ODS ACR (particle  
528 size 5 μm, 250 mm x ID 4.6 mm), Eluent: 1 mmol L<sup>-1</sup> diammonium hydrogen phosphate (pH  
529 2.0) / 0.05 % methanol, Flow rate; 0.75 mL min<sup>-1</sup>.

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531 Figure 5 Effect of extraction time on extraction efficiency.

532 ◆ 0.15 mol L<sup>-1</sup> HNO<sub>3</sub> extraction, ■ 0.20 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub> + 0.10 mol L<sup>-1</sup> HNO<sub>3</sub> extraction,  
533 Sample: NMIJ CRM 7532-a brown rice,  
534 Extraction temperature: 100 °C, 0\*: Non-heating extraction at room temperature for 2 h.

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536 Figure 6 Chromatogram of the arsenic species in NMIJ CRM 7533-a and NIST SRM 1568b rice  
537 flour.

538 Column: C<sub>18</sub> ODS ACR (particle size 5 μm, 250 mm x ID 4.6 mm), Eluent: 1 mmol L<sup>-1</sup>  
539 diammonium hydrogen phosphate (pH 2.0) / 0.05 % methanol at a flow rate of 0.75 mL min<sup>-1</sup>  
540 and 1.0 mL min<sup>-1</sup> (proposed optimized condition), Extracting solvent: 0.15 mol L<sup>-1</sup> HNO<sub>3</sub>, Peaks  
541 from the left: 1st peak i-As, 2nd peak Organoarsenic species.

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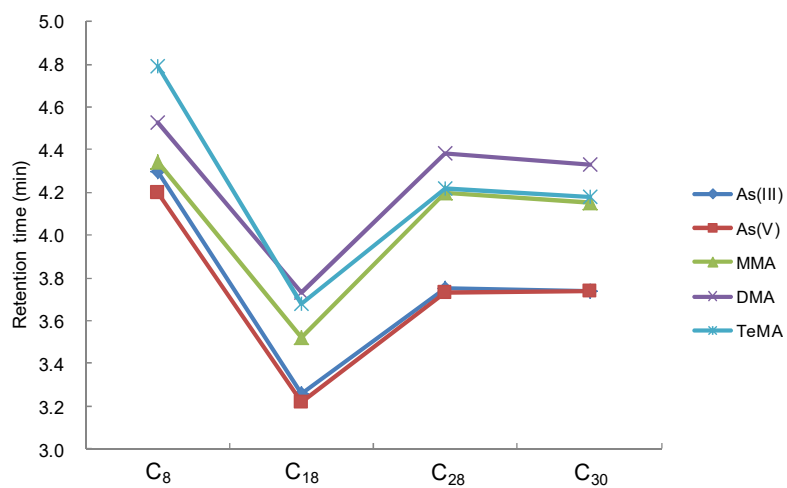


Figure 1 Retention times of arsenic species according to carbon length of ODS column.

Eluent: 0.01 M HNO<sub>3</sub> / 0.05 % methanol, Flow rate: 0.75 mL min<sup>-1</sup>.

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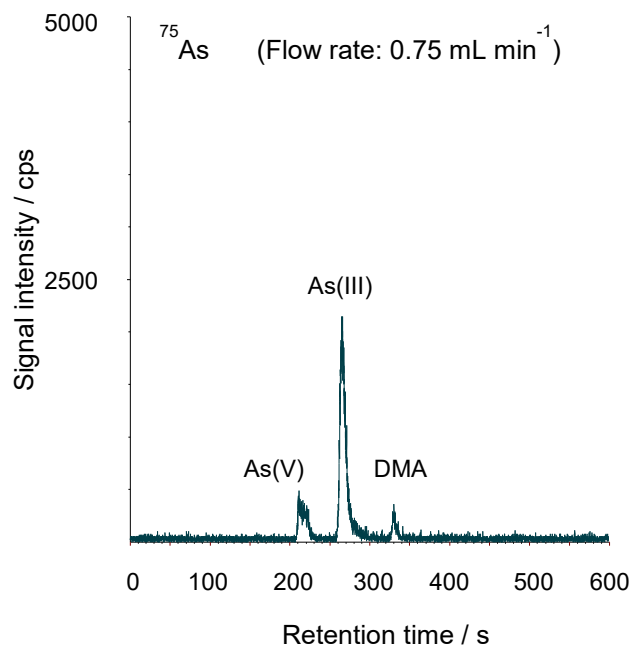


Figure 2 Chromatogram of the arsenic species in brown rice flour following extraction with solvent of 0.15 mol L<sup>-1</sup> HNO<sub>3</sub>.  
Column: C<sub>18</sub> ODS (particle size 5 μm, 250 mm x ID 4.6 mm), Eluent: 10 mmol L<sup>-1</sup> sodium 1-butanefulfonate / 4 mmol L<sup>-1</sup> malonic acid / 4 mmol L<sup>-1</sup> tetramethylammonium hydroxide / 0.05 % methanol (pH 3.0), Flow rate: 0.75 mL min<sup>-1</sup>, Sample; NMIJ CRM 7532-a, Peaks from the left: 1st peak As(V), 2nd peak As(III), 3rd peak DMA.

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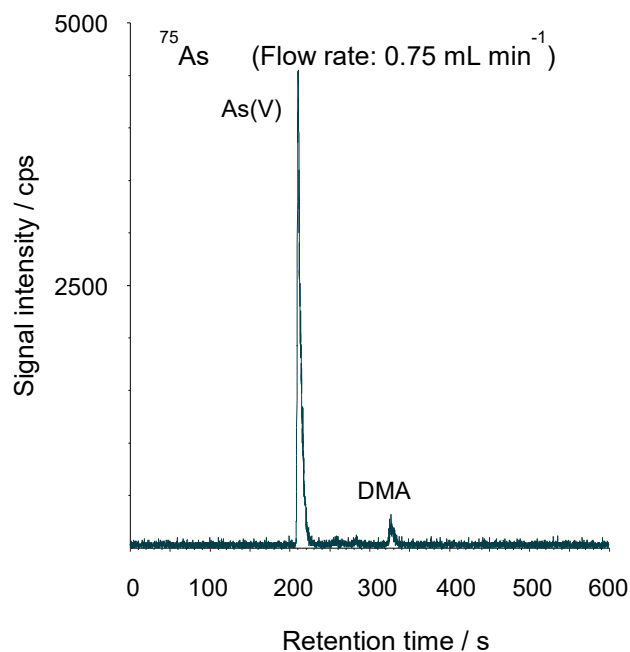


Figure 3 Chromatogram of the arsenic species in brown rice flour following extraction with solvent of 0.30 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub>. Column: C<sub>18</sub> ODS (particle size 5 μm, 250 mm x ID 4.6 mm), Eluent: 10 mmol L<sup>-1</sup> sodium 1-butanesulfonate / 4 mmol L<sup>-1</sup> malonic acid / 4 mmol L<sup>-1</sup> tetramethylammonium hydroxide / 0.05 % methanol (pH 3.0), Flow rate of 0.75 mL min<sup>-1</sup>, Sample; NMIJ CRM 7532-a, Peaks from the left: 1st peak As(V), 2nd peak DMA.

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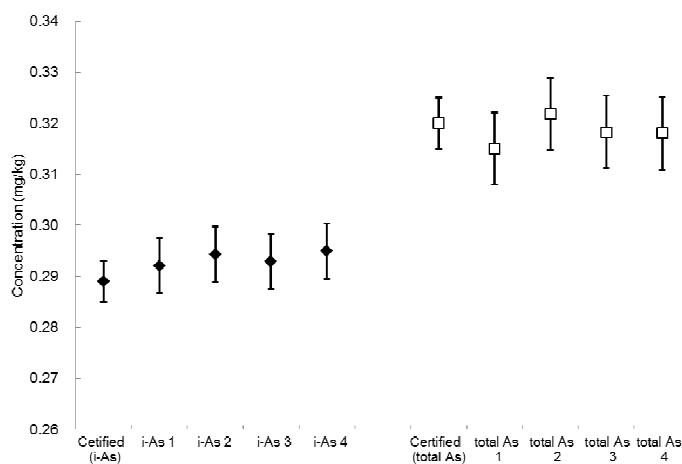


Figure 4 Influence of solvents on extraction efficiencies.

◆ Inorganic arsenic (i-As), □ Total arsenic(total As), Certified: Certified values of inorganic arsenic and total arsenic ( $k=1$ ), i-As 1 and total As 1:  $0.15 \text{ mol L}^{-1} \text{ HNO}_3$  extraction, i-As 2 and total As 2:  $0.28 \text{ mol L}^{-1} \text{ HNO}_3$  extraction, i-As 3 and total As 3:  $0.30 \text{ mol L}^{-1} \text{ H}_2\text{O}_2$  extraction, i-As 4 and total As 4:  $0.20 \text{ mol L}^{-1} \text{ H}_2\text{O}_2 + 0.10 \text{ mol L}^{-1} \text{ HNO}_3$  extraction, Sample: NMIJ CRM 7532-a brown rice, Extraction; temperature  $100 \text{ }^\circ\text{C}$ , time 2 h, Column:  $\text{C}_{18}$  ODS ACR (particle size  $5 \text{ }\mu\text{m}$ ,  $250 \text{ mm} \times \text{ID } 4.6 \text{ mm}$ ), Eluent:  $1 \text{ mmol L}^{-1}$  diammonium hydrogen phosphate (pH 2.0) / 0.05 % methanol, Flow rate;  $0.75 \text{ mL min}^{-1}$ .

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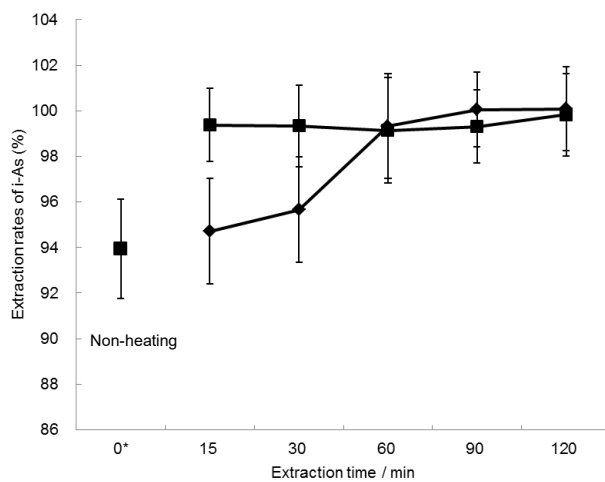


Figure 5 Effect of extraction time on extraction efficiency.

◆ 0.15 mol L<sup>-1</sup> HNO<sub>3</sub> extraction, ■ 0.20 mol L<sup>-1</sup> H<sub>2</sub>O<sub>2</sub> + 0.10 mol L<sup>-1</sup> HNO<sub>3</sub> extraction, Sample: NMIJ CRM 7532-a brown rice, Extraction temperature: 100 °C, 0\*: Non-heating extraction at room temperature for 2 h.

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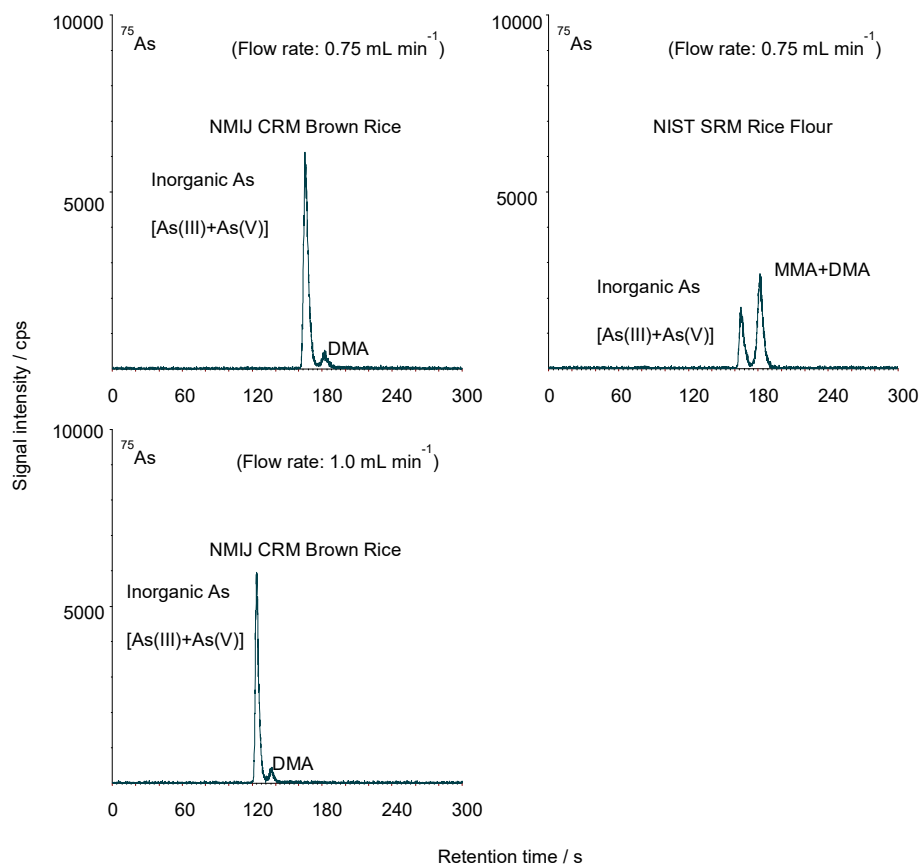


Figure 6 Chromatogram of the arsenic species in NMIJ CRM 7533-a and NIST SRM 1568b rice flour. Column:  $\text{C}_{18}$  ODS ACR (particle size  $5 \mu\text{m}$ ,  $250 \text{ mm} \times \text{ID } 4.6 \text{ mm}$ ), Eluent:  $1 \text{ mmol L}^{-1}$  diammonium hydrogen phosphate (pH 2.0) / 0.05 % methanol at a flow rate of  $0.75 \text{ mL min}^{-1}$  and  $1.0 \text{ mL min}^{-1}$  (proposed optimized condition), Extracting solvent:  $0.15 \text{ mol L}^{-1} \text{ HNO}_3$ , Peaks from the left: 1st peak i-As, 2nd peak Organoarsenic species.



762 Supplementary material

763 Table S1 Measurement parameters and its analytical performances.

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ICP-MS Agilent 7500c

Plasma conditions:

Incident Rf power	1600 W
Reflected power	< 2 W
Outer gas flow rate	Ar 15 L/min
Intermediate gas flow rate	Ar 0.9 L/min
Carrier gas flow rate	Ar 0.8 L/min
Make-up gas flow rate	Ar 0.4 L/min

Sampling conditions:

Nebulizer	Glass 100 $\mu$ L (natural aspirate)
Spray chamber	Scott type (2 °C)
Sample depth	7 mm from work coil

Collision / reaction mode:

Calibration method	He 3.0 mL/min
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Data acquisition:

Dwell time	20 ms / point
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Measured isotopes (*m/z*):

<sup>75</sup>As

HPLC conditions:

Column	ODS columns (particle size of the filler 5 $\mu$ m, ID 4.6 mm x 250 mm)
Eluent	1 mmol L <sup>-1</sup> diammonium hydrogen phosphate (pH 2.0) / 0.05 % methanol
Flow rate	0.75 mL min <sup>-1</sup>
Injection volume	20 $\mu$ L

Analytical performances:

Limit of detection (3 $\sigma$ )	0.01 ng g <sup>-1</sup> (as As)
limit of quantitation (10 $\sigma$ )	0.03 ng g <sup>-1</sup> (as As)
Analytical precision	3 % for 5 ng g <sup>-1</sup> As, 10 % for 1 ng g <sup>-1</sup>

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772 Table S2: Tested columns (Column group II)

Column	Production	Typical separation mode	Particle size of filler	Column size (mm) Inside diameter × Length
PRP-X100	Hamilton	Anion-exchange	10 μm	4.1 x 150
IonPac CS12A	Dionex	Cation-exchange		3.0 x 150
PRP-X300	Hamilton	Ion-exclusion	10 μm	4.1 x 150
RSpak NN-414	Showa Denko K.K.	Reversed phase		4.6 x 150
RSpak NN-614	Showa Denko K.K.	Reversed phase		6.0 x 150
CAPCELL CORE	Shiseido Co., Ltd.	Reversed phase	2.7 μm	4.6 x 100
CAPCELL PAK ADAM S3	Shiseido Co., Ltd.	Reversed phase	3 μm	4.6 x 50
CAPCELL PAK C27 AQ	Shiseido Co., Ltd.	Reversed phase	2.7 μm	4.6 x 100
Shim-pack VP-ODS C18	Shimadzu Co.	Reversed phase	5 μm	4.6 x 250
Sunrise C18 SAC	ChromaNik Technologies Inc.	Reversed phase	5 μm	4.6 x 250

\*For comparison study, typical column conditions were selected the particle size of filler 5 μm and the column size ID 4.6 mm × 250 mm. But, some columns are not on the market: therefore, the different size columns were controlled by its column pressure.

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774 Table S3: Components of the tested eluents for column group II

Component of eluent	Concentration range (mmol L <sup>-1</sup> )	Organic solvent (Methanol %)	pH range
HNO <sub>3</sub>	1 to 10	0 to 0.5	-----
H <sub>2</sub> SO <sub>4</sub>	1 to 10	0 to 0.5	-----
Malonic acid	1 to 5	0 to 0.5	2 to 4
Malonic acid / TMAH	1 to 5	0 to 0.5	2 to 4
NH <sub>4</sub> H <sub>2</sub> PO <sub>4</sub>	1 to 20	0 to 0.5	2 to 8
(NH <sub>4</sub> ) <sub>2</sub> HPO <sub>4</sub>	1 to 20	0 to 0.5	2 to 8
Prigine	5 to 20	0 to 0.5	2 to 4
NH <sub>4</sub> NO <sub>3</sub>	5 to 20	0 to 0.5	2 to 4

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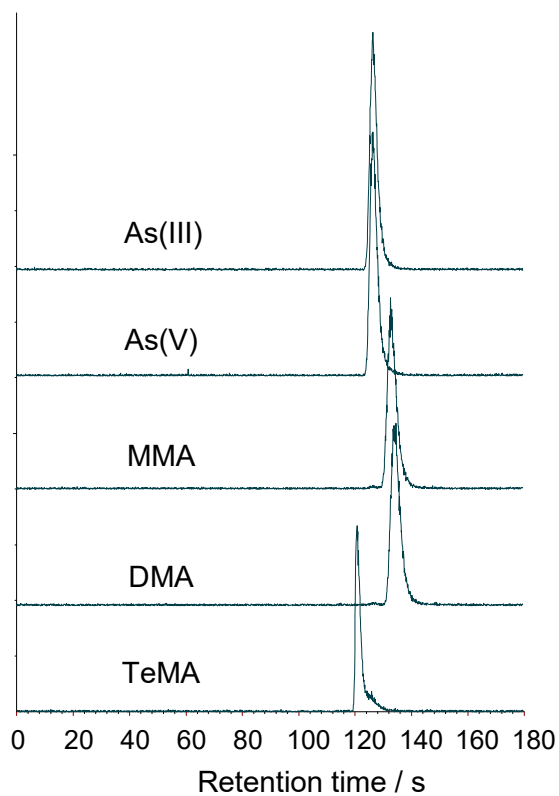


Figure S1: Chromatogram of each arsenic species under the proposed condition.  
Column: C<sub>18</sub> ODS ACR (particle size 5 μm, 250 mm x ID 4.6 mm), Eluent: 1 mmol L<sup>-1</sup> diammonium hydrogen phosphate (pH 2.0) / 0.05 % methanol at a flow rate of 1.0 mL min<sup>-1</sup>.

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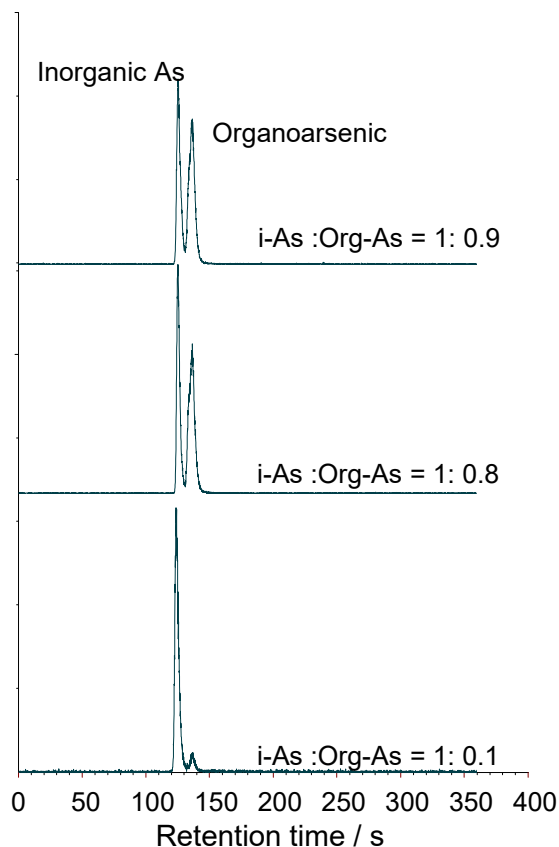


Figure S2: Chromatogram of the arsenic species with different concentrations.

Column: C<sub>18</sub> ODS ACR (particle size 5 μm, 250 mm x ID 4.6 mm), Eluent: 1 mmol L<sup>-1</sup> diammonium hydrogen phosphate (pH 2.0) / 0.05 % methanol at a flow rate of 1.0 mL min<sup>-1</sup> (proposed optimized condition), i-As: As(III)+As(V), Organoarsenic: MMA+DMA.

The peak resolutions in Figure S2 are 1.35 to 1.40. In general, it is said that complete separation is more than 1.5: therefore, approximately 0.2 % may be overlapping each other.

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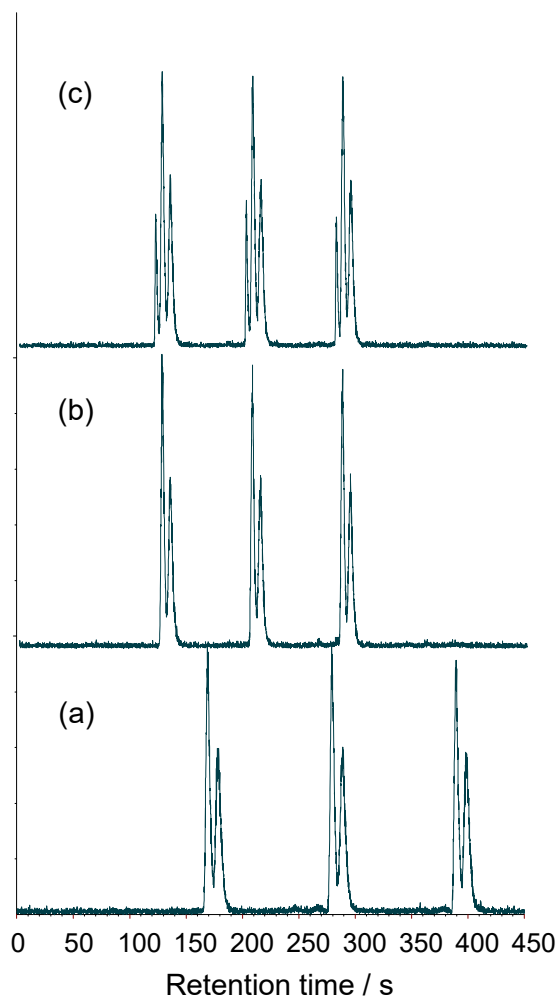


Figure S3: Demonstration results of the rapid monitoring by the proposed method.

Injection interval time: 60 s, Replication: 3, Monitoring time: 450 s, Column: C<sub>18</sub> ODS ACR (particle size 5 μm, 250 mm x ID 4.6 mm), Eluent: 1 mmol L<sup>-1</sup> diammonium hydrogen phosphate (pH 2.0) / 0.05 % methanol, (a) 1<sup>st</sup> peak As(III)+As(V), 2<sup>nd</sup> peak MMA+DMA at a flow rate of 0.75 mL min<sup>-1</sup>, (b) 1<sup>st</sup> peak As(III)+As(V), 2<sup>nd</sup> peak MMA+DMA at a flow rate of 1.0 mL min<sup>-1</sup> (proposed condition), (c) 1<sup>st</sup> peak TeMA, 2<sup>nd</sup> peak As(III)+As(V), 3<sup>rd</sup> peak MMA+DMA at a flow rate of 1.0 mL min<sup>-1</sup> (proposed condition).